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Compositional characterization of commercial sparkling wines from cv. Ribolla
Gialla produced in Friuli Venezia Giulia

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Characterization of the metabolomic profile of Ribolla Gialla commercial sparkling wines

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4 1 Characterization of the metabolomic profile of Ribolla Gialla
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7 2 commercial sparkling wines
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3 22 **Abstract**
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5 23 Ribolla Gialla (RG) is a white grape variety used in the production of high-quality sparkling
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7 24 wines. It is widespread in North-East Italy, particularly Friuli Venezia Giulia (FVG), as well as
8
9 25 Slovenia. Because of its limited area of cultivation, however, there is little information about
10
11 26 the composition and chemical characteristics of the sparkling wines produced. This work used
12
13 27 different analytical approaches to characterize thirty-three commercial sparkling RG wines
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15 28 from different areas of FVG. The characteristics included the overall volatile profile and content
16
17 29 of terpenes, C₁₃-norisoprenoids, lipids, and tryptophan metabolites. The aroma profile of RG
18
19 30 wines was mainly characterized by fermentative esters and β -damascenone, whereas other
20
21 31 norisoprenoids and varietal aromas were below the odor threshold. Appreciable amounts of
22
23 32 certain fatty acids were found (e.g., palmitic acid), which could be potentially correlated with
24
25 33 greater foam stability. However, high concentrations of tryptophan metabolites highlighted a
26
27 34 higher risk of developing atypical ageing defects.
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35 36 **KEYWORDS:** Ribolla Gialla; sparkling wine; volatile compounds; tryptophan metabolites;
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39 **Introduction**

40 Ribolla Gialla is a promising white grape variety that has recently been used for the production
41 of premium sparkling wines, which have received good appreciation in international wine
42 markets. Ribolla has been cultivated since ancient times in North-Eastern Italy (in the Friuli
43 Venezia Giulia region), as well as Slovenia and the Ionian Islands (Kefalonia), where it is
44 known as Rebula and Robola, respectively [1]. A certain number of studies have been carried
45 out on the genetic identity and characterization of Ribolla grapes [1–4], but very few studies
46 have dealt with the metabolomic profile of the wines [5, 6]. In addition, none of them have
47 focused on the aromatic characteristics of sparkling wines obtained from Ribolla grapes,
48 including their volatile organic compound (VOC) content, lipids, polyphenols, and tryptophan
49 metabolites. All of these compounds are important because they may be linked with positive
50 and negative aspects of wine quality.

51 VOCs are fundamental for determining the sensory characteristics of wine and its profile. These
52 compounds are classified into four groups [7]: i) primary grape VOCs, which are present in the
53 cells of grapes; ii) secondary grape aroma, which are formed during the pre-fermentation phases
54 (crushing, pressing, and skin contact) or by thermal, chemical, and enzymatic reactions in must;
55 iii) fermentation bouquet, which includes aroma compounds that form during alcoholic
56 fermentation; iv) and maturation bouquet, which refers to the aroma compounds that develop
57 when ageing the wine. Primary aroma is particularly important for the production of certain
58 wine typologies, such as white sparkling wines. Furthermore, different aroma compounds may
59 be produced after fermentation and ageing that further affect the aroma profile of the final wine,
60 which are mainly due to pre-fermentative processes and yeast metabolism (e.g., C6-compounds,
61 norisoprenoids, fatty acids, alcohols, esters, carbonyl compounds) [8–11].

62 Moreover, yeast autolysis takes place during the ageing of sparkling wine, which leads to the
63 release of substances in the medium. Lipids represent a wide class of compounds among these
64 substances [8]. Fatty acids may have a significant impact on the sensory properties of the wine.

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3 65 A large part of the fatty acids originates from the firm tissues of the grapes, but the greatest
4
5 66 amount is formed during alcoholic fermentation. Therefore, fatty acids may be present in wine
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7 67 in free or bound forms as ethyl esters. Both forms directly contribute to the flavor of the wine,
8
9 68 while unsaturated fatty acids such as oleic, linolenic, and linoleic acids act as precursors of C6-
10
11 69 aldehydes and alcohols with herbaceous notes, and are critical for yeast growth during
12
13 70 fermentation [12]. Another part of the sensory assessment of sparkling wine is the relationship
14
15 71 between the foaming properties and lipid compounds. Therefore, it has been reported that
16
17 72 medium-chain free fatty acids C8, C10, and C12 are negatively correlated with foamability,
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19 73 while the ethyl esters of hexanoic, octanoic, and decanoic acids have a positive effect on the
20
21 74 formation and stability of the foam [13].
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26 75 The essential amino acid tryptophan (TRP) and its metabolites, especially indole-3-acetic acid
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28 76 (IAA), are considered to be potential precursors of 2-aminoacetophenone (2-AAP), which is an
29
30 77 aroma compound that causes an atypical ageing off-flavor (ATA) in *Vitis vinifera* wines. The
31
32 78 off-flavor is described with aroma descriptors such as “acacia blossom,” “furniture polish,”
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34 79 “wet wool,” “mothballs,” or “fusel alcohol,” which are combined with a loss of the typical
35
36 80 bouquet of the grape variety [14]. Depending on the wine matrix, the detection threshold of 2-
37
38 81 AAP varies from 0.5 to 1.5 $\mu\text{g L}^{-1}$. Strongly aromatic wines are able to integrate more than 1.5
39
40 82 $\mu\text{g L}^{-1}$ of 2-AAP, while meager wines might be rejected as tainted by ATA with less than 0.5
41
42 83 $\mu\text{g L}^{-1}$. It is generally accepted that the ultimate cause of ATA development in white wines is a
43
44 84 stress reaction in the vineyard triggered by drought, nutritional deficiency, and other viticultural
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46 85 factors, such as the time of harvest and leaf removal [14, 15].
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51 86 Most papers on sparkling wines focus on the characterization of internationally known varieties
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53 87 with a worldwide distribution [16–19], but there are few studies on local or less relevant
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55 88 cultivars. Therefore, the aim of this work was to investigate the compositional profile of
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57 89 different Ribolla Gialla sparkling wines produced in different DOC (*Denominazione di Origine*
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59 90 *Controllata*) districts in the Friuli Venezia Giulia region in North-East Italy. The wines were
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91 characterized for their chemical composition (pH, titratable acidity, residual sugars, alcohol,
92 and sulfur dioxide), volatile profile (overall profile of aroma compounds, free and bound
93 terpenes, and norisoprenoids), and content of lipids and tryptophan metabolites. The results are
94 critically discussed with the aim of defining the metabolomic profile of Ribolla Gialla sparkling
95 wines.

