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# Impact of high-pressure carbon dioxide on polyphenoloxidase activity and stability of fresh apple juice

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Abstract: Freshly-extracted apple juice was exposed to high pressure carbon dioxide (HP-CO2) treatment at 20, 35 and 45 °C at different pressure conditions (6.0, 12.0 and 18.0 MPa) for up to 30 min. Samples were analysed for residual enzymatic activity. The time needed for 90% enzyme inactivation (Dp) decreased when CO2 pressure increased, while the CO2 pressure sensitivity of the enzyme (zp) showed no variation with temperature. The HP-CO2 treatment at 12 MPa and 35 °C allowed the minimum residual enzyme activity (20%) to be reached in 10 min. Samples treated under these conditions showed lower polyphenoloxidase activity and higher microbial stability than untreated apple juice while presenting a sensory fresh-likelihood higher than thermally pasteurized apple juice.

Dear Editor,

We considered further comments and suggestions and modified the manuscript accordingly. We are thus sending you the revised work.

Kind regards,

Stella Plazzotta

### **Answer to Editor's comments:**

Reference style: all authors (up to 6) should be given at the first citation of a publication (and in the subsequent citations 1st author et al, for more than 2 authors); if more than 6: 1st author et al also at first citation; in the reference list, according to APA "give surnames and initials for up to and including seven authors. When authors number eight or more, include the first six authors' names, then insert three ellipsis points, and add the last author's name". References were modified according to Editor's suggestion (lines 50, 52, 60, 62, 333, 401).

L167-168: Two experiments for each treatment but how many juice batches? How many replicates for the juice production?

Requested information was added in lines 79-83 and 174-175.

Express enzyme activity in katals

Enzyme activity was expressed in katals as requested (lines 141-147, 441).

As Figures 5 and 6 are all related to sensory analysis could you please group them in Figure 5 A and B Figures 5 and 6 were grouped as suggested (lines 346, 349, 357)

HP-CO<sup>2</sup> treatment promotes partial inactivation of polyphenoloxidase in apple juice

HP-CO<sup>2</sup> treatment increases microbial stability of apple juice

HP-CO<sup>2</sup> treatment does not impair apple juice fresh-likelihood



22 inactivation  $(D_p)$  decreased when  $CO_2$  pressure increased, while the  $CO_2$  pressure sensitivity of the 23 enzyme  $(z_p)$  showed no variation with temperature. The HP-CO<sub>2</sub> treatment at 12 MPa and 35 °C allowed the minimum residual enzyme activity (20%) to be reached in 10 min. Samples treated

under these conditions showed lower polyphenoloxidase activity and higher microbial stability than

 untreated apple juice while presenting a sensory fresh-likelihood higher than thermally pasteurized apple juice.

# **Introduction**

 Consumption of unprocessed fruit juices has substantially risen over the last few years, mostly due to the increasing demand for good nutritional quality foods with fresh-like characteristics (Beuchat, 1996; Raybaudi-Massilia, Mosqueda-Melgar, Soliva-Fortuny & Martín-Belloso, 2009). As a consequence of inappropriate manipulation and storage, both spoilage and pathogenic microorganisms can grow, leading to hygienic and quality issues. Enzymatic activity can also contribute to quality depletion, along with physical and chemical changes during the storage (Raybaudi-Massilia et al., 2009). To guarantee product safety and provide an adequate shelf-life, unpasteurized juices are generally distributed under refrigerated conditions. They are traditionally obtained by a combination of formulation strategies such as water activity reduction, nutrient restriction, acidification as well as use of antimicrobial additives (Davidson, 2001). These preservation strategies hardly fit with the current demand for fresh-like juices that are free from additives, generating the need for developing novel non-thermal treatments for juice stabilization.

42 High pressure carbon dioxide  $(HP-CO<sub>2</sub>)$  has been reported as a promising non-thermal technology 43 for the stabilization of fresh products. During the treatment, food is in contact with pressurised  $CO<sub>2</sub>$ 44 at temperature/pressure conditions that may be below or above the critical point  $(31.1 \degree C, 7.38)$ 45 MPa). Typical  $CO_2$  pressure is generally within 4 and 30 MPa, rarely exceeding 50 MPa. 46 Temperature is generally between 20 and 50  $^{\circ}$ C, low enough to maintain the freshlikelihood of treated products (Manzocco et al., 2016).

