



UNIVERSITÀ  
DEGLI STUDI  
DI UDINE

Università degli studi di Udine

Design of Roll-In Margarine Analogous by Partial Drying of Monoglyceride-Structured Emulsions

*Original*

*Availability:*

This version is available <http://hdl.handle.net/11390/1206318> since 2025-01-15T11:55:29Z

*Publisher:*

*Published*

DOI:10.1002/ejlt.202000206

*Terms of use:*

The institutional repository of the University of Udine (<http://air.uniud.it>) is provided by ARIC services. The aim is to enable open access to all the world.

*Publisher copyright*

(Article begins on next page)



### Design of roll-in margarine analogous by partial drying of monoglyceride-structured emulsions

Journal:	<i>European Journal of Lipid Science and Technology</i>
Manuscript ID	Draft
Wiley - Manuscript type:	Research Article
Date Submitted by the Author:	n/a
Complete List of Authors:	Calligaris, Sonia; University of Udine, Department of Agricultural, Food, Environmental and Animal Sciences Plazzotta, Stella; University of Udine, Department of Agricultural, Food, Environmental and Animal Sciences Barba, Luisa; Institute of Crystallography National Research Council Trieste Branch Manzocco, Lara; University of Udine, Department of Agricultural, Food, Environmental and Animal Sciences
Keywords:	fat substitution, monoglyceride, rheological properties, laminated products, puff pastry
Additional Keywords (select from list):	

SCHOLARONE™  
Manuscripts

# Graphical Abstract

Monoglyceride-structured emulsion



- 47.6% water
- Soft



*Partial gentle drying*

Partially-dried emulsion



- 20% water
- Plastic and laminable



*Baking trials*

Puff pastry



- Comparable to roll-in-margarine puff pastry

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41

**Legend Graphical Abstract**

The application of a gentle drying procedure at 30 °C was used to turn soft monoglyceride-structured O/W emulsions into plastic and laminable systems. The latter showed a water content and rheological properties analogous to those of a conventional roll-in margarine and similar performances in baking trials of puff pastry.

For Peer Review

1  
2  
3  
4 1 **Title: Design of roll-in margarine analogous by partial drying of monoglyceride-structured**  
5  
6 2 **emulsions**

7  
8  
9  
10 3 **Running Title: Novel monoglyceride-emulsions for margarine replacement**

11  
12  
13 4 Sonia Calligaris<sup>1</sup>, Stella Plazzotta<sup>1\*</sup>, Luisa Barba<sup>2</sup>, Lara Manzocco<sup>1</sup>

14  
15  
16 5 <sup>1</sup>Department of Agricultural, Food, Environmental and Animal Sciences, University of Udine, Via  
17  
18 6 Sondrio 2/A, 33100 Udine, Italy

19  
20  
21  
22 7 <sup>2</sup>Crystallography Institute of CNR, 34100 Trieste, Italy.

23  
24  
25 8 \*Corresponding Author: stella.plazzotta@uniud.it; Tel: +39 0432-558137

26  
27  
28 9 **Abstract**

29  
30  
31 10 A novel process is proposed to turn soft gelled emulsions containing 4 or 9 g/100 g saturated  
32  
33 11 monoglyceride (MG4 and MG9) and 52.4 g/100 g water into a fat material with the typical water content  
34  
35 12 (20 g/100 g) and rheological properties of roll-in margarines. A gentle drying procedure at 30 °C was  
36  
37 13 found to allow partial water removal from the emulsions, without causing phase separation. As compared  
38  
39 14 to the initial emulsions, the partially dried systems (MG4-dried and MG9-dried) showed a more solid-  
40  
41 15 like appearance and a microstructure with oil droplets tightly packed into the MG-network. Calorimetric  
42  
43 16 (DSC) and crystal organization (XRD) analyses showed that drying modified neither the amount of  
44  
45 17 crystallized material nor induced MG transition toward a different lamellar organization. Moreover, the  
46  
47 18 MG9-dried emulsion showed rheological properties ( $G' = 4.6 \times 10^5$  Pa, critical stress = 634 Pa)  
48  
49 19 comparable to those of commercial roll-in margarine and was successfully used in baking trials of puff  
50  
51 20 pastry.

1  
2  
3 21 **Practical applications:** The last years have seen an increasing demand of strategies for the nutritional  
4  
5 22 improvement of lipid-containing foods, due to the emerging health concern related to the consumption  
6  
7 23 of high quantities of saturated and trans fatty acids. However, the unique technological and sensory  
8  
9 24 properties of fats make this task very challenging, especially considering the production of laminated  
10  
11 25 products requiring the use of plastic fat materials with peculiar rheological properties. The results  
12  
13 26 obtained in this study are very promising, as they show that the MG9-dried fat can be used as a trans-  
14  
15 27 free alternative to traditional laminating margarine.  
16  
17  
18  
19

20 28 Key-words: fat substitution/monoglyceride/rheological properties/laminated products/puff pastry  
21  
22

23 29 Abbreviations: MG, monoglycerides  
24  
25  
26

## 27 30 **1 Introduction**

  
28

29 31 Food fats, including animal fats (e.g. butter and lard), tropical oils (e.g. palm oil, palm kernel oil and  
30  
31 32 coconut oil) as well as margarine and shortenings containing them, are multi-purpose ingredients used  
32  
33 33 to modulate food structure (aeration, lightness), rheological properties (plasticity, texture), and sensory  
34  
35 34 characteristics (taste, colour, flavour, crispiness, creaminess) [1]. They have the common characteristic  
36  
37 35 to be spreadable at room temperature thanks to the presence of a fat crystal network made of  
38  
39 36 triacylglycerols [2]. In this context, saturated and trans fatty acids play a determining role being  
40  
41 37 crystalline at room temperature. Despite unique fat technological functionalities, the last years have seen  
42  
43 38 an increasing demand for strategies for the nutritional improvement of lipid-containing foods. This is due  
44  
45 39 to the emerging health concern related to the consumption of high quantities of saturated and trans fats.  
46  
47 40 The latter are well known to be associated with an increased risk of diet-related diseases and pathological  
48  
49 41 conditions, including cardiovascular diseases, obesity and type II diabetes [3,4]. The World Health  
50  
51 42 Organization (WHO) recommends the amount of total, saturated and trans fats to be less than 30% and  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

43 10% of the total energy intake, respectively [5]. At the same time, the European Commission in 2019  
44 imposed a limitation of trans fats to 2 g per 100 g of total fats [6].

45 In this context, the most challenging fats to be substituted are those used to produce laminated products,  
46 such as puff pastry and croissants, called roll-in fats or laminating fats [7]. They mainly include  
47 margarines that are water-in-oil emulsions containing at least 80 g/100 g fat, of which more than 50% is  
48 represented by saturated and/or trans fatty acids [8,9]. Upon lamination, these roll-in-fats are shaped into  
49 several thin sheets which inter-lie among an equal number of dough layers. Laminated products owe their  
50 peculiar physical and sensory properties to this laminated structure: during baking, in fact, dough water  
51 evaporates, causing the expansion of the dough layers, which bake separately, due to the alternate fat  
52 layers [10]. Roll-in-fats should present specific characteristics, as described by Blake and Marangoni  
53 [11]. They need to be plastic enough to withstand the lamination process without losing continuity. To  
54 this aim, their rheological properties should be similar to the dough ones and their melting temperature  
55 higher than that of the lamination process, to avoid fat melting and prevent dough layers from coming in  
56 contact. Moreover, it should be underlined that the presence of water in margarine is required for optimal  
57 lubrication, plasticity and leavening [12]. Water is entrapped in the continuous matrix of fat crystals as  
58 tiny droplets, presenting a diameter of about 1  $\mu\text{m}$ , which accounts for  $5\text{-}10 \times 10^9$  droplets per mL of  
59 product [13]. Although finely dispersed, water droplets are combined loosely enough for the emulsion to  
60 break easily upon melting of the crystalline continuous phase, which favours puff during product baking  
61 [14].