96 **Materials and Methods**

97 *Reagents and materials*

98 Hydrogen peroxide (30% w/w), ethanol (96% v/v), ACS grade hydrochloric acid (37%),
99 anhydrous sodium sulfate and citric acid were purchased from Carlo Erba Reagents (Milan,
100 Italy). Sodium chloride (99.5%) was obtained from Honeywell Fluka (Morris Plains, New
101 Jersey). HPLC-grade solvents dichloromethane, *n*-pentane, and methanol, LC-MS grade
102 methanol, acetonitrile, 2-propanol, chloroform, formic acid, ammonium formate, ethyl
103 heptanoate, 1-heptanol, 2-octanol, ethyl hexanoate- d_{11} , 3-(2-hydroxy ethyl)-indole, kynurenic
104 acid, D-tryptophan methyl ester, L-tyrosine ethyl ester, *N*-acetyl tyrosine ethyl ester, and 3,5-
105 di-*tert*-4-butylhydroxytoluene (BHT) were obtained from Sigma-Aldrich (St. Louis, MO,
106 USA). C_7 - C_{30} *n*-alkane solution in *n*-hexane was purchased from Supelco (Bellefonte, PA,
107 USA), while cholesterol- d_7 and octadecanoic acid- d_3 were obtained from CDN Isotopes
108 (Quebec, Canada). Finally, the glycosidase preparation (Rapidase Revelation Aroma) used for
109 the determination of bound monoterpenes was obtained from Oenobrand (Montellier, France).

110 *Wine samples*

111 Local wineries located in the North-East Italian region of Friuli Venezia Giulia provided
112 thirty-three commercial Ribolla Gialla sparkling wines. The locations of the wineries are
113 shown in Figure 1.

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3 114 *Basic analysis of wine samples*
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6 115 The basic quality control parameters (reducing sugars, alcoholic strength, total acidity, volatile
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8 116 acidity, pH, malic, lactic and tartaric acid) were determined by FTIR spectroscopy with a
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10 117 Winescan™ FT-120 instrument (FOSS, Hillerød, Denmark). All of the samples were analyzed
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12 118 two times, and the mean value of the two measurements was considered for the data analysis.
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17 119 *Determination of free and bound terpenes and norisoprenoids by SPE-GC-MS*
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19 120 Terpenes and C₁₃-norisoprenoids in free and bound forms were extracted on Isolute® 500-mg,
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21 121 6-mL, C₁₈ SPE cartridges (Biotage, Uppsala, Sweden) according to the method reported by
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23 122 Comuzzo et al. [20]. A GCMS-QP-2010 system (Shimadzu, Kyoto, Japan) was used for GC-
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25 123 MS analyses. Volatile compounds were separated on a J&W DB-Wax capillary column (30 m
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27 124 × 0.25 mm i.d., 0.25 µm film thickness) provided by Agilent Technologies Inc. (Santa Clara,
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29 125 CA, USA) using the operating conditions described by Loira et al. [21].
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33 126 Electron impact mass spectra were recorded at 70 eV and volatile compounds were
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35 127 tentatively identified by comparison of their mass spectra and retention times with those of
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37 128 standard compounds or by comparison of the mass spectrum with those reported in the Wiley
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39 129 6, NIST 21, and NIST 107 mass spectrum libraries (provided by the manufacturer). Linear
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41 130 retention indices were calculated according to the retention times of *n*-alkanes and compared
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43 131 with those reported in literature. The semi-quantitative analysis was based on the internal
44
45 132 standard method using 1-heptanol (312 mg L⁻¹ in 96 % v/v ethanol) as the internal standard
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47 133 while considering the response factor to be equal to 1.00.
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53 134 *Determination of volatile compounds*
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56 135 The volatile compounds in the commercial wines were determined using two analytical
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58 136 approaches: liquid-liquid extraction (LLE) and solid-phase microextraction (SPME). Samples
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3 137 were degassed before analysis by placing them in an ultrasonic bath (Falc Labsonic, Treviglio,
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5 138 Italy) for 2 min.

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8 139 *Liquid-liquid extraction (LLE-GC-MS)*

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11 140 Aroma compounds were determined as reported by Loira et al. [21] by mixing 5 mL of wine
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13 141 with 5 mL of a 30% (w/v) sodium chloride solution and 200 μL of internal standard (ethyl
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15 142 heptanoate, 422 mg L^{-1} in 96% v/v ethanol). The mixture was subjected to three extractions
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17 143 using 2.5 mL of pentane:dichloromethane (2:1 v/v) each. The organic phase was collected in a
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19 144 Pyrex tube, dehydrated with anhydrous sodium sulfate, and concentrated under nitrogen flow
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21 145 to a final volume of about 1 mL. The samples obtained were subjected to GC-MS analysis as
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23 146 reported previously[21].