48 Significant lethal effects of HP-CO<sub>2</sub> on different microorganisms have been demonstrated in fruit juices (Spilimbergo & Bertucco, 2003; Damar & Balaban, 2006; Ferrentino, Bruno, Ferrari, Poletto 50 & Balaban, 2009; Xu, Zhang, Wang, Bi, Buckow & Liao, 2011). In particular, the technology is known to promote up to 5 Log reductions in microbial counts, approaching those required for

52 pasteurization (Kincal, Hill, Balaban, Portier, Wei & Marshall, 2005; Ferrentino & Spilimbergo, 53 2011). The germicidal activity of HP-CO<sub>2</sub> is due to the combination of temperature, pressure and 54 specific effects of HP-CO<sub>2</sub>. The treatment is associated with extracellular and intracellular 55 acidification, destabilization of membranes and denaturation of microbial enzymes (Jones & 56 Greenfield, 1982; Hutkins & Nannen, 1993; Bothun, 2004; Bothun, Knutson, Strobel & Nokes, 57 2005). More controversial is the effect of  $HP-CO<sub>2</sub>$  in inactivating fruit enzymes leading to juice 58 quality decay. For instance, inactivation of polyphenoloxidase responsible for browning of fruit 59 juices, depends on the nature of the enzyme and is strongly affected by  $CO<sub>2</sub>$  pressure, temperature 60 and treatment time (Gui, Chen, Wu, Wang, Liao & Hu, 2006; Liao, Zhang, Bei, Hu & Wu, 2009; 61 Zhou, Zhang, Hu, Liao & He, 2009; Spilimbergo, Komes, Vojvodic, Levaj & Ferrentino, 2013). 62 The mechanisms involved in enzyme inactivation by  $HP-CO<sub>2</sub>$  include pH lowering (**Balaban**, 63 Arreola, Marshall, Peplow, Wei & Cornell, 1991) and changes in the conformation of the secondary 64 structure of the enzyme (Chen, Balaban, Wei, Marshall & Hsu, 1992; Manzocco et al., 2016). 65 Based on these considerations, the present paper was addressed to investigate the impact of  $HP\text{-}CO<sub>2</sub>$ 66 treatment on polyphenoloxidase activity and stability of fresh apple juice intended for refrigerated 67 storage. To this aim, apple juice was exposed to  $HP-CO<sub>2</sub>$  treatments in a wide range of pressure, 68 temperature and treatment time conditions. Apple juice was then submitted to the  $HP-CO<sub>2</sub>$  treatment 69 leading to the minimum polyphenoloxidase activity at the mildest pressure/temperature combination 70 and stored at 4  $\degree$ C for up to 15 days. HP-CO<sub>2</sub> treated apple juice was monitored during storage for 71 residual polyphenoloxidase activity, colour, microbial counts and sensory attributes. To verify the 72 potential applicability of  $HP-CO<sub>2</sub>$  technology to produce fresh apple juice, data were compared to 73 those relevant to an untreated apple juice. An apple juice submitted to conventional thermal 74 pasteurization was also considered as additional control.

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#### 76 **2. Material and methods**

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79 A 10 kg batch of fresh apples "Golden delicious" were purchased at the local market and stored at 4 80 <sup>o</sup>C overnight. When the experiments were performed, apples had a dry matter content of 164.7  $\pm$  1.6 81 g/kg, a soluble solid content of 13.3  $\pm$  0.2 °Brix, a pH of 4.2  $\pm$  0.2 and a titratable acidity of 82  $4.6 \pm 0.3$  g/kg. Apple juice was prepared fresh for every trial from the same batch of fruits, to 83 minimize sample variability. The juice was obtained by using a domestic juicer (Moulinex, mod. Vitae JU2000, Milan, Italy), filtered through two layers of cloth filter and centrifuged at 5000 g for 85  $\pm$  5 min at 4 °C (Beckman, Avanti<sup>TM</sup> J-25, High performance centrifuge, Brea, USA). The supernatant was filtered again through two layers of cloth filter and the resulting clear juice was immediately treated.

#### *2.2 High-pressure CO<sup>2</sup> treatments*

90 HP-CO<sub>2</sub> inactivation process was carried out in a double-batch apparatus. The system consists of two identical stainless steel cylinders with a screwed cap and an internal volume of 150 mL, connected in parallel. Each reactor was connected to an on-off valve that can be used to depressurise it independently from the other. The two reactors were submerged in a thermostatic water bath (CB 8-30e, Heto, Allerød, Denmark). For more details, please refer to Manzocco et al. (2016). Before starting the pressurisation, the temperature of the sample was allowed to reach 96 equilibrium. The time needed to reach the desired temperature (20, 35 or 45  $^{\circ}$ C) and the pressurisation time were lower than 3 min. After reaching the desired pressure (6, 12 and 18 MPa), the pump was switched off and valves connected to each vessel were tightly closed. After increasing treatment time up to 30 min, vessels were depressurised. In all experiments, depressurisation was completed within 10 min and the outlet flow was controlled using a digital flowmeter (PFM 750, SMC Italia S.p.A., Milan, Italy). Control samples were prepared by treating 102 the apple juice in the vessels at atmospheric pressure  $(0.1 \text{ MPa})$  and thus at  $CO<sub>2</sub>$  partial pressure equal to 0.0039 MPa.