62 Different Authors proposed roll-in-fat mimetics with low saturated fat content [15–20]. However, these  
63 strategies resulted in fat materials which are not comparable to margarines, since not containing water  
64 [21].

65 An alternative strategy could be based on the use of MG-structured emulsions, which have actually  
66 shown good performances as fat substitutes in different baked goods, including short dough pastry [22],

1  
2  
3 67 sweet bread [23], cookies [24]; ice-cream [25]. The use of MG is not new and ternary gelled systems  
4  
5 68 containing saturated monoglycerides were firstly proposed by Marangoni and co-workers in 2007 [26–  
6  
7 69 28]. At high water content, the system appears as a liquid emulsion in which MG crystalline bilayers  
8  
9  
10 70 surround and stabilize oil droplets. Increasing the MG content, the MG walls extend from one droplet to  
11  
12 71 the next one, leading to a continuous solid network, which generates a gel-like material with the  
13  
14 72 rheological features typical of spreadable fats. The final gel structure is affected by several compositional  
15  
16 73 and processing factors, such as MG and other ingredient type and concentration, pH, temperature and  
17  
18 74 application of shear. In any case, it exists a defined range of compositional variables in which the system  
19  
20 75 equilibrium is not assured and phase separation could occur [27,29,30]. It should be noted that it is not  
21  
22 76 possible to prepare systems simulating margarine rheological properties and fat/water ratio by simply  
23  
24 77 increasing the MG content. In fact, the relative increase in MG requires more water and a concomitant  
25  
26 78 decrease in the oil content to form a water-continuous phase in which the MG interact giving a stable gel  
27  
28 79 [27,28]. In this regard, Blake and Marangoni [11] affirm that to ensure proper hydration and sufficient  
29  
30 80 oil-droplet encapsulation, the water content of these emulsions must not be lower than 30 g/100 g. To  
31  
32 81 improve the rheological properties of MG emulsions, waxes were added but the final water content (35–  
33  
34 82 40 g/100 g) was in any case higher than that expected in margarine [11].  
35  
36  
37  
38  
39

40 83 Based on these considerations, in this work, a novel process is proposed to turn soft MG gelled emulsions  
41  
42 84 into fat materials with the typical water content (20 g/100 g) and rheological properties of roll-in  
43  
44 85 margarines. In particular, we propose to apply a gentle drying to remove part of the water from the  
45  
46 86 aqueous network surrounding oil droplets in MG emulsions, without disrupting their ternary structure.  
47  
48 87 The partially dried structured emulsions were analyzed for microstructure, calorimetric (DSC), crystal  
49  
50 88 organization (XRD) and rheological properties. Finally, the best performing sample was selected and  
51  
52 89 used in baking trials of puff pastry.  
53  
54  
55  
56  
57  
58  
59  
60

## 90 2 Materials and Methods

### 91 2.1 Materials

92 Myverol™ saturated monoglyceride (MG) were kindly supplied by Kerry Ingredients and Flavour  
93 (Bristol, UK), palmitic, stearic acid, NaOH were purchased from Sigma Aldrich (Milan, Italy),  
94 ingredients for puff pastry were purchased in a local market; commercial margarines were selected  
95 among those typically used for the industrial preparation of laminated baked goods and kindly provided  
96 by local food companies.

### 97 2.2 Structured emulsion preparation

98 MG structured emulsions were prepared according to Calligaris, Da Pieve, Arrighetti, and Barba [31].  
99 Briefly, the oil phase was composed of MG (4 and 9 g/100 g), a co-surfactant mixture (0.8 and 1.8 g/100  
100 g) of palmitic and stearic acid (weight ratio 1:1) and sunflower oil (47.6 and 41.6 g/100 g). The water  
101 phase (52.4 g/100 g) was composed of 1 mM NaOH (pH 11) in deionized milli-Q water to promote the  
102 partial ionization of the co-surfactant mixture and obtain a properly swollen phase. The samples were  
103 prepared by mixing the water solution and the oil phase previously heated at 78 °C in a water bath and  
104 homogenizing by using a high-speed homogenizer DI 25 (Ika-Werke, Staufen, Germany) at 59,000g for  
105 1 min. Finally, the mixture was cooled at 4 °C in an ice bath and then stored at 4 °C for 24 h before usage.  
106 The obtained structured emulsions were called MG4 and MG9.

### 107 2.3 Structured emulsion drying

108 The drying process was conducted according to the method patented by our research group [32]. The  
109 structured emulsions were spread in a thin layer ( $3.0 \pm 0.5$  mm), as measured by a CD-15APXR digital  
110 calliper (Absolute AOS Digimatic, Mitutoyo Corporation, Kanagawa, Japan) and dried at 30, 40 and 50  
111 °C using a professional oven (Air-o-steam, 10 GN 1/1, Electrolux, Pordenone, Italy), with an airflow set  
112 at the medium level. The drying process was continued until the structured emulsions reached a lipid

1  
2  
3 113 phase weight concentration of 80 g/100 g, which was determined by weighting the samples at defined  
4  
5 114 time intervals. The partially dried structured emulsions were called MG4-dried and MG9-dried.  
6  
7

#### 8 115 2.4 *Puff pastry preparation*

9

10  
11 116 Puff pastry was prepared using commercial laminating palm margarine (Control) and the MG9-dried  
12  
13 117 sample (36.9 g/100 g). Plain flour (47.7 g/100 g) was mixed with fine salt (0.7 g/100 g) and water (14.7  
14  
15 118 g/100 g) using a domestic mixer (Kenwood Chef KVC3110S, Milan, Italy) until obtaining homogeneous  
16  
17 119 dough (250 g). The latter was placed on a paper-lined sheet pan dusted with flour, covered, and left to  
18  
19 120 rest for 30 min at 4 °C. Then, the dough was rolled into a square of 45×45 cm. Similarly, the commercial  
20  
21 121 margarine and the MG9-dried sample were shaped into squares of about 31×31 cm, and puff pastry was  
22  
23 122 produced as described by Patient [12], with 2 h rest at 4 °C between subsequent folding steps. Puff pastry  
24  
25 123 was finally rolled up to 3-mm thickness and cut into 4×5 cm rectangles, which were baked at 180 °C for  
26  
27 124 9 min (Air-o-steam, 10 GN 1/1, Electrolux, Pordenone, Italy) and cooled at 20 °C before analyses.  
28  
29  
30  
31

#### 32 125 2.5 *Analytical determinations*

33  
34

##### 35 126 2.5.1 *Image acquisition*

36  
37

38 127 Sample images were acquired using an image acquisition cabinet (Immagini & Computer, Bareggio,  
39  
40 128 Italy) equipped with a digital camera (EOS 550D, Canon, Milano, Italy). The digital camera was placed  
41  
42 129 on an adjustable stand positioned 45 cm above a black or white cardboard base where the samples were  
43  
44 130 placed. The light was provided by 4 100 W frosted photographic floodlights, in a position allowing  
45  
46 131 minimum shadow and glare. Images were saved in jpeg format resulting in 3456×2304 pixels.  
47  
48  
49  
50

##### 51 132 2.5.2 *Optical microscopy*

52

53 133 Samples were placed on a glass slide, covered with a cover slide and observed using a Leica DM 2000  
54  
55 134 optical microscope (Leica Microsystems, Heerbrugg, Switzerland). The images were taken at 200×  
56  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

135 magnification using a Leica EC3 digital camera and elaborated with the Leica Suite Las EZ software  
136 (Leica Microsystems, Heerbrugg, Switzerland).