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27 147 *SPME-GC-MS/MS*

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30 148 Degassed wine (1 mL) was spiked with 50 μL of 2-octanol at 2.13 mg L^{-1} (IS) in ethanol and
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32 149 placed in a 20-mL headspace vial containing 1.5 g of sodium chloride. Two technical replicates
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34 150 were prepared for each sample, along with a blank sample containing only sodium chloride.
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37 151 GC analysis was performed using a Trace GC Ultra gas chromatograph coupled with a TSQ
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39 152 Quantum Tandem mass spectrometer, which was upgraded to the XLS configuration and
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41 153 equipped with a Triplus autosampler (Thermo Fisher Scientific, Waltham, MA, USA). The
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43 154 method used was adopted by Carlin et al. [22]. Samples were incubated for 5 min at 35°C,s and
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45 155 VOCs were extracted for 20 min with a 2-cm-long 50/30- μm coated
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47 156 divinylbenzene/carboxen/polydimethylsiloxane (DVB/CAR/PDMS) SPME fiber (Supelco,
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49 157 Sigma Aldrich, Milan, Italy). After solid-phase microextraction, the fiber was desorbed for 3
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51 158 min at 250°C in the GC system with the injector set in splitless mode. The fiber was
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53 159 reconditioned between each sample at 270°C for 7 min. Helium was used as a carrier gas at a
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55 160 flow rate of 1.2 mL min^{-1} .

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3 161 The GC oven was equipped with a 30-m × 0.25-mm VF-WAXms column (Agilent
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5 162 Technologies Inc., Santa Clara, CA, USA) with a film thickness of 0.25 μm. The oven
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7 163 temperature was held for 2 min at 40°C after the injection, ramped at 6°C min⁻¹ up to 250°C,
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9 164 and held for 5 min. The ion source was set at 230°C, and electron impact mass spectra were
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11 165 recorded at 70 eV. Data acquisition and analyses were performed using the Xcalibur
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13 166 Workstation software supplied by the manufacturer.
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18 167 *Determination of tryptophan metabolites by UHPLC-MS/MS*
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21 168 Sparkling wine samples were filtered at 0.22 μm by a Millex-GV filtration unit (Merc,
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23 169 Darmstadt, Germany) and directly collected in 2-mL HPLC amber vials. The analyses were
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25 170 performed on an Acquity UHPLC system provided with an autosampler and coupled with a
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27 171 XEVO TQMS mass spectrometer equipped with an electrospray source (Waters Corporation,
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29 172 Milford, MA, USA). Tryptophan metabolites were separated at a flow rate of 0.4 mL min⁻¹ on
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31 173 a 1.8-μm 150 × 2.1-mm Waters Acquity HSS T3 column (Waters Corporation), which was
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33 174 conditioned at 40°C. A linear gradient of water containing 0.1% formic acid (solvent A) and
34
35 175 0.1% formic acid in acetonitrile (solvent B) was used according to the conditions reported by
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37 176 Arapitsas et al.[23]. The injection volume was 10 μL, and the wine samples were stored in the
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39 177 autosampler at 6°C during analyses.
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44 178 The MS conditions included a capillary voltage of 3.5 kV in positive mode and -2.7 kV in
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46 179 negative mode. The ion source temperature was set at 150°C, while the desolvation temperature
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48 180 was 500°C. Nitrogen was used as the cone gas with a flow rate of 50 L h⁻¹. The data were
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50 181 processed using Waters MassLynx (version 4.1) and TargetLynx software (Milford, MA,
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52 182 USA).
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3 183 *Analysis of lipid molecules by UHPLC-MS/MS*
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6 184 Lipid analysis was performed as described by Della Corte et al. [24] with slight modifications.
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8 185 An aliquot (3 mL) of degassed wine was introduced in a 50-mL Falcon tube and spiked with
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10 186 30 μL of a methanolic solution containing cholesterol- d_7 and octadecanoic acid- d_3 (IS) at
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12 187 concentrations of 1.01 mg mL^{-1} and 1.002 mg mL^{-1} , respectively. Lipids were extracted two
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14 188 times in 21 mL of a chloroform–methanol solution (2:1 v/v) containing BHT (10 mg L^{-1}). The
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16 189 total lower lipid-rich layer was collected in 100-mL flasks, and the solvent was evaporated to
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18 190 dryness using a rotary evaporator. The samples were reconstituted in $300 \mu\text{L}$ of acetonitrile/2-
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20 191 propanol/water (65:30:5 v/v/v) and filtered ($0.22 \mu\text{m}$) into 2-mL HPLC amber vials for
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22 192 UHPLC-MS/MS analysis.
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24 193 UHPLC separation was performed on a Dionex 3000 chromatograph (Thermo Fisher Scientific,
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26 194 Waltham, MA, USA) equipped with an autosampler and coupled with an API 5500 triple-
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28 195 quadrupole mass spectrometer (Sciex, Concord, Vaughan, ON, Canada), which was provided
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30 196 with an electrospray ion source (ESI). Lipids were separated on a $2.7\text{-}\mu\text{m}$ $150 \times 2.1\text{-mm}$ RP
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32 197 Ascentis Express column (Sigma-Aldrich, Milan, Italy) set at 55°C . The injection volume was
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34 198 $5 \mu\text{L}$, and the samples were stored in an autosampler at 10°C during the analyses. The mobile
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36 199 phase and chromatographic conditions are those described by Della Corte et al. [24], and the
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38 200 flow rate was $0.260 \text{ mL min}^{-1}$.
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40 201 The spray voltage of the ESI source was set at 5500 V for positive mode and -4500 V for
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42 202 negative mode, and the source temperature was 250°C . The nebulizer (Gas 1) and heater gas
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44 203 (Gas 2) pressures were set at 40 and 20 psi, respectively. Ultra-high-purity nitrogen (99.999%)
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46 204 was used as both a curtain gas and collision gas (CAD) at 20 and 9 psi, respectively. Instrument
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48 205 control and data acquisition were performed by Analyst software (Applera Corporation,
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50 206 Norwalk, CT, USA), and the data were processed using MultiQuant, version 2.1 (Sciex,
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52 207 Concord, Vaughan, ON, Canada).
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3 208 *Statistical analysis*
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6 209 Statistica for Windows Version 8 (Statsoft, Tulsa, OK, USA) was used to calculate the means,
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8 210 standard deviations (SDs), coefficients of variation (CVs), minimum values (MIN), and
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10 211 maximum values (MAX) of the different analytical parameters determined for wine samples.
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14 212 **Results and Discussion**
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17 213 *Basic quality parameters*
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20 214 Table 1 shows the results related to the basic chemical composition of the wines. The mean
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22 215 values, SDs, and CVs show that the samples are very homogeneous concerning alcoholic
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24 216 strength, titratable acidity, tartaric acid, pH, and even volatile acidity, which shows quite low
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26 217 values despite a slightly higher CV. This relatively homogeneous composition is probably
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28 218 related to the fact that the sparkling wines analyzed are all DOC productions and were
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30 219 manufactured in fulfillment of the standards reported in the DOC production guidelines.
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34 220 However, some differences that are more relevant from a practical point of view were observed
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36 221 for residual sugars (min-max range: 6-19 g L⁻¹), malic acid (0.01-3.79 g L⁻¹), and lactic acid
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38 222 (0.00-2.75 g L⁻¹). These differences could be related to the application of different winemaking
39
40 223 protocols by the producers, leading to different styles of wine for Ribolla Gialla DOC, even if
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42 224 they are in agreement to the DOC production guidelines. In regard to sugars, most of the wines
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44 225 analyzed showed a residual sugar content of less than 12 g L⁻¹ or between 12 and 17 g L⁻¹.
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46 226 According to the European Commission Regulation (EC) n. 607/2009 [25], these wines are
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48 227 classified as *brut* and *extra dry*, respectively. Only one sample had higher sugar content (about
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50 228 19 g L⁻¹) and was categorized as *sec* (or *dry*).
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54 229 The variations of malic and lactic acid highlight the different ways of managing malolactic
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56 230 fermentation (MLF), which are probably connected with the different storage times of the wines
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58 231 on yeast lees after refermentation. Most of the wines did not contain lactic acid in appreciable
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3 232 amounts, which was probably due to the lack of malolactic fermentation. MLF generally occurs
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5 233 during ageing on the lees because of the positive effect that they have on promoting the growth
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7 234 of lactic acid bacteria [26]. For this reason, most winemakers would consider Ribolla Gialla
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9 235 sparkling wine as a fresh and young wine according to the data.