#### *2.3 Thermal treatment*

 Aliquots of 100 mL of apple juice were placed in plastic pouches (PA/PE, 20 x 28 cm, Savonitti, Codroipo, Italy). A thin layer of sample was obtained, being the maximum thickness of the filled pouches lower than 0.5 cm. Pouches were heated in a water bath (IKA-Werke, Staufen, Germany) 109 at 71.1  $\degree$ C for 6 s (FDA, 2004). After thermal treatment, samples were quickly cooled under running water at room temperature.

#### *2.4 Apple juice storage*

 Aliquots of 10 mL of apple juice were introduced in Eppendorf® vials of 10 mL capacity and 114 stored for up to 15 days at 4 °C in a refrigerated cell. At increasing time during storage, samples 115 were removed from the refrigerator, equilibrated at  $22 \degree C$  and submitted to the analysis.

# *2.5 Apple physical-chemical parameters*

 Soluble solid content (SSC) was measured using a table refractometer (Unirefrax, Bertuzzi, Milan, Italy) calibrated with distilled water.

 Dry matter content of apple samples was determined gravimetrically by recording difference in 121 weight before and after drying at 70 °C, until a constant weight was achieved (M.U.A.C.V., 1989).

Titratable acidity was determined by titration with NaOH 0.1 mol/L and phenolphthalein as

indicator (Sigma-Aldrich, Milan, Italy), accordingly to the official M.U.A.C.V. method (1989) and

expressed as g of acids/kg of fresh product.

Analyses of SSC and TA were carried out on the solution obtained after homogenization (Polyton,

Kinematica, Luzern, Switzerland) and filtration of apple cubes through filter paper (Whatman #1,

Whatman International Ltd, Maidstone, UK).

130 During  $HP-CO<sub>2</sub>$  treatments and thermal pasteurization, temperature was measured by a thermocouple probe (Hanna Instruments, Tersid s.r.l., Milan, Italy); pH was assessed using a pH-meter (Mettler Toledo 355, Lou Analyzer, Halstead, England).

#### *2.7 Polyphenoloxidase activity*

 The polyphenoloxidase activity was assayed spectrophotometrically (Shimadzu UV-2501PC, UV- Vis recording spectrophotometer, Shimadzu Corporation, Kyoto, Japan) at 25 °C according to the 137 methodology of Kahn (1995). The reaction was started by the addition of 500 µL of apple juice to 2 138 mL of 0.1 mol/L potassium phosphate buffer pH 7 and  $1.5 \times 10^{-3}$  mol/L L-Dopa (Carlo Erba, Milan, Italy). The absorbance at 420 nm was monitored each minute for 10 min. The changes in absorbance per min were calculated by linear regression, applying the pseudo zero order kinetic 141 model. The eventual final stationary phase was excluded from regression data. The slope of the very 142 first linear part of the reaction curve was used to determine polyphenoloxidase specific activity. The 143 latter was defined as the amount of enzyme that produced 1 *µ*mol of quinone per second (*µkatal*) 144 (Lee et al., 2010). The average polyphenoloxidase activity in untreated juice was found to be 0.047 *μ*katals. Polyphenoloxidase residual activity (*RA*%) upon treatments was calculated as the 146 percentage ratio between enzymatic activity of the treated sample and that of the untreated one (de la Rosa et al., 2011; Niu et al., 2010; Xu et al., 2011).

#### *2.8 Browning*

 Browning was assessed spectrophotometrically (Shimadzu UV-2501PC, UV-Vis recording spectrophotometer, Shimadzu Corporation, Kyoto, Japan) measuring absorbance values at 420 nm 152 and 25 °C of apple juice samples, diluted with water to obtain absorbance signals within the scale.

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- *2.9 Microbiological analyses*

 For microbiological analyses, appropriate aliquots (0.1 or 1 mL) of properly diluted apple juice were spread on agar plates. Plate Count Agar (Oxoid, Milan, Italy) and Man Ragosa Sharpe (MRS) (Oxoid, Milan, Italy) were used for enumeration of mesophilic and lactic acid bacteria respectively, and plates were incubated for 48 h at 37 °C. Oxytracycline-Glucose-Yeast Extract (OGYE) agar (Oxoid, Milan, Italy) was used for enumeration of yeasts and moulds, and plates were incubated for 160 72 h at 28 °C.