### 2.5.3 Differential scanning calorimetry

The calorimetric analysis was carried out using a TA4000 differential scanning calorimeter (Mettler-Toledo, Greifensee, Swiss) connected to a GraphWare software TAT72.2/5 (Mettler-Toledo, Greifensee, Swiss). Heat flow calibration was achieved using indium (heat of fusion 28.45 J/g). Temperature calibration was carried out using hexane (m.p.  $-93.5\text{ }^{\circ}\text{C}$ ), water (m.p.  $0.0\text{ }^{\circ}\text{C}$ ) and indium (m.p.  $156.6\text{ }^{\circ}\text{C}$ ). Samples were prepared by carefully weighing 15–20 mg of the sample in 160  $\mu\text{L}$  aluminium DSC pans, closed with hermetic sealing. An empty pan was used as a reference. Samples were heated under nitrogen flow (0.5 mL/min) from 20 to  $80\text{ }^{\circ}\text{C}$  at  $5\text{ }^{\circ}\text{C}/\text{min}$ . The start of the melting transition was taken as on-set ( $T_{on}$ ) points of transition, that are the points at which the extrapolated baseline intersects the extrapolated tangent of the calorimetric peak in the transition state. Results were normalized to account for the weight variation of the samples. Total peak enthalpy was obtained by integration. The program STAR ever. 8.10 (Mettler-Toledo, Greifensee, Swiss) was used to plot and analyze the thermal data.

### 2.5.4 X-ray diffraction

X-ray diffraction patterns were recorded at the X-ray Diffraction beamline 5.2 at the Synchrotron Radiation Facility Elettra located in Trieste (Italy). The X-ray beam emitted by the wiggler source on the Elettra 2 GeV electron storage ring was monochromatized by a Si(1 1 1) double crystal monochromator, focused on the sample and collimated by a double set of slits giving a spot size of  $0.2\times 0.2\text{ mm}$ . A drop of the sample was lodged into a nylon pre-mounted cryoloop 20 micron for crystallographic experiments (0.7–1.0 mm) (Hampton Research HR4-965, Aliso Viejo, CA, USA). Sample temperature was controlled using a 700 series cryo-cooler (Oxford Cryosystems, Oxford, UK) with an accuracy of  $1\text{ }^{\circ}\text{C}$ . Analyses were performed at  $20\text{ }^{\circ}\text{C}$ . Data were collected at a photon energy of 8.856 keV ( $\lambda = 1.4\text{ \AA}$ ), using a 2M

1  
2  
3 158 Pilatus silicon pixel X-ray detector (DECTRIS Ltd., Baden, Switzerland). Bidimensional patterns  
4  
5 159 collected with Pilatus were calibrated by means of a LaB6 standard and integrated using the software  
6  
7  
8 160 FIT2D [33] obtaining a powder-like pattern for every sample experiment. The angular range (two theta)  
9  
10 161 of observed diffraction ring spanned from  $1.3^\circ$  to  $21^\circ$ , corresponding to interplanar distances  $d$  from 60  
11  
12 162 Å to 3.9 Å and to reciprocal lattice spacings  $q$  from  $0.1 \text{ \AA}^{-1}$  to  $1.62 \text{ \AA}^{-1}$ .

### 16 163 2.5.5 Rheology

18 164 The viscoelastic properties (moduli  $G'$  and  $G''$ ) of the samples were tested using an RS6000 Rheometer  
19  
20 165 (Thermo Scientific RheoStress, Haake, Germany), equipped with a Peltier system for temperature  
21  
22 166 control. Measures were performed using a parallel plate geometry at  $15^\circ\text{C}$  with a gap of 1.0 mm.  
23  
24 167 Oscillatory sweep tests to identify the linear viscoelastic region (LVR) were performed increasing stress  
25  
26 168 from 0.1 to 3,000 Pa at 1 Hz frequency. Critical stress (Pa) was identified as the strain value  
27  
28 169 corresponding to a 10% drop in  $G'$  value.  
29  
30  
31  
32

### 33 170 2.5.6 Colour

35  
36 171 The puff pastry surface colour was measured using a tristimulus Chromameter-2-Reflectance  
37  
38 172 colourimeter (Minolta, Osaka, Japan) with a CR-300 measuring head, standardized against a white tile,  
39  
40 173 and data were expressed in  $L^*$ ,  $a^*$  and  $b^*$  scale.  
41  
42

### 44 174 2.5.7 Leavening capacity

46 175 The leavening capacity of puff pastry was determined by measuring the volume gain (%) before and after  
47  
48 176 baking ( $\text{cm}^3\text{g}^{-1}$ ) by using the rapeseed displacement method [34].  
49  
50

### 52 177 2.5.8 Sensory analysis

54 178 A group of 13 judges was used for the sensory evaluation of puff pastry samples. Judges were not trained  
55  
56 179 on sensory analysis of puff pastry but they were experts in the use of the selected sensory method. Control  
57  
58  
59  
60

1  
2  
3 180 puff pastry prepared with commercial margarine and the sample prepared with the MG9-dried emulsion  
4  
5 181 were evaluated. Judges were asked to evaluate the intensity of firmness, crispiness and oiliness of the  
6  
7  
8 182 puff pastry samples, which were identified by 3-digit random codes. The selected descriptors were  
9  
10 183 evaluated on a 1-9 point hedonic scale, in which 1 corresponded to “extremely low descriptor intensity”,  
11  
12 184 and 9 to “extremely high descriptor intensity” [35].  
13  
14

### 15 185 2.5.9 Data elaboration

16  
17  
18 186 All determinations were expressed as the mean  $\pm$  standard error of at least three repeated measurements  
19  
20 187 from two experiment replicates. Statistical analysis was performed by using R v. 3.0.2 (The R Foundation  
21  
22 for Statistical Computing). Student’s t-test was used to determine statistically significant differences  
23 188 between means ( $p < 0.05$ ).  
24  
25 189

26  
27 190 The kinetics constants of the drying processes ( $k, h^{-1}$ ) were obtained by fitting the drying data, expressing  
28  
29 the moisture weight percentage as a function drying time ( $t, h$ ), by using the model proposed by Lewis  
30 191  
31 [36] (eq. 1):  
32 192  
33

$$34 \frac{X - X_e}{X_0 - X_e} = \exp(-kt) \quad (\text{eq. 1})$$

35  
36 193 where  $X, X_e$  and  $X_0$  are the moisture content (g/100 g) at any time, equilibrium and initial, respectively.  
37  
38

39 194 Data fitting was performed by using TableCurve2D software (Jandel Scientific, ver. 5.01). The goodness  
40  
41 of fit was evaluated based on statistical parameters of fitting ( $R^2, p$ ).  
42 195  
43  
44 196  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

### 197 **3 Results and Discussion**

#### 198 *3.1 Structured emulsion drying*

199 Water-in-oil emulsions containing 47.6 g/100 g water and 4 or 9 g/100 g monoglycerides were prepared  
200 and characterized. Taking into consideration the MG content, the samples were named as MG4 and MG9,  
201 respectively. As expected and in agreement with the literature, the obtained systems were self-standing  
202 white emulsions, as well visible from sample images and micrographs reported in Figure 1. This peculiar  
203 structure is due to the self-assembling of saturated monoglycerides, which form walls surrounding oil  
204 droplets [26–28,31]. The latter resulted smaller in the case of the emulsion containing 9 g/100 g MG, due  
205 to the higher amount of MG available for the stabilization of the increased surface area associated with  
206 smaller oil droplets [31].