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12 236 Ribolla Gialla is generally refermented via the Martinotti/Charmat short method, which
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14 237 involves refermentation in stainless steel autoclaves without contact with the lees at the end of
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17 238 refermentation. However, six wine samples had a higher content of lactic acid. This accounted
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19 239 for the increase of the CV calculated for lactic acid in Table 1 and was combined with the
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21 240 decrease of malic acid concentration. This evidence may be related to the manufacture of certain
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23 241 Ribolla Gialla wines, with some products being kept in contact with the lees after
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25 242 refermentation in an autoclave. Such *sur-lies* aging may be prolonged for up to several months
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28 243 (2-4 months on average). In addition, certain winemakers also produce Ribolla Gialla by the
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30 244 traditional refermentation method in bottles.

31 32 33 34 245 *Wine aroma profile*

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37 246 Table 2 shows the results of the qualitative and quantitative determination of the aroma
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39 247 compounds detected in wine samples by the different analytical techniques. Concerning non-
40
41 248 varietal aromas, a total number of fifty-eight volatiles were tentatively identified in the wines,
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44 249 including acids, alcohols, esters, C6 compounds, diols, and carbonyls. Fatty acids are some of
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46 250 the most representative volatiles in the wines analyzed, and generally, they are described as
47
48 251 having cheese, rancid, and fatty notes. If their amounts are higher than the odor detection
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50 252 threshold (ODT), they may negatively affect the organoleptic characteristics of the wines.

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52
53 253 In the samples analyzed, the average concentration of fatty acid was often close and sometimes
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55 254 higher than their ODT. Nevertheless, their odor activity value (OAV), which is the ratio
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57 255 between the concentration and ODT, was in the range of 1-4 for many of the compounds
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59 256 analyzed. The exception was octanoic acid, which is the most representative fatty acid in the

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3 257 volatile composition of Ribolla wines ($OAV \approx 30$). These results suggest that fatty acids might
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5 258 have minor relevance in the aromatic characteristics of Ribolla Gialla. Acetic acid (which is
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7 259 responsible for the typical vinegar off-flavor) was found to be lower than ODT in all samples
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10 260 analyzed.

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12 261 Interestingly, the variability among the samples (i.e., the CVs) for fatty acids and volatile
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14 262 compounds in general (Table 2) was higher than that detected for the basic parameters in Table
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17 263 1. This highlights a certain level of differentiation among Ribolla Gialla DOC wines, which is
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19 264 probably related to a number of variables, such as those typically connected with the
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21 265 development of wine aromas. Concerning fatty acids, for instance, the differences among the
22
23 266 samples might have resulted from the different origin of the grapes used, the amount of lipid
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26 267 substances in the musts, the diverse winemaking conditions, and the yeast strains used for
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28 268 fermentations [27].

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30 269 Higher alcohols (HA) are also important volatiles among the compounds listed in Table 2. They
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32 270 are produced by yeasts during alcoholic fermentation as products of amino acid metabolism
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34 271 [11]. Their contributions to wine aroma vary from honey, rose, and floral characters (2-
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36 272 phenylethyl alcohol and benzyl alcohol) to pungent and solvent-like smells (1-propanol, 1-
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38 273 butanol, 2- and 3-methyl-1-butanol) [28], and the effects depend on their concentration [9]. The
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40 274 average amounts of HA observed in the wines analyzed were generally found to be lower than
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42 275 their ODT except for 2- and 3-methyl-1-butanol and 2-phenylethanol (ODT: 40 and 10 mg L⁻¹,
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44 276 respectively). This composition may be linked with the yeast strains used, as well as the
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46 277 winemaking conditions adopted and the amino acidic composition of the grapes. The presence
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48 278 of significant amounts of 2-phenylethanol is very interesting for its fresh rose-like odor [29],
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50 279 even if its OAV range was about 1-6.