# *2.10 Sensory analysis and off-odour perception*

 A panel of 20 Italian assessors, not trained but expert in method, was selected. For sensory testing, 164 10 mL apple juice was served in odourless plastic glasses at 6 °C. Water was used between samples for mouth rinsing. Samples were indicated by a three-digit code and served the panel paired with a just prepared control sample, identified as "reference". The judges were asked to evaluate sample colour, apple flavour, cooked taste and acidity assigning each descriptor a score on a 9-point scale anchored with "reference" at point 5. Judges were also asked to indicate the possible perception of off-flavours. Off-flavour perception data were expressed as percentage of judges that identified the defect as respect to the reference. Sensory analysis requiring juice drinking was only carried out until all samples had total viable count lower than 5 Log CFU/g.

- 
- *2.11 Data analysis*

174 Data reported in this work are expressed as mean  $\pm$  S.D of at least three measurements carried out 175 on two experiments replicated on different juice extraction batches.

 Apparent inactivation rate constants of polyphenoloxidase were analysed by using a conventional first-order equation:

 $\boldsymbol{d}$ 178  $\frac{u(NA\%)}{dt} = -k \cdot (RA\%)$  (1)

- 179 Where  $RA\%$  is the polyphenoloxidase residual activity in the juice at time *t* (min) and *k* is the 180 inactivation rate constant (min<sup>-1</sup>). The value of  $k$  was obtained as the slope of the regression of the 181 decimal logarithm of *RA%* vs. *t.*
- 182 The eventual final stationary phase was excluded from regression data.
- 183 The value of *RA%* after 30 min was taken as an indicator of the minimum *RA%* achievable by the 184 treatment.
- 185 The kinetic parameter *D<sup>P</sup>* was obtained using procedures analogous to that employed in thermal 186 death time studies. In particular,  $D<sub>P</sub>$  is the decimal reduction time, i.e. the treatment time needed for 187 90% enzyme activity reduction at a given pressure and temperature. *D<sup>P</sup>* was computed as the 188 negative reciprocal of *k*.
- 189 The pressure increase needed for a 90% reduction of the  $D_P$  value was computed as  $z_p$  (MPa). The 190 value of  $z_p$  was obtained by regressing the decimal logarithm of  $D_p$  versus pressure (P):

$$
191 \qquad log D_p = -\frac{P}{z_P} \tag{2}
$$

192 The  $z_P$  was then derived as the negative reciprocal slope of the regression line.

193 The pressure dependence of the inactivation rate constants (*k*) was expressed by the activation 194 volume  $(\Delta V^2, \text{cm}^3/\text{mol})$ , according to the Eyring equation (Weemaes, Ludikhuyze, Van den Broeck 195 & Hendrickx 1998):

$$
196 \quad lnk = lnk_{atm} - \frac{\Delta V^2}{RT} \cdot (P - P_{atm}) \tag{3}
$$

197 where *P* is pressure (MPa), *katm* is the inactivation rate constant at ambient pressure *Patm* (0.1 MPa), 198 R is the gas constant (8.31 cm<sup>3</sup> MPa K<sup>-1</sup> mol<sup>-1</sup>) and T is temperature (K).  $\Delta V^2$  was estimated from 199 the slope of the line obtained by the regression of the natural logarithm of *k* vs. *P*.

200 Goodness-of-fit was evaluated by means of the determination coefficients  $(R^2)$ . Analysis of 201 variance (ANOVA and Tuckey test) were accomplished using the v. 3.1.1 of R software (The R 202 foundation for statistical computing), to determine the significance at a 95% level.

203

#### **3. Results and discussion**

*3.1. Effect of high pressure carbon dioxide on apple juice polyphenoloxidase activity*

 Figure 1 shows the residual polyphenoloxidase activity of apple juice as a function of treatment 208 time at different pressures at 20, 35 and 45 °C.

209 Control apple juice treated for increasing time at 20 °C under environmental conditions (0.1 MPa 210  $CO<sub>2</sub>$  pressure) showed a significant decrease in polyphenoloxidase activity. According to Le Bourvellec, Le Quéré, Sanoner, Drilleau & Guyot (2004), this effect is probably due to the formation of chemically oxidised polyphenols with anti-enzymatic activity upon contact of apple 213 derivatives with oxygen. Exposure of apple juice to increasing  $CO<sub>2</sub>$  pressure resulted in progressively higher enzyme inactivation (Figure 1a). However, even applying 18 MPa for 30 min, 215 the complete inactivation was not achieved. When apple juice was exposed to HP-CO<sub>2</sub> at 35 °C, polyphenoloxidase inactivation was more intense (Figure 1b). For instance, the minimum residual 217 activity was reached upon few min of exposure to 18 MPa  $CO<sub>2</sub>$ . The effect of temperature on 218 enzyme inactivation by HP-CO<sub>2</sub> was further confirmed by additional trials carried out at 45  $^{\circ}$ C (Figure 1c). In accordance with evidences from other Authors (Vamos-Vigyazo, 1981; Gui et al., 2007), these data demonstrate the existence of a negative relation between polyphenoloxidase 221 activity and the increase in both  $CO<sub>2</sub>$  pressure and temperature, at least under the experimental conditions here tested.