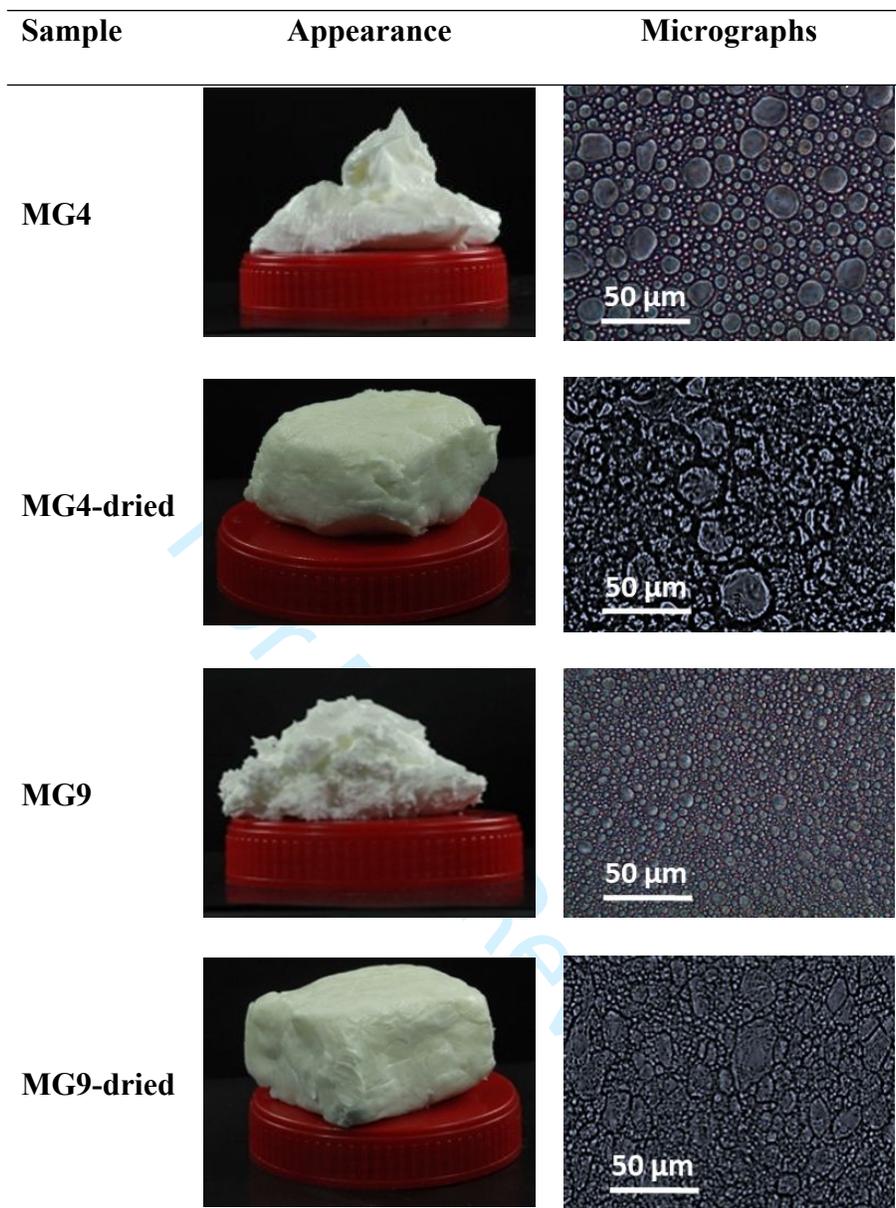


Figure 1. Appearance and microscopic images of structured emulsions containing 4 and 9 g/100 g monoglycerides before (MG4, MG9) and after drying (MG4-dried, MG9-dried).

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

209 These systems were dried in a thin layer at 30, 40 and 50 °C to partially remove water and obtain systems  
210 with 80 g/100 g lipid content, analogous to that of traditional margarine [21]. As previously mentioned,  
211 it was not possible to prepare systems containing such lipid ratio by simply following the emulsion  
212 preparation procedure. In this case, in agreement with the literature, a non-homogeneous liquid system  
213 was obtained, in which the lipid and water phase were separated [11].

For Peer Review

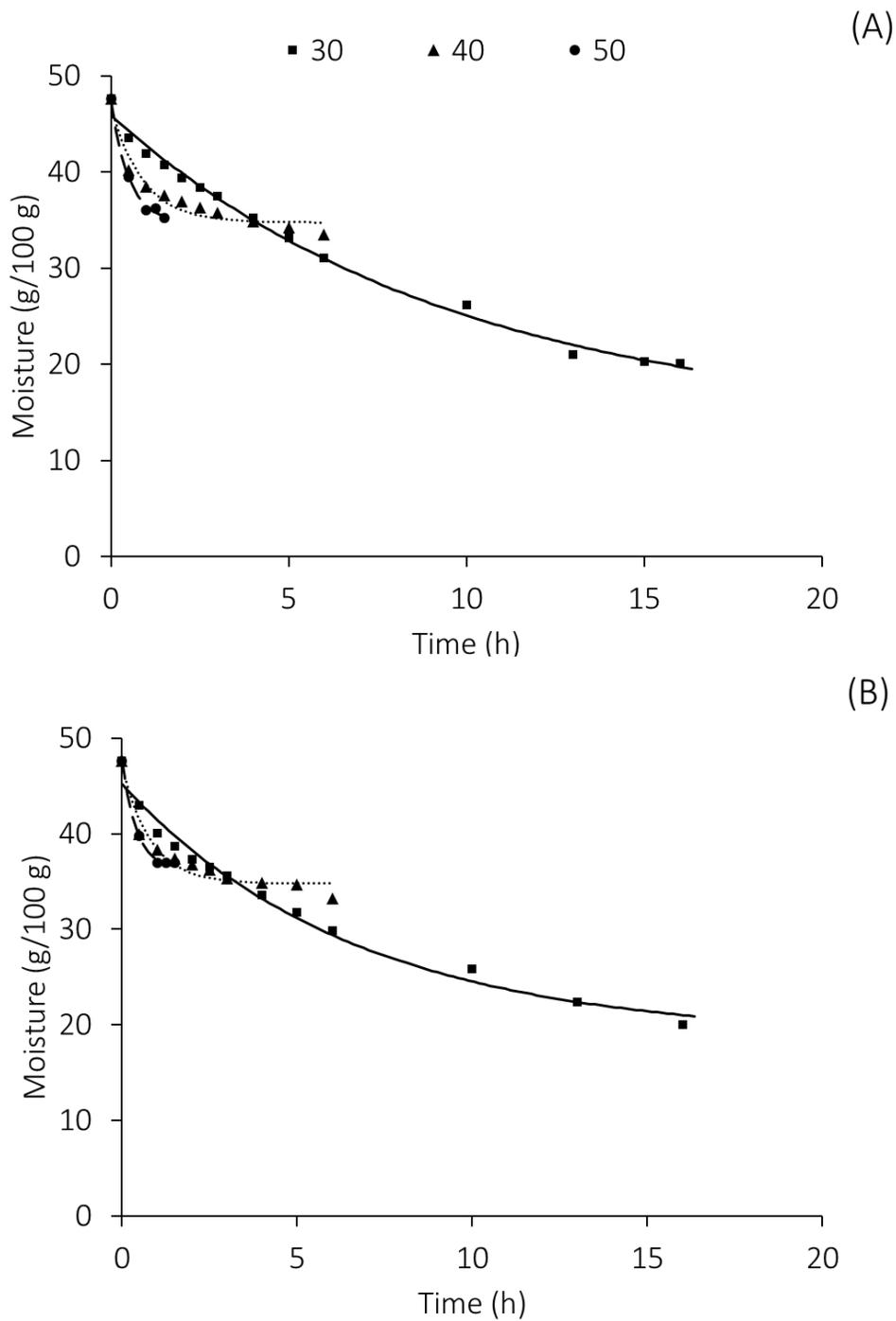


Figure 2. Moisture of structured emulsions containing 4 g/100 g (A) and 9 g/100 g (B) monoglycerides as a function of drying time at 30, 40 and 50 °C (symbols). Fitting curves of drying model are also reported (lines).