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52 280 Esters are the most representative class of volatile compounds found in Ribolla Gialla sparkling
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54 281 wines. Isoamyl acetate (3-methyl-1-butanol acetate), ethyl hexanoate, ethyl lactate, ethyl
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56 282 octanoate, diethyl succinate, and 2-phenylethyl acetate showed the highest concentrations. The

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3 283 compounds eluted at higher retention times (e.g., diethyl malate) presumably had a minor
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5 284 impact on the aroma profile because of their low volatility. Some esters, such as ethyl butanoate
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7 285 and ethyl hexanoate, were close to or higher than their ODT (20 µg and 65 µg) in all the samples
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9 286 analyzed. Others such as isoamyl acetate, 2-phenylethyl acetate, and hexyl acetate were
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11 287 detected in only certain wines with concentrations higher than their ODT. Compounds such as
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13 288 ethyl hexanoate and isoamyl acetate were found to have significant OAVs (e.g., up to 18 for
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15 289 ethyl hexanoate, 85 for hexyl acetate, and approximately 200 for isoamyl acetate). This
16
17 290 observation is very interesting from a sensory point of view because esters generally confer
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19 291 floral, fresh, and fruity notes (rose, banana, pear, green apple) to the wines [28, 30, 31], which
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21 292 are generally recognized as typical in Ribolla wines. For this reason, esters appeared to be an
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23 293 important component of the volatile profile of Ribolla Gialla.

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26 294 Among esters, ethyl lactate and diethyl succinate are considered “ageing esters” (AE), and their
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28 295 content in wines generally increases during ageing and after malolactic fermentation [29, 32].
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30 296 The same consideration is true for diethyl malate, but this compound is normally detected less
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32 297 frequently in wines. The concentration of AE was generally found to be lower than the ODT in
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34 298 most of the wines analyzed. However, in some samples, ethyl lactate had higher concentrations
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36 299 than its odor threshold (60 mg/L) in accordance with the amount of malic and lactic acid. In
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38 300 fact, the highest concentration of these esters was observed in the same products in which
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40 301 malolactic fermentation was hypothesized to have taken place (i.e., the wines where malic acid
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42 302 was not detected), as shown in Table 1. This may confirm that certain Ribolla Gialla sparkling
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44 303 wines are produced with a more or less prolonged period of *sur-lies* ageing or by refermenting
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46 304 them in bottle by the traditional method.

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49 305 Carbonyls, diols, and C6 compounds appeared not to have an important contribution to the
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51 306 aromatic characterization of the samples, and most of the cases, they were detected at levels
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53 307 below their ODT. Hexanal was found to overcome its ODT in some samples, but at the low
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55 308 concentrations detected, it may or may not contribute to the fresh vegetal notes of Ribolla wines.

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3 309 Table 3 reports the detected varietal aromas (terpenes and C₁₃-norisoprenoids) in free and bound
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5 310 form in the samples. Terpenes are well known to confer floral odors to wines [29]. In Ribolla,
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7 311 they were especially found in the free form, but in most cases, their average concentrations
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9 312 were below their ODT with a few exceptions for linalool and geraniol. Although it was present
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11 313 below its ODT, α -terpineol was the most abundant free terpenic alcohol in the wines.
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13 314 Interestingly, even if Ribolla Gialla cannot be considered as an aromatic variety, the
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15 315 concentrations detected for α -terpineol and linalool were higher than those normally observed
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17 316 in certain international white varieties, such as Pinot blanc, Pinot Gris, Chardonnay, and
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19 317 Sauvignon blanc [19]. Also β -citronellol, was found at lower levels with respect to its ODT
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21 318 (100 $\mu\text{g L}^{-1}$) [33], it showed similar average values to those of some Riesling wines [19].
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23 319 Considering the glycosylated forms, only geraniol was significantly detected as both free and
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25 320 bound terpenol with a prevalence of the latter (combined) form of the molecule. According to
26
27 321 other experiments, geraniol (with minor traces of nerol) was the only terpenic alcohol normally
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29 322 found in Ribolla Gialla grapes [34]. For this reason, in Ribolla Gialla, linalool, α -terpineol, and
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31 323 β -citronellol may be formed from such terpenic alcohols during grape processing. Citronellol,
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33 324 for instance, is reported to be produced by *S. cerevisiae* from geraniol and nerol during alcoholic
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35 325 fermentation, and other formation pathways have also been suggested for linalool (from
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37 326 geraniol) and α -terpineol (by cyclization of nerol) [35]. For both free and bound geraniol, the
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39 327 average values were generally lower than the ODT reported for the alcohol, but some of the
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41 328 samples showed significant levels of total geraniol (Table 3). Considering that the bound form
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43 329 generally prevails for this terpenol, one interesting way to improve the volatile profile of these
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45 330 wines might be the application of techniques to increase the release of free geraniol from its
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47 331 glycosides in the production process of Ribolla Gialla sparkling wines.
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49 332 In contrast to terpenes, which did not appear particularly characteristic for the aroma profile of
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51 333 the wines, some C₁₃-norisoprenoids were detected in appreciable concentrations (Table 3).
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53 334 TDN and β -damascenone, for instance, overcame their sensory thresholds in different samples.
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3 335 Norisoprenoids are generally not connected to specific aromatic grapevine varieties and
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5 336 generally have a low odor threshold. The ODT of β -damascenone is 0.04-0.06 $\mu\text{g L}^{-1}$ in a model
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8 337 dilute alcohol solution [29], while a relatively wide range of ODTs was found in wine: Sefton
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10 338 et al. [36] reported 0.14 $\mu\text{g L}^{-1}$ in deodorized white wine, 0.85 – 2.10 $\mu\text{g L}^{-1}$ in deodorized red
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12 339 wine, and 7.00 $\mu\text{g L}^{-1}$ in red wine. The presence of β -damascenone may be particularly
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15 340 interesting for Ribolla wines for not only the relatively high concentrations found but also
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17 341 because it is reported to modify the sensory perception of some esters. Escudero et al.[37] found
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19 342 that the addition of low levels of β -damascenone (0.85 $\mu\text{g L}^{-1}$) to a solution of esters increased
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21 343 the fruity notes of the mixture, while higher levels (3.5 $\mu\text{g L}^{-1}$) accounted for the development
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23 344 of strong raisin/dry-plum odors.
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26 345 The amount of norisoprenoids is also dependent on the winemaking conditions. In general, β -
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28 346 damascenone is observed at higher concentrations in young wines [36], whereas for 1,1,6-
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30 347 trimethyl-1,2-dihydronaphthalene (TDN), a positive correlation between its occurrence in wine
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32 348 and ageing was reported.[38] Their odor descriptors range from floral (e.g., vitispiranes) and
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34 349 fruity-honey scents (β -damascenone) to notes of kerosene and petrol (for TDN) [39]. The
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36 350 presence of TDN among the compounds listed in Table 3 may represent further evidence about
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38 351 the ageing of certain Ribolla Gialla sparkling wines after refermentation.
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41 352 Another interesting aspect related to norisoprenoids is the high concentration observed for some
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43 353 of them in bound form, particularly 3-oxo- α -ionol and 3-hydroxy- β -damascone, which are the
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45 354 most representative compounds. As for terpenes, Ribolla Gialla appeared to have an
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47 355 unexpressed aromatic potential for norisoprenoids. Thus, from a technical point of view, the
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49 356 optimization of winemaking process to maximize the release of this bound aromatic potential
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51 357 might be an interesting way to improve the quality of Ribolla Gialla sparkling wines.
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3 358 *Metabolomic fingerprint of tryptophan metabolites and indoles*
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6 359 The second group of metabolites analyzed was the amino acid tryptophan and its catabolites,
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8 360 as shown in Table 4. There were twenty-eight compounds detected, among which the sulfonated
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10 361 derivatives of indole 3-lactic acid, tryptophol, indole 3-lactic acid glucoside, and indole 3-acetic
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12 362 acid are presented. Sulfonation is the addition of a sulfonic acid group (-SO₃H) to an organic
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14 363 compound and is a widespread industrial process used in a diverse range of products. It also
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16 364 has a major function in modulating biological activities that are known to occur in wine, which
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18 365 involve several metabolites such as polyphenols and indoles [23].
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22 366 Sparkling wines are well known to have much lower amounts of tryptophan than other wines,
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24 367 which is probably due to the second fermentation [23]. Tryptophan values for the commercial
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26 368 samples ranged from 0.01 to 0.39 mg L⁻¹ with a mean value of 0.13 mg L⁻¹, and which similarly
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28 369 occurred for its ethyl ester (0.44 mg L⁻¹ mean value). TOL-SO₃H had higher amounts among
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30 370 all the other tryptophan metabolites with a mean value of 3.73 mg L⁻¹. The unsulfonated form
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32 371 of tryptophol varied from 0.01 to 1.08 mg L⁻¹ in all samples. This could mean that the
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34 372 sulfonated/unsulfonated TOL ratio that favored TOL-SO₃H may also give rise to further
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36 373 products similar to 2-AAP.
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40 374 The ILA (0.03 mg L⁻¹) and ILA-SO₃H (8.53 mg L⁻¹) pairing showed similar behavior to the
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42 375 previously described TOL and TOL-SO₃H pairing. Arapitsas et al. [23], found that young red
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44 376 wines and rosé wines had a much higher concentration of ILA and especially ILA-GLU, which
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46 377 is synthesized by plants [40]. This was probably due to maceration with the skins during
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48 378 winemaking. In contrast, white and especially sparkling wines had the lowest concentrations
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50 379 due to soft pressing and secondary fermentation, which necessitate further nitrogen
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52 380 consumption.
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56 381 Sulfonated indole 3-acetic acid was detected at concentrations between 0.13 and 0.39 mg L⁻¹,
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58 382 while its parent compounds were detectable at very low concentrations. Hoenicke et al. [41]
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60 383 showed that the sulfonation of indoles in a model wine solution could be responsible for their