 Data shown in Figure 1 were analysed considering the minimum residual activity achievable by each treatment (Table 1). Due to the monotonic decrease of residual activity curves (Figure 1), the value after 30 min of juice treatment was taken as an indicator of the residual activity achievable at 226 each temperature/pressure combination (Table 1). In particular, under the most intense  $CO<sub>2</sub>$  treatment conditions (18 MPa, 45 °C), a minimum residual activity (*RA*%) of 20% was still observed. It can be hypothesised that more intense treatments than those here performed are needed to reach complete inactivation. To this regard, contradictory information is reported in the literature.

 In particular, Xu et al. (2011) found that polyphenoloxidase was completely inactivated by a 231 treatment carried out at 22 MPa and 60 °C for 10 min. A similar effect was also observed by Niu et 232 al. (2010) in apple slices treated at 20 MPa and 25 °C for 20 min. However, other authors reported 233 that even applying  $CO_2$  at 60 MPa and 55 °C for 60 min, a 40% minimum residual activity of polyphenoloxidase was still present (Gui et al., 2007). These different inactivation degrees can be attributed to many factors, including not only operative conditions, but also apple cultivar and derivative as well as equipment layout and operative parameters (Yemenicioğlu, Özkan, Cemeroğlu, Mehmet & Yemeniciog, 1997; Weemaes et al., 1998; Buckow, Weiss & Knorr, 2009; Xu et al., 2011).

 The residual activity of polyphenoloxidase was regressed as a function of treatment time in the initial linear part of the curve (Figure 1) to obtain rate constants (*k*) of polyphenoloxidase 241 inactivation (Table 1). The latter were then used to calculate  $D_p$  values using procedures analogous 242 to that employed in thermal death time studies (Table 1). In particular,  $D_p$  was defined as the treatment time needed for 90% enzyme activity reduction at a given pressure. The treatment at 6 MPa and 20 °C led to a tenfold decrease of activity in *circa* 48 min. At the same temperature, this goal was achieved at 12 MPa in *circa* 34 min. On the other hand, keeping the pressure constant at 6 MPa, inactivation was achieved at 35 or 45 °C in less than 25 or 12 min, respectively. These *D<sup>p</sup>* values suggest a lower resistance of apple juice polyphenoloxidase than that reported in the literature. To this regard, Gui et al. (2007) reported a 220 min *D<sup>p</sup>* value for polyphenoloxidase of 249 cloudy juice from *Fuji* apples upon exposure at 35  $\degree$ C to 30 MPa CO<sub>2</sub>. These differences confirm the significant effect of processing conditions and *cultivar* on inactivation of apple polyphenoloxidase.

252  $D_p$  values (Table 1) were used to calculate the parameter  $z_p$ , describing the sensitivity of 253 polyphenoloxidase to pressurised  $CO<sub>2</sub>$ . The decimal logarithmic values of  $D<sub>p</sub>$  resulted well 254 correlated ( $\mathbb{R}^2$  >0.89; p<0.05) with pressure for treatments carried out at 20, 35 and 45 °C (Table 2).

255 The  $z_p$  value, which represents the pressure range within which the  $D_p$  changes tenfold, resulted *circa* 20 MPa for treatments carried out at 20 °C. This indicates that an increase in pressure of 20 MPa is necessary to get a 90 % decrease in *D<sup>p</sup>* at this temperature. The increase in temperature from 258 20 to 45 °C did not cause a significant decrease in  $z_p$  value, indicating a constant effect of pressure in inactivating the enzyme, at least within the temperature range here tested. To this regard, contradictory data are reported in the literature. Weemaes et al. (1998) detected antagonistic effects of pressure and temperature studying the effects of high static pressure combined with heating treatments on avocado polyphenoloxidase. On the other hand, when conditions similar to those here considered were applied on a plyphenoloxidase model system, a synergistic effect of pressure and 264 temperature was observed (Manzocco et al., 2016). Values of  $z_p$  (Table 2) thus emphasise the 265 critical role of enzyme origin and reaction media in determining its sensitivity to  $CO<sub>2</sub>$  pressure.