1  
2  
3 218 Figure 2 reports the moisture (g/100 g) of the emulsions as a function of drying time at the different  
4  
5 219 temperatures. During drying, water content showed an exponential decrease, with an initial fast decrease-  
6  
7  
8 220 phase followed by a tendency to reach a plateau. Experimental data were fitted to a classical exponential  
9  
10 221 decay model developed for thin-layer drying [36] to obtain the kinetic constants reported in Table 1. The  
11  
12 222 model well-explained experimental data, as shown by the  $R^2$  values. Moreover, independently on the  
13  
14 223 MG content, as expected, the increase in drying temperature provoked an increase in the drying rate,  
15  
16  
17 224 resulting in higher kinetic constant values ( $k$ , Table 1). However, it was not possible to take the samples  
18  
19 225 at the desired 20 g/100 g moisture value at both 40 and 50 °C. At these temperatures, independently on  
20  
21 226 the MG content, the systems broke down and leaked oil at moisture values of 33 and 35 g/100 g,  
22  
23  
24 227 respectively, corresponding to a lipid content of 67 and 65 g/100 g. These results show that a drying  
25  
26 228 temperature higher than 30 °C is associated with the destruction of the MG network responsible for  
27  
28 229 emulsion structuration. By contrast, gentle drying at 30 °C allowed obtaining emulsions with 80 g/100 g  
29  
30  
31 230 lipid phase after about 16 h. It must be underlined that a further water removal at 30 °C from both MG4  
32  
33 231 and MG9 sample resulted in emulsion break-down and oil leaking. These results suggest that the stability  
34  
35 232 of the samples during drying was affected by both the drying rate and the water removal extent. In this  
36  
37  
38 233 regard, it is known that water is required to induce MG self-assembling into a supra-molecular  
39  
40 234 organization able to encapsulate oil and water among the MG lamellae [27,28]. Moreover, as evidenced  
41  
42 235 in the literature [27], the temperature should be strictly controlled during both emulsification and cooling  
43  
44  
45 236 phase to properly structure the gel. In particular, the increase in temperature during emulsification has  
46  
47 237 been reported to affect the hydration state of the MG, decrease the electrostatic charge magnitude and  
48  
49 238 also induce changes in the mesomorphic phase, leading to a greater tendency to aggregation and thus  
50  
51 239 phase separation [27]. This evidence supports the hypothesis that also during drying, an accurate  
52  
53  
54 240 optimization of temperature and thus of water removal rate, is required. Based on the acquired results, a  
55  
56 241 slow drying rate seems to allow the reorganization of the MG network surrounding oil droplets, while  
57  
58  
59  
60

1  
2  
3 242 preventing phase separation. By contrast, an increase in drying rate probably induced MG aggregation,  
4  
5 243 thus favouring system break-down.  
6

7  
8 244 Based on these results, the temperature of 30 °C was selected to partially dry the structured emulsions  
9  
10 245 MG4 and MG9 until reaching a water content of 20 g/100 g, comparable to that of conventional  
11  
12 246 margarine. Upon drying, the MG concentration of the emulsions initially containing 4 and 9 g/100 g MG  
13  
14 247 reached a concentration of 6.1 and 13.7 g/100 g, respectively. The obtained samples, called MG4-dried  
15  
16  
17 248 and MG9-dried, are shown in Figure 1. As compared to the initial MG4 and MG9 emulsions, the partially  
18  
19 249 dried systems showed a more solid-like appearance and a microstructure with oil droplets tightly packed  
20  
21  
22 250 into a network consisting of MG (Figure 1). The oil droplets present in the original structure were  
23  
24 251 deformed after drying, evolving from a spherical shape to a polygonal one. These results agree with  
25  
26 252 literature studies in which the emulsion-template method is used for oil structuring. This approach is  
27  
28  
29 253 based on complete water evaporation from highly concentrated oil-in-water emulsions, which results in  
30  
31 254 the trapping of the oil in a physical network of biopolymers including polysaccharides and proteins or  
32  
33 255 their combination, originally forming the oil-water interface [37–40].  
34

35 256 Differential scanning calorimetry measurements were carried out on the emulsions before and after  
36  
37  
38 257 drying to identify the effects of drying on the melting temperature and the enthalpy of the transition.  
39  
40 258 Figure 3 shows the heating curves of the emulsions before and after drying.  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

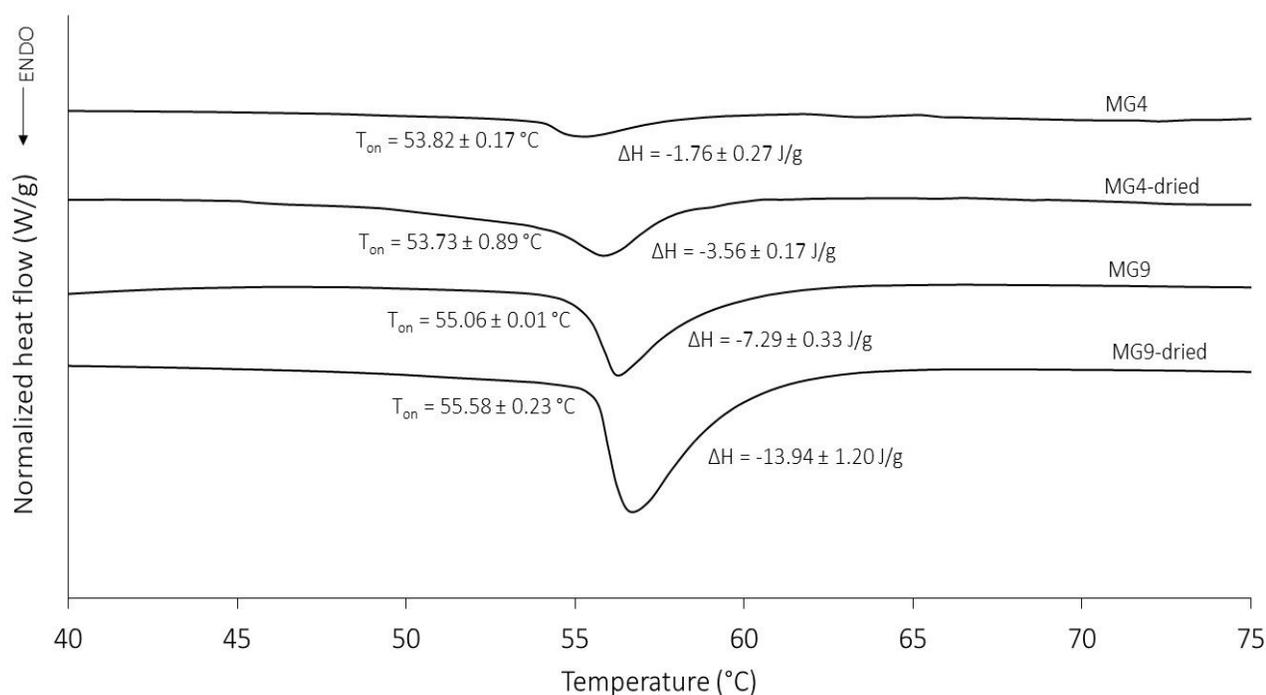


Figure 3. Calorimetric melting curves of structured emulsions before (MG4, MG9) and after (MG4-dried, MG9-dried) drying. Onset melting temperature ( $T_{on}$ ) and melting enthalpy ( $\Delta H$ ) are also reported.