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3 384 degradation and the formation of aromatic aminobenzenes such as 2-aminoacetophenone,
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5 385 which are responsible for some of the heavy aromatic characteristics of white wines. On other
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7 386 hand, the structural similarity of TOL to IAA could also cause the detection of 2-AAP or other
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9 387 similar aromatic compounds in wine. Because of the lack of anthocyanins and the low flavanol
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11 388 content in white wines, the sulfonation of indoles can increase the risk of developing an atypical
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13 389 off-flavor. Thus, the sparkling wines produced from the Ribolla Gialla variety might have a
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15 390 higher tendency to develop atypical ageing defects, which could also be promoted by
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17 391 inappropriate storage temperature.
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23 392 *Metabolomic fingerprint of lipids*

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25 393 Table 5 shows the twenty-nine lipid compounds found in the samples of commercial sparkling
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27 394 wines according to UHPLC-MS-MS. Most of the compounds found are saturated long-chain
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29 395 fatty acids (LCFAs, more than 12 carbon atoms), although the mid-chain fatty acids (MCFAs,
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31 396 4–12 carbons) and their esters have a major influence on the organoleptic properties of wine.
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33 397 In addition to the fact that lipids are an integral part of solid grape tissues, they are also an
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35 398 important building block in wine yeasts, where the majority of LCFAs are esterified with
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37 399 glycerol or glycerophosphate to form mono-, di-, and tri-acylglycerides or
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39 400 glycerophospholipids, respectively [42].
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44 401 The two most abundant LCFAs detected in the samples were palmitic acid (C16:0), which
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46 402 ranged from 5.21 to 11.49 mg L⁻¹ with a mean value of 9.24 mg L⁻¹, and stearic acid (C18:0),
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48 403 which ranged from 4.43 to 8.40 mg L⁻¹ (mean value: 6.49 mg L⁻¹). Previous studies showed that
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50 404 after the beginning of second fermentation, which is after the inoculation of the medium with
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52 405 yeast, the concentration of saturated fatty acids C16:0 slightly increased, while the proportion
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54 406 of C18:0 remained constant. However, after the end of the growth phase of the yeasts, the
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56 407 distributions of both saturated fatty acids decreased exponentially [43].
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3 408 Unlike saturated fatty acids (SFA), the values of unsaturated fatty acids (UFA) were lower in
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5 409 the wine samples (*cis*-11-eicosanoic acid 0.11 mg L⁻¹, myristoleic acid 0.05 mg L⁻¹, linoleic
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7 410 acid 0.04 mg L⁻¹, palmitoleic acid 0.02 mg L⁻¹, and linolenic acid 0.01 mg L⁻¹). UFAs also had
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9 411 a much smaller proportion than SFAs. Because of their membrane fluidity, wine yeasts are able
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11 412 to modify the UFA/SFA ratio in their membrane composition in response to stress conditions,
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13 413 such as the presence of ethanol or cooler fermentation conditions under which fermentation
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15 414 takes place, especially for white wines. Normally, the UFA/SFA ratio is close to 1:1, but a
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17 415 deficiency of UFA leads to impaired biosynthesis of phospholipids and consequently stuck
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19 416 fermentations [12].