266 The effect of pressure on polyphenoloxidase inactivation was also expressed through the activation 267 volume  $(\Delta V^2)$  concept (Table 2). According to the transition state theory, the activation volume is a 268 measure of the volume difference between the initial reactants and the activated complex at the 269 transition state (Eyring, 1935). Data reported in Table 2 show that polyphenoloxidase inactivation is 270 characterized by negative  $\Delta V^2$  with high absolute value ( $\mathbb{R}^2$  >0.89; p<0.05). This indicates that the 271 increase in pressure strongly favoured the denaturation of the enzymatic protein (Ohmae, 272 Murakami, Gekko & Kato, 2007). The values of activation volume for apple juice 273 polyphenoloxidase resulted lower than that reported in the literature (-94.3 cm<sup>3</sup> mol<sup>-1</sup> at 55 °C; Gui 274 et al., 2007). This result indicates that, in our experimental conditions, polyphenoloxidase was more 275 susceptible to  $CO_2$  pressure variation. In addition, in agreement with  $z_p$  values, activation volume 276 did not significantly decrease with the increase in temperature from 20 to 45 °C (Table 2). The 277 increase in temperature promoted instead a significant increase in the pre-exponential or frequency 278 factor (*ln katm*). The latter indicates how often the enzyme is properly oriented to undergo structural 279 modifications leading to denaturation. In fact, the increase in temperature from 20 to 45 °C

 enhanced the frequency factor (Table 2), indicating a higher frequency of steric conditions favouring denaturation.

# *3.2. Effect of high pressure carbon dioxide on apple juice quality during storage*

284 In the light of the previous results, it can be hypothesized that  $HP-CO<sub>2</sub>$  treatment could represent a non-thermal technological strategy to control enzymatic activity during storage of apple juice. In 286 this context, HP-CO<sub>2</sub> could be proposed as a technology to stabilize refrigerated apple juice colour. To verify this hypothesis, a combination of processing conditions which could be potentially applicable on a larger scale to produce fresh refrigerated apple juice was selected. To this regard, 289 treatments carried out at 20 °C were excluded since associated with residual activity higher than 30% even at the highest tested pressure (Table 1). The mildest pressure/temperature combination leading to the minimum residual activity (20%) was thus selected. As shown in Table 1, this 292 combination corresponded to the treatment at 12 MPa and 35  $^{\circ}$ C. The juice was thus treated at these conditions for 10 min since longer treatment times did not promote further enzyme inactivation (Figure 1).

 Even if similarly effective in terms of enzyme inactivation, treatments at temperature and pressure 296 higher than 35 °C and 12 MPa respectively, were not considered since they are reasonably more energy-intensive and thus less sustainable from an environmental point of view.

298 Juice submitted to the selected treatment was then stored for up 15 days at 4  $\degree$ C to simulate conventional distribution conditions of not thermally stabilized apple juice.

 During storage, apple juice was analyzed for the evolution of polyphenoloxidase activity and browning. Microbial and sensory analyses were also performed to evaluate the hygienic level and the intensity of typical sensory attributes of the juice. Data were compared to those relevant to an untreated apple juice as well as a control thermal pasteurized apple juice (71 °C for 6 s). As expected, the latter presented no enzymatic activity during the entire storage time, in agreement with literature data (Golan-Goldhirsh, Whitaker & Kahn, 1984; McEvily, Iyengar & Otwell, 1992).

 By contrast, HP-CO<sub>2</sub> treated and untreated apple juice showed different initial polyphenoloxidase activity, which progressively decreased during storage, approaching in both cases 5% after 10 days (Figure 2). These different inactivation degrees were probably associated with different evolution of browning (Whitaker, 1995; Yoruk & Marshall, 2006).

For this reason, juice browning was assessed spectrophotometrically at 420 nm (Figure 3).

 Immediately after preparation, untreated, pasteurized and HP-CO<sub>2</sub> treated apple juices showed not significantly different browning. As expected, pasteurized juice did not show changes in browning during storage, due to the complete and irreversible inactivation of polyphenoloxidase upon thermal 314 treatment. On the contrary, an increase in browning was detected in both untreated and  $HP-CO<sub>2</sub>$  treated samples. Beyond 3 days of storage, the latter showed browning values not significantly 316 lower than those of the untreated sample, suggesting that the  $HP-CO<sub>2</sub>$  treatment here applied was not able to significantly reduce browning phenomena during storage.

318 To evaluate the ability of  $HP-CO<sub>2</sub>$  treatment to stabilize fresh apple juice against microbial spoilage, total viable and lactic acid bacteria, yeasts and moulds were determined during storage. Whilst pasteurized apple juice always presented microbial counts below the detection limit (data not shown), untreated and HP-CO<sub>2</sub> treated apple juice showed different evolution of these microbial populations during storage (Figure 4).