All systems showed one endothermic peak corresponding to the disruption of the network among MG lamellar bilayers, followed by the melting of the crystalline MG [30]. The melting enthalpy ( $\Delta H$ ) expressed per g of emulsion resulted well correlated with the MG and co-surfactant content ( $R^2 = 0.993$ ), being proportional to the quantity of crystallised material in the matrix [26,27,30]. By contrast, the melting enthalpy per g of saturated fatty acids was not significantly different in the MG4 and MG9 emulsions (data not shown), neither was affected by drying ( $p \geq 0.05$ ). Moreover, no significant differences between sample melting temperature ( $T_{on}$ ) before and after drying ( $p \geq 0.05$ ) were observed. These results could be associated with the fact that drying did not significantly modify the amount of crystallized material in the samples.

1  
2  
3 272 Synchrotron XRD diffraction patterns were also acquired to obtain information on the effect of drying  
4  
5 273 on MG crystal packing in the emulsions. The patterns, reported in Figure 4, resulted similar to those  
6  
7  
8 274 reported in the literature for analogous samples [27,28,30,31]. The diffraction signal superimposes onto  
9  
10 275 two large amorphous halos centred respectively about  $d = 23 \text{ \AA}$  ( $q = 0.27 \text{ \AA}^{-1}$ ) and  $d = 4.42 \text{ \AA}$  ( $q = 1.4$   
11  
12 276  $\text{\AA}^{-1}$ ). Two lamellar forms appear to coexist, whose lower angle visible peaks correspond to average  
13  
14 277 interplanar distances of  $d = 62 \text{ \AA}$  ( $q = 0.1 \text{ \AA}^{-1}$ ) and  $d = 48 \text{ \AA}$  ( $q = 0.13 \text{ \AA}^{-1}$ ), respectively (Figure 4). All  
15  
16  
17 278 other stakes around  $4 \text{ \AA}$  are characteristic of MG high-angle diffraction patterns, the most intense being  
18  
19 279 at  $4.54$  and  $4.12 \text{ \AA}$ . All the samples presented similar patterns in the wide angle region, due to the in-  
20  
21 280 plane chain ordering of MG crystals in the  $L\beta$  phase [28,31]. Based on the fact that there were no  
22  
23  
24 281 differences in the d-spacing of the samples in the wide-angle region, it can be concluded that they  
25  
26 282 presented the same in-plane molecular ordering [31]. In the small angle region, the diffraction signals of  
27  
28 283 the two lamellar phases appear to be much more intense with respect to the amorphous phase in sample  
29  
30  
31 284 MG9 than in the sample MG4. In both cases, no structural differences were noted before and after drying,  
32  
33 285 and no transition toward a different lamellar organization occurred upon drying.  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

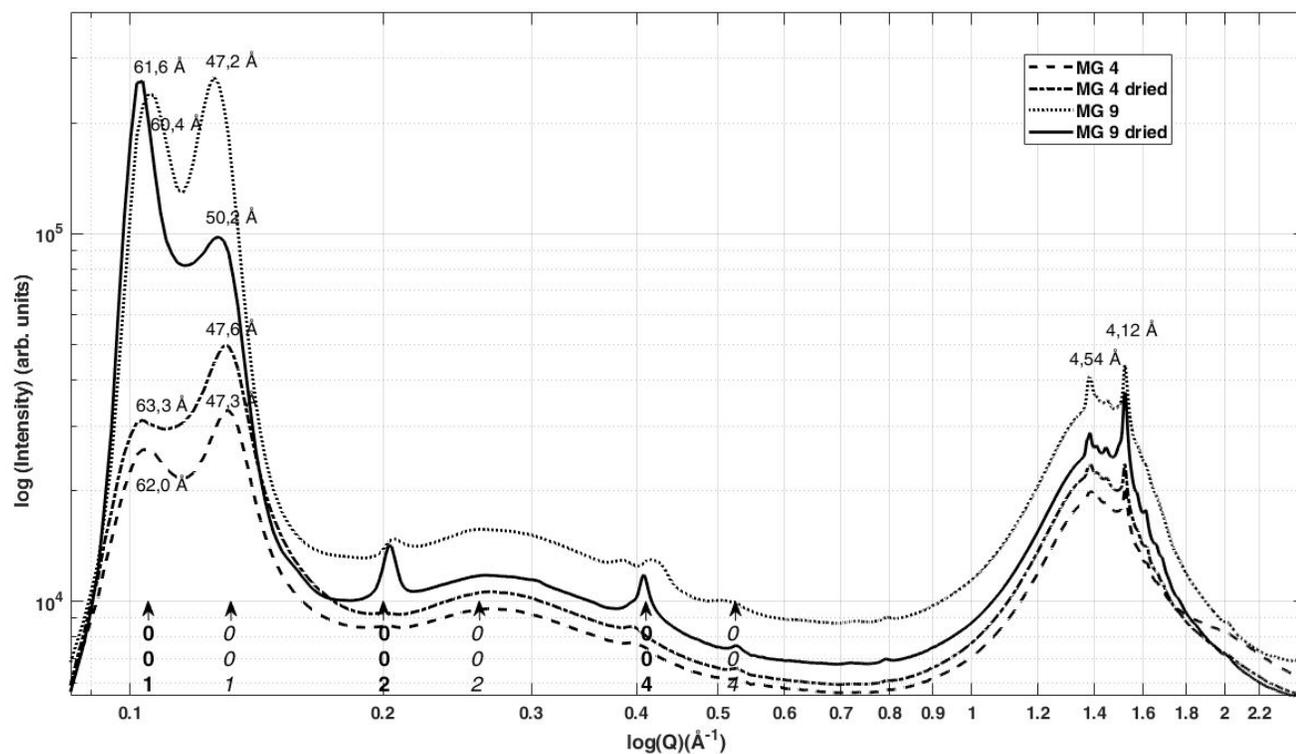


Figure 4. Powder X-ray diffraction spectra of structured emulsions before (MG4, MG9) and after (MG4-dried, MG9-dried) drying. Patterns plotted as a function of reciprocal lattice spacing  $q$ , where  $q = 2\pi/d = (4\pi/\lambda) \cdot \sin\theta$ ,  $d$  is the lattice spacing,  $\lambda$  is the X-ray wavelength and  $2\theta$  is the Bragg scattering angle, represented as log-log plots opportunely scaled and shifted for clarity.

Finally, the rheological characteristics of the samples before and after drying were studied (Supplementary Figure 1; Table 2).



1  
2  
3 308 G' and critical stress have been previously indicated as key parameters for comparing the rheological  
4  
5 309 properties of MG structured emulsions with those of traditional solid fats [11]. Thus, rheological  
6  
7  
8 310 properties of dried MG emulsions were compared to those of three commercial laminating fats with 20  
9  
10 311 g/100 g water content. In agreement with literature data [41,42], the latter gave G' and critical stress  
11  
12 312 values in the range  $8.2\text{-}26.7\times 10^5$  Pa and  $263.7\text{-}995.4$  Pa, respectively. With reference to these ranges, the  
13  
14 313 MG4-dried sample showed critical stress and G' values about 2 and 10 times lower, respectively. By  
15  
16  
17 314 contrast, the MG9-dried emulsion showed rheological features comparable to those of the analyzed  
18  
19 315 commercial laminating fats (Table 2). Such rheological features have been previously obtained in  
20  
21 316 analogous MG structured emulsions by adding at least 15 g/100 g waxes or substituting sunflower oil  
22  
23  
24 317 with palm oil [11]. However, the first systems showed unpleasant waxy taste, while the seconds require  
25  
26 318 the addition of saturated fatty acids.