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23 417 Similarly, Pueyo et al. [44] studied the effect of the total contents of linolenic acid and palmitic
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25 418 acid as compounds that best define foam stability in wines and foam height in the cavas,
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27 419 respectively. They found a positive correlation between these compounds and the measured
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29 420 properties, meaning that the wines with a greater amount of these fatty acids in the lipidic
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31 421 fraction could form more foam with high stability than those with lower concentrations. Since
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33 422 the analyzed samples in the present study contained a greater concentration of palmitic acid
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35 423 compared to the linolenic acid, it can be expected that foam formation of the Ribolla Gialla
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37 424 sparkling wines will be increased, which is a key parameter for determining the quality of
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39 425 sparkling wines.

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43 426 In conclusion, Ribolla Gialla sparkling wines showed homogeneous characteristics concerning
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45 427 basic parameters (e.g., sugar content, alcoholic strength, pH, and titratable acidity) as well as
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47 428 their aroma composition. Sparkling Ribolla wines are generally characterized by low levels of
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49 429 free terpenols, and their aroma seems to be mostly characterized by the development of volatile
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51 430 esters and β -damascenone during fermentations, storage, and processing. The presence of these
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53 431 compounds is probably connected with the fresh and fruity notes that normally characterize the
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55 432 sparkling wines produced from Ribolla grapes.

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3 433 A minor amount of malolactic fermentation was found, which reflects that two distinct wine
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5 434 styles coexist for Ribolla sparkling wines. The first characterizes young wines normally
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7 435 produced by refermentation in stainless-steel autoclaves (Martinotti/Charmat short
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9 436 refermentation method), and the second is based on a more or less prolonged ageing period on
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11 437 refermentation lees in an autoclave or bottles (according to the traditional refermentation
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13 438 method). The analysis of lipid molecules, tryptophan, and its metabolites highlighted other
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15 439 interesting features of Ribolla sparkling wines from a practical point of view. Ribolla Gialla
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17 440 shows a high ratio between saturated (e.g., palmitic acid) and unsaturated fatty acids (e.g.,
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19 441 linolenic acid), which may lead to higher foam height in the sparkling wines produced and
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21 442 represents one of the key quality features of sparkling wines in general. In contrast, due to the
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23 443 high amounts of certain tryptophan metabolites, Ribolla Gialla could be prone to the formation
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25 444 of atypical ageing aromas. Considering that this specific Italian product has only recently been
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27 445 appearing in wine markets, further investigations should carefully consider these results to
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29 446 properly address the production practices and techniques towards producing high quality
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31 447 products, thus increasing the local and international competitiveness of Ribolla Gialla sparkling
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33 448 wines.
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45
46 451 25, art. 3, paragraphs 6-10. Project Title: Maturation parameters and optimization of the
47
48 452 agronomic and enological techniques for the production of quality sparkling wines, Jul 2017 –
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54 455 analyses.
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457 **References**

- 458 1. Rusjan D, Jug T, Štajner N (2010) Evaluation of genetic diversity: Which of the varieties
459 can be named “Rebula” (*Vitis vinifera* L.)? *Vitis* 49:189–192
- 460 2. De Lorenzis G, Imazio S, Rusjan D, Vouillamoz J F, Nikolaou N, Failla O, Scienza A
461 (2013) Genetic investigation of grapevine varieties ‘Ribolla Gialla’ (Italy), ‘Rebula’
462 (Slovenia) and ‘Robola’ (Ionian Islands). *Sci Hortic* 150:425–431.
- 463 3. Imazio S, Scienza A, Brancadoro L, de Lorenzis G, Failla O (2016) Evidence for a
464 Sympatric Origin of Ribolla gialla, Gouais Blanc and Schiava cultivars (*V. vinifera* L.).
465 *South African J Enol Vitic* 35:149–156.
- 466 4. Imazio S, De Lorenzis G, Scienza A, Failla O, Vouillamoz J, Korosec-Koruza Z, Rusjan
467 D, Nikolao N (2014) ‘Ribolla Gialla’ from North Eastern Italy, ‘Rebula’ from Northern
468 Balkans and ‘Robola’ from Ionian Islands; Do They Belong to the Same Population
469 Variety or Are They Genetically Different? *Acta Horti* 1046:645–652.
- 470 5. Bavčar D, Baša Česnik H, Čuš F, Vanzo A, Gašperlin L, Košmerl T (2011) Impact of
471 alternative skin contact procedures on the aroma composition of white wine. *South*
472 *African J Enol Vitic* 32:190–203.
- 473 6. Bavčar D, Baša Česnik H, Čuš F, Košmerl T (2011) The influence of skin contact during
474 alcoholic fermentation on the aroma composition of Ribolla Gialla and Malvasia Istriana
475 *Vitis vinifera* (L.) grape wines. *Int J Food Sci Technol* 46:1801–1808.
- 476 7. Rapp A, Versini G (1995) Influence of nitrogen compounds in grapes on aroma
477 compounds of wines. *Dev Food Sci* 37:1659–1694
- 478 8. Pueyo E, Martínez-Rodríguez A, Polo MC, Santa-María G, Bartolomé B (2000) Release
479 of Lipids during Yeast Autolysis in a Model Wine System. *J Agric Food Chem* 46:116–
480 122.
- 481 9. Swiegers JH, Bartowsky EJ, Henschke PA, Pretorius IS (2005) Yeast and bacterial
482 modulation of wine aroma and flavour. *Aust J Grape Wine Res* 11:139-173.