 In the just prepared apple juice, lactic acid bacteria and moulds were below the detection limit and the total bacterial count was mainly represented by yeasts. These microorganisms are well known to be the main spoilage agents in fruit derivatives, due to their low pH (Raybaudi-Massilia et al*.,*  2009). During the refrigerated storage, total bacterial count of untreated juice progressively increased up to 6 Log CFU/mL (Figure 4a) due to the growth of both yeasts and lactic acid bacteria (Figures 4b and c). HP-CO<sub>2</sub> treatment allowed to decrease the initial count of total viable bacteria and yeasts and to inhibit their growth during the storage (Figure 4a and 4b). In fact, after 15 days of refrigerated storage, total bacteria and yeast resulted about 4 and 1 Log lower than that of the untreated sample, respectively. On the contrary, lactic acid bacteria were below the detection limit

 in HP-CO<sub>2</sub> treated juice, independently on storage time. Lactic acid bacteria have been actually 333 reported to be more HP-CO<sub>2</sub> sensitive than yeasts (García-Gonzalez, Geeraerd, Elst, Van Ginneken, 334 Van Impe & Devlieghere, 2009). As reported in the literature, the antimicrobial effects of HP-CO<sub>2</sub> are attributed not only to pressurization but also to media acidification (Balaban et al., 1991). To 336 this regard, HP-CO<sub>2</sub> treatments were associated to a decrease in product pH due to the presence of residual carbonic acid after the treatment (Hong, Park & Pyun, 1997; Xu et al., 2011). However, in 338 this study, the pH of the juice  $(4.2 \pm 0.2)$  did not change upon the HP-CO<sub>2</sub> treatment and resulted 339 analogous to that of pasteurized and untreated apple juice ( $p > 0.05$ ). It is thus likely that  $CO<sub>2</sub>$  residues were removed from the juice during the depressurization of the reactor after the treatment. 341 In order to evaluate the possible impact of  $HP$ - $CO<sub>2</sub>$  treatment on sensory parameters of apple juice, 342 the samples, stored for increasing time at  $4^{\circ}$ C, were submitted to sensory evaluation. No significant changes in the evolution of the sensory attributes "acidity", "fresh apple flavour" and "sweetness" were detected by the panelists (p>0.05). In addition, judges did not detect any off-flavour in the samples. By contrast, significant changes in the scores of the sensory attributes "browning" and

346 "cooked apple flavour" were noticed ( $Figure 5$ ).

 Due to complete polyphenoxidase inactivation, the browning sensory score of pasteurized sample resulted always significantly lower than that observed in untreated and HP-CO<sub>2</sub> treated juices, 349 which showed progressively higher values ( $\frac{Figure 5a}{?}$ ), mimicking the evolution of absorbance at 420 nm during storage (Figure 3).

 Immediately after the treatment, the pasteurized juice presented a high "cooked-apple flavour" score, confirming the well-known sensory quality depletion induced by thermal treatment (Aguilar- Rosas, Ballinas-Casarrubias, Nevarez-Moorillon, Martín-Belloso & Ortega-Rivas, 2007). After few days of storage, the intensity of this defect in the pasteurized juice decreased, possibly because of the evolution of the juice sensory profile. On the contrary, judges were not able to detect this defect in the HP-CO<sub>2</sub> treated juice. The latter also presented mean "cooked-apple flavour" values 357 comparable to those of the untreated juice during the entire storage period ( $p > 0.05$ ) (Figure 5b).

358 This result is certainly attributable to the low temperature (35  $^{\circ}$ C) experienced by the juice during 359 the  $HP$ - $CO<sub>2</sub>$  treatment.

# **4. Conclusions**

362 HP-CO<sub>2</sub> treatments at temperatures lower than  $45\degree$ C may allow partial inactivation of polyphenoloxidase in apple juice. The treatment time needed for reaching the minimum residual activity decreases with pressure and temperature but no further inactivation is obtained by 365 increasing pressure and temperature beyond 12 MPa and 35  $\degree$ C respectively. HP-CO<sub>2</sub> treatment could be applied under mild pressure/temperature conditions for short times to allow a significant microbial stabilisation of fresh refrigerated apple juice without impairing its fresh-likelihood. Being HP-CO<sup>2</sup> treatment cheap and sustainable, these outcomes make it an interesting stabilisation technology for the production of fresh refrigerated apple juice.

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486 Table 1. Minimum residual activity (*RA%*) of polyphenoloxidase, apparent inactivation rate 487 constants (k), relevant determination coefficient  $(R^2)$  and  $D_p$  values following HP-CO<sub>2</sub> treatments of 488 apple juice at increasing pressures  $(0.1, 6.0, 12.0, 18.0, MPa)$  at 20, 35 and 45 °C.