### 29 319 3.2 Dried structured emulsions as fat alternatives in puff pastry

30  
31  
32 320 Based on its promising features, the MG9-dried sample was used in baking trials in the production of  
33  
34 321 puff pastry and compared to a common roll-in laminating palm margarine showing a G' and critical stress  
35  
36 322 of  $15.7\times 10^5$  and  $995.4$  Pa, respectively. Some of the preparation phases of the puff pastry using MG9-  
37  
38  
39 323 dried sample are reported in the Supplementary Figure 2.



Supplementary Figure 2. Preparation phases of puff pastry using MG9-dried sample as laminating fat substitute.

During both puff pastry preparation and baking, the MG9-dried emulsion showed the typical features required to a laminating fat. Thanks to its plasticity, the MG9-dried sample was easily laminated into thin sheets between dough layers during the folding process, during which it did not melt, preventing the dough layers from touching. Therefore, upon lamination and folding, clearly defined and continuous dough and MG9-dried layers were visible, confirming the optimal rheological features of the MG9-dried sample. Finally, during baking, no evident oil leaking was observed, which indicates that the fat phase was retained between the dough layers. This allowed the typical physical leavening of puff pastry, as well-evident from the images reported in Figure 5.

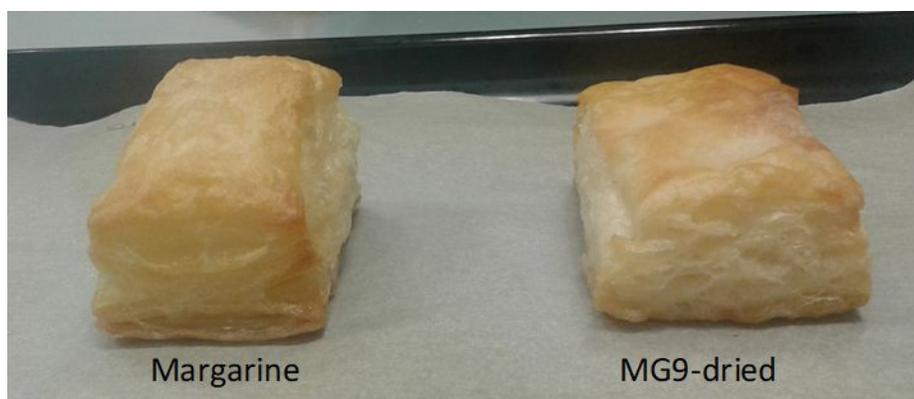


Figure 5. Image of puff pastry prepared with commercial laminating margarine and MG9-dried emulsion.

The puff pastry prepared with the MG9-dried sample displayed the flaky golden-brown appearance of puff pastry containing margarine. As shown in Table 3, the only observable difference was that MG9-dried puff pastry showed a lighter colour, as indicated by the lower absolute values of the colourimetric parameters  $a^*$  and  $b^*$  and a 16% lower leavening capacity. Moreover, it presented comparable instrumental firmness, which was also confirmed by the judges during sensory evaluation ( $p \geq 0.05$ ). The latter also evidenced that the substitution of the commercial margarine with the MG9-dried fat did not affect puff pastry friability ( $p \geq 0.05$ ) but resulted in an increased oiliness mouthfeel. It must also be noted that the judges did not highlight the presence of off-flavours or off-taste upon tasting the puff pastry samples. These findings are very promising, as they show that MG9-dried fat can be used as laminating fat replacement.

#### 4 Conclusions

Results here reported face the timely challenge of developing substitutes of roll-in margarine able to deliver a nutritional improvement of laminated products conventionally rich in saturated and trans fatty acids. The process here proposed for the production of the margarine substitute requires simple unit operations (e.g heating, homogenization, cooling, drying), which are commonly applied in food

1  
2  
3 354 industries, making its scaling up feasible both technically and economically. Based on the acquired  
4  
5 355 results, drying temperature is the key factor to be controlled to obtain a well-structured margarine  
6  
7  
8 356 alternative. In fact, the desired 20 g/100 g water content, typical of margarine, is only obtained by  
9  
10 357 applying a gentle and gradual drying at 30 °C. Under these conditions, the MG structure entrapping both  
11  
12 358 water and oil is maintained, avoiding phase separation. The resulting system showed good performances  
13  
14  
15 359 in both lamination and baking of puff pastry, making it an optimal candidate for the substitution of  
16  
17 360 margarine in the industrial production of laminated products with improved nutritional quality. It is not  
18  
19 361 excluded that by changing the starting emulsion formulation (e.g. oil type, other ingredients in the water  
20  
21  
22 362 phase) it would be possible to further improve the range of MG-based margarine analogues.

### 23 24 25 363 **Funding**

26  
27 364 This research did not receive any specific grant from funding agencies in the public, commercial, or  
28  
29  
30 365 not-for-profit sectors.

### 31 32 33 366 **Declaration of interest**

34  
35 367 The authors have declared no conflicts of interest.

### 36 37 38 39 368 **Permission statements**

40  
41 369 The manuscript *does not* contain experiments using animals.

42  
43 370 The manuscript *does not* contain human studies.

### 44 45 46 47 371 **References**

- 48  
49 372 [1] R.V. Rios, M.D.F. Pessanha, P.F. de Almeida, C.L. Viana, S.C. da S. Lannes, *Food Sci.*  
50  
51  
52 373 *Technol.* **2014**, *34*, 3.  
53  
54 374 [2] N.C. Acevedo, F. Peyronel, A.G. Marangoni, *Curr. Opin. Colloid Interface Sci.* **2011**, *16*, 374.  
55  
56 375 [3] D. Mozaffarian, M.B. Katan, A. Ascherio, M.J. Stampfer, W.C. Willett, *N. Engl. J. Med.* **2006**,  
57  
58  
59  
60

1  
2  
3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

376 61, 525.

377 [4] M. Briggs, K. Petersen, P. Kris-Etherton, *Healthcare* **2017**, *5*, 29.

378 [5] WHO **2020** [https://www.who.int/news-room/fact-sheets/detail/healthy-](https://www.who.int/news-room/fact-sheets/detail/healthy-diet#:~:text=Energy%20intake%20(calories)%20should%20be,1%2C%20%2C%203).)  
diet#:~:text=Energy%20intake%20(calories)%20should%20be,1%2C%20%2C%203).

380 [6] EC/2019/649, *Off. J. Eur. Union* **2019**, L 110/17.

381 [7] B. Huschka, C. Challacombe, A.G. Marangoni, K. Seetharaman, *Cereal Chem.* **2011**, *88*, 253.

382 [8] B. Macias-Rodriguez, A.G. Marangoni, *JAOCs, J. Am. Oil Chem. Soc.* **2016**, *93*, 575.

383 [9] F. Anwar, M.I. Bhangar, S. Iqbal, B. Sultana, *J. Food Qual.* **2006**, *29*, 87.

384 [10] N. Haegens, Pastries, in: *Bak. Prod. Sci. Technol.*, John Wiley & Sons, Ltd, **2014**: pp. 603–610.

385 [11] A.I. Blake, A.G. Marangoni, *Food Res. Int.* **2015**, *74*, 284.

386 [12] D. Patient, *Nutr. Food Sci.* **1994**, *4*, 33.

387 [13] M. Vaisey-Genser, in *Encyclopedia of Food Sciences and Nutrition (Second Edition)*, (Ed: B. Caballero), Academic Press, **2003**, pp. 3704–3709.