<b>Temperature</b>	<b>Pressure</b>	<b>Minimum</b>	$\boldsymbol{k}$	${\bf R}^2$	$\boldsymbol{D_p}$
$({}^{\circ}C)$	(MPa)	RA%	$(min^{-1})$		(min)
20	0.1	$69.0 \pm 2.2$ <sup>a</sup>	$-0.0053 \pm 0.0008$ <sup>b</sup>	0.94	$188.8 \pm 26.8^{\text{a}}$
	6	$55.3 \pm 1.0^{\mathrm{b}}$	$-0.0207 \pm 0.0024$ <sup>b</sup>	0.97	$48.4 \pm 5.6^{\rm bc}$
	12	48.0 $\pm$ 1.8 $^{\circ}$	$-0.0297 \pm 0.0053$ <sup>b</sup>	0.94	$33.7 \pm 6.0^{\rm bc}$
	18	$30.2 \pm 1.5$ <sup>de</sup>	$-0.0452 \pm 0.0063$ <sup>b</sup>	0.96	$22.1 \pm 3.1$ <sup>c</sup>
35	0.1	$35.5 \pm 0.6$ <sup>d</sup>	$-0.0133 \pm 0.0019^{b}$	0.94	$75.1 \pm 10.5^{\mathrm{b}}$
	6	$24.6 \pm 0.8^{\circ}$	$-0.0394 \pm 0.0107$ <sup>b</sup>	0.87	$25.4 \pm 6.9$ <sup>bc</sup>
	12	$18.0 \pm 1.2$ <sup>f</sup>	$-0.0660 \pm 0.0180$ <sup>b</sup>	0.84	$15.2 \pm 4.1$ <sup>c</sup>
	18	$18.0 \pm 0.6$ <sup>f</sup>	$-0.1228 \pm 0.0530$ <sup>b</sup>	0.97	$8.1 \pm 3.5$ <sup>c</sup>
45	0.1	$32.5 \pm 1.3$ <sup>de</sup>	$-0.0465 \pm 0.0055^{\mathrm{b}}$	0.97	$21.5 \pm 2.5$ <sup>c</sup>
	6	$19.0 \pm 1.2$ <sup>f</sup>	$-0.0803 \pm 0.0141$ <sup>b</sup>	0.92	$12.5 \pm 2.2$ <sup>c</sup>
	12	$19.3 \pm 0.8$ <sup>f</sup>	$-0.4107 \pm 0.1034$ <sup>a</sup>	0.89	$2.4 \pm 0.6^{\circ}$
	18	$20.3 \pm 1.1$ <sup>f</sup>	$-0.6005 \pm 0.1051$ <sup>a</sup>	0.94	$1.7 \pm 0.3$ <sup>c</sup>

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492 Table 2:  $z_p$  (MPa),  $\Delta V^{\neq}$  (cm<sup>3</sup> mol<sup>-1</sup>) and *ln k<sub>atm</sub>* (min<sup>-1</sup>) values of apple juice polyphenoloxidase 493 inactivation by HP-CO<sub>2</sub> treatments carried out at increasing temperature (20, 35 and 45 °C). 494 Coefficients of determination  $(R^2)$  are also shown.

<b>Temperature</b> $(^{\circ}C)$	$z_p(MPa)$	$\Delta V^{\neq}$ (cm <sup>3</sup> mol <sup>-1</sup> )	<i>ln</i> $k_{atm}$ (min <sup>-1</sup> )	${\bf R}^2$
	$20.3 \pm 5.0^{\circ}$	$-276.7 \pm 68.9^{\text{ a}}$	$-4.94 \pm 0.14^{b}$	0.89
35	$19.2 \pm 2.4^{\text{a}}$	$-307.9 \pm 38.4^{\text{a}}$	$-4.17 \pm 0.17$ <sup>ab</sup>	0.97
45	$14.8 \pm 2.6^{\text{a}}$	$-412.3 \pm 73.8$ <sup>a</sup>	$-3.13 \pm 0.31$ <sup>a</sup>	0.94

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Figure 1: Polyphenoloxidase residual activity (*RA%*) of apple juice as a function of exposure time to increasing  $CO_2$  pressures (6.0, 12.0 and 18.0 MPa) at 20 (a), 35 (b) and 45 °C (c). Samples treated at environmental pressure (0.1 MPa) were considered as control. Symbols: experimental data. Solid lines: regression lines obtained in the linear part of the curve. Inset of figure 1c: magnification of *RA%* in the 0-1 min time range.



Figure 2. Polyphenoloxidase residual activity (%) during storage at 4 °C of HP-CO<sub>2</sub> treated and untreated apple juice.



Figure 3. Absorbance at 420 nm of pasteurized, untreated and HP-CO<sub>2</sub> treated apple juices, during refrigerated storage.



Figure 4. Total viable bacteria (a), yeast (b) and lactic acid bacteria (c) counts of untreated and HP-CO<sub>2</sub> treated apple juice, during refrigerated storage.

Total viable bacteria and yeast detection limit: 10 CFU/mL

Lactic acid bacteria detection limit: 1 CFU/mL





Figure 5. Browning (a) and cooked-apple flavour (b) sensory scores of pasteurized, untreated and HP-

 $CO<sub>2</sub>$  treated apple juice, during refrigerated storage.