389 [14] R. Carr, M. Vaisey-Genser, in *Encyclopedia of Food Sciences and Nutrition (Second Edition)*, (Ed: B. Caballero), Academic Press, **2003**, pp. 3709–3714.

391 [15] N.C. Acevedo, A.G. Marangoni, *Food Biophys.* **2014**, *9*, 368.

392 [16] P. Garcia-Macias, M.H. Gordon, R.A. Frazier, K. Smith, L. Gambelli, *Eur. J. Lipid Sci. Technol.*  
393 **2011**, *113*, 1474.

394 [17] P. Garcia-Macias, M.H. Gordon, R.A. Frazier, K. Smith, L. Gambelli, *Eur. J. Lipid Sci. Technol.*  
395 **2012**, *114*, 741.

396 [18] M. Aguilar-Zárate, B.A. Macias-Rodriguez, J.F. Toro-Vazquez, A.G. Marangoni, *Carbohydr.*  
397 *Polym.* **2019**, *205*, 98.

398 [19] A.J. Gravelle, S. Barbut, A.G. Marangoni, *Food Res. Int.* **2012**, *48*, 578.

399 [20] S. Plazzotta, S. Calligaris, L. Manzocco, *Food Res. Int.* **2020**, *132*, 109099.

- 1  
2  
3 400 [21] M. Vaisey-Genser, in *Encyclopedia of Food Sciences and Nutrition (Second Edition)*, (Ed: B.  
4 Caballero), Academic Press, **2003**, pp. 5257–5261.  
5 401  
6  
7 402 [22] L. Manzocco, M. Anese, S. Calligaris, B. Quarta, M.C. Nicoli, *Food Chem.* **2012**, *132*, 175.  
8  
9  
10 403 [23] S. Calligaris, L. Manzocco, F. Valoppi, M.C. Nicoli, *Food Res. Int.* **2013**, *51*, 596.  
11  
12 404 [24] A. Goldstein, K. Seetharaman, *Food Res. Int.* **2011**, *44*, 1476.  
13  
14 405 [25] S. Calligaris, M. Marino, M. Maifreni, N. Innocente, *LWT - Food Sci. Technol.* **2018**, *96*, 329.  
15  
16  
17 406 [26] H.D. Batte, A.J. Wright, J.W. Rush, S.H.J. Idziak, A.G. Marangoni, *Food Res. Int.* **2007**, *40*,  
18  
19 407 982.  
20  
21 408 [27] H.D. Batte, A.J. Wright, J.W. Rush, S.H.J. Idziak, A.G. Marangoni, *Food Biophys.* **2007**, *2*, 29.  
22  
23  
24 409 [28] A.G. Marangoni, S.H.J. Idziak, C. Vega, H. Batte, M. Ollivon, P.S. Jantzi, J.W.E. Rush, *Soft*  
25  
26 410 *Matter* **2007**, *3*, 183.  
27  
28 411 [29] F. Valoppi, S. Calligaris, L. Barba, M.C. Nicoli, *Food Biophys.* **2015**, *10*, 94.  
29  
30  
31 412 [30] F. Valoppi, S. Calligaris, L. Barba, M.C. Nicoli, *Food Res. Int.* **2015**, *74*, 224.  
32  
33 413 [31] S. Calligaris, S. Da Pieve, G. Arrighetti, L. Barba, *Food Res. Int.* **2010**, *43*, 671.  
34  
35 414 [32] S. Calligaris, L. Manzocco, S. Plazzotta (University of Udine), WO. 007399, **2018**.  
36  
37  
38 415 [33] T. Roisnel, J. Rodriguez-Carvajal, in *Proc. Seventh Eur. Powder Diffr. Conf. (EPDIC 7)* (Eds. R.  
39  
40 416 Delhez, E. Mittemeijer), Trans Tech Publications, Zurich **2001**.  
41  
42 417 [34] AOAC, Official methods of analysis, Washington, DC, **1997**.  
43  
44 418 [35] L. Manzocco, C. Lagazio, *Food Qual. Prefer.* **2009**, *20*, 24.  
45  
46  
47 419 [36] W.K. Lewis, *J. Ind. Eng. Chem.* **1921**, *13*, 427.  
48  
49 420 [37] I. Tavernier, A.R. Patel, P. Van der Meeren, K. Dewettinck, *Food Hydrocoll.* **2017**, *65*, 107.  
50  
51 421 [38] A.R. Patel, P.S. Rajarethinem, N. Cludts, B. Lewille, W.H. De Vos, A. Lesaffer, K. Dewettinck,  
52  
53  
54 422 *Langmuir* **2015**, *31*, 2065.  
55  
56 423 [39] A.R. Patel, N. Cludts, M.D. Bin Sintang, B. Lewille, A. Lesaffer, K. Dewettinck,  
57  
58  
59  
60

- 1  
2  
3 424            *ChemPhysChem* **2014**, *15*, 3435.  
4  
5 425    [40]    W. Wijaya, P. Van Der Meeren, K. Dewettinck, A.R. Patel, *Food Funct.* **2018**, *9*, 1993.  
6  
7 426    [41]    S. Rønholt, J.J.K. Kirkensgaard, K.F. Høyer, K. Mortensen, J.C. Knudsen, *JAOCS, J. Am. Oil*  
8  
9            *Chem. Soc.* **2014**, *91*, 29.  
10 427  
11  
12 428    [42]    A.R. Patel, R.A. Nicholson, A.G. Marangoni, *Curr. Opin. Food Sci.* **2020**, *33*, 61.  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25  
26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60

For Peer Review

Table 1. Model parameter estimates for the drying process at 30, 40 and 50 °C of structured emulsions containing 4 (MG4) and 9% (MG9) monoglycerides.

Sample	Temperature (°C)	$k$ (h <sup>-1</sup> )	R <sup>2</sup>
<b>MG4</b>	30	0.103 ± 0.009	0.996
	40	1.119 ± 0.220	0.952
	50	2.018 ± 0.288	0.997
<b>MG9</b>	30	0.150 ± 0.024	0.982
	40	1.207 ± 0.234	0.953
	50	2.464 ± 0.315	0.997

Table 2. Rheological parameters of structured emulsions before (MG4, MG9) and after (MG4-dried, MG9-dried) drying.

Sample	G' (×10 <sup>5</sup> Pa)	Critical stress (Pa)
<b>MG4</b>	0.085 ± 0.012 <sup>c</sup>	22.6 ± 3.3 <sup>d</sup>
<b>MG9</b>	0.438 ± 0.029 <sup>b</sup>	86.0 ± 4.6 <sup>c</sup>
<b>MG4-dried</b>	0.533 ± 0.098 <sup>b</sup>	139.4 ± 6.4 <sup>b</sup>
<b>MG9-dried</b>	4.565 ± 0.212 <sup>a</sup>	633.9 ± 25.5 <sup>a</sup>

<sup>a, b, c, d</sup> In the same column, means indicated by different letters are statistically different (p<0.05)

Table 3. Leavening capacity, firmness, colour and sensory attributes of puff pastry prepared with commercial roll-in margarine and MG9-dried emulsion.

Roll-in fat	Colour			Leavening capacity	Firmness (kN)	Sensory score		
	L*	a*	b*			Firmness	Friability	Oiliness
<b>Margarine</b>	82.84 ± 0.61	-1.22 ± 0.12	24.22 ± 0.93	12.3 ± 1.2	1.1 ± 0.2	5.0 ± 0.8	5.9 ± 1.6	4.5 ± 1.2
<b>MG9-dried</b>	85.66 ± 1.08	-0.50 ± 0.14	15.91 ± 0.76	10.4 ± 0.5	0.8 ± 0.1	4.9 ± 1.9	4.9 ± 2.3	7.3 ± 1.0