- 1
2
3 483 10. Callejón RM, Margulies B, Hirson GD, Ebeler SE (2012) Dynamic changes in volatile
4
5 484 compounds during fermentation of Cabernet Sauvignon grapes with and without skins.
6
7 485 Am J Enol Vitic 63:301–312.
8
9
10 486 11. Hirst MB, Richter CL (2016) Review of Aroma Formation through Metabolic Pathways
11
12 487 of *Saccharomyces cerevisiae* in Beverage Fermentations. Am J Enol Vitic 67:361–370.
13
14 488 12. Waterhouse AL, Sacks GL, Jeffery DW (2016) Understanding Wine Chemistry. John
15
16 489 Wiley & Sons, Chichester (UK).
17
18
19 490 13. Gallart M, Lopez-Tamames E, Suberbiola G, Buxaderas Susana (2002) Influence of
20
21 491 Fatty Acids on Wine Foaming. J Agric Food Chem 50:7042–7045.
22
23
24 492 14. Hoenicke K, Simat TJ, Steinhart H, Köhler HJ, Schwab A (2001) Determination of Free
25
26 493 and Conjugated Indole-3-Acetic Acid, Tryptophan, and Tryptophan Metabolites in
27
28 494 Grape Must and Wine. J Agric Food Chem 49:5494–5501.
29
30
31 495 15. Schneider V (2014) Atypical Aging Defect: Sensory Discrimination, Viticultural Causes,
32
33 496 and Enological Consequences. A Review. Am J Enol Vitic 65:277–284.
34
35 497 16. Dziadas M, Jeleń HH (2010) Analysis of terpenes in white wines using SPE-SPME-
36
37 498 GC/MS approach. Anal Chim Acta 677:43–49.
38
39
40 499 17. Louw L, Tredoux AGJ, Rensburg PV, Kidd M, Naes T, Nieuwoudt HH (2010)
41
42 500 Fermentation-derived Aroma Compounds in Varietal Young Wines from South Africa.
43
44 501 South African J Enol Vitic 31:213–225.
45
46 502 18. D’Onofrio C, Matarese F, Cuzzola A (2017) Study of the terpene profile at harvest and
47
48 503 during berry development of *Vitis vinifera* L. aromatic varieties Aleatico, Brachetto,
49
50 504 Malvasia di Candia aromatica and Moscato bianco. J Sci Food Agric 97:2898–2907.
51
52
53 505 19. Vilanova M, Genisheva Z, Graña M, Oliveira JM (2013) Determination of Odorants in
54
55 506 Varietal Wines from International Grape Cultivars (*Vitis vinifera*) Grown in NW Spain.
56
57 507 South African J Enol Vitic 34:212–222.
58
59
60 508 20. Comuzzo P, Marconi M, Zanella G, Querzè M (2018) Pulsed electric field processing of

- 1
2
3 509 white grapes (cv. Garganega): Effects on wine composition and volatile compounds.
4
5 510 Food Chem 264:16–23.
6
7
8 511 21. Loira I, Morata A, Comuzzo P, Callejo MJ, González C, Calderón F, Suárez-Lepe JA
9
10 512 (2015) Use of *Schizosaccharomyces pombe* and *Torulaspora delbrueckii* strains in
11
12 513 mixed and sequential fermentations to improve red wine sensory quality. Food Res Int
13
14 514 76:325–333.
15
16
17 515 22. Carlin S, Vrhovsek U, Franceschi P, Lotti C, Bontempo L, Camin F, Toubiana D, Zottele
18
19 516 F, Toller G, Fait A, Mattivi F (2016) Regional features of northern Italian sparkling
20
21 517 wines, identified using solid-phase micro extraction and comprehensive two-
22
23 518 dimensional gas chromatography coupled with time-of-flight mass spectrometry. Food
24
25 519 Chem 208:68–80.
26
27
28 520 23. Arapitsas P, Guella G, Mattivi F (2018) The impact of SO₂ on wine flavanols and indoles
29
30 521 in relation to wine style and age. Sci Rep 8:858.
31
32
33 522 24. Della Corte A, Chitarrini G, Di Gangi IM, Masuero D, Soini E, Mattivi F, Vrhovsek U
34
35 523 (2015) A rapid LC–MS/MS method for quantitative profiling of fatty acids, sterols,
36
37 524 glycerolipids, glycerophospholipids and sphingolipids in grapes. Talanta 140:52–61.
38
39
40 525 25. Commission Regulation (EC) No 607/2009 (2009) Laying down certain detailed rules
41
42 526 for the implementation of Council Regulation (EC) No 479/2008 as regards protected
43
44 527 designations of origin and geographical indications, traditional terms, labeling and
45
46 528 presentation of certain wine sector products. Off J Eur Communities 60–139.
47
48
49 529 26. Patynowski RJ, Jiranek V, Markides AJ (2002) Yeast viability during fermentation and
50
51 530 sur lie ageing of a defined medium and subsequent growth of *Oenococcus oeni*. Aust J
52
53 531 Grape Wine Res 8:62–69.
54
55
56 532 27. Lambrechts MG, Pretorius IS (2000) Yeast and its Importance to Wine Aroma - A
57
58 533 Review. South African J Enol Vitic 21:97-129.
59
60 534 28. Fariña L, Villar V, Ares G, Carrau F, Dellacassa E, Boido E (2015) Volatile composition

- 1
2
3 535 and aroma profile of Uruguayan Tannat wines. *Food Res Int* 69:244–255.
4
5 536 29. Ribéreau-Gayon P, Glories Y, Maujean A, Dubourdieu D (2006) *Handbook of Enology*.
6
7 537 Vol. 2. *The Chemistry of Wine Stabilization and Treatments*, 2nd ed. John Wiley & Sons,
8
9 538 Chichester (UK).
10
11
12 539 30. Ferreira V, Fernandez P, Peiia C, Escudero A (1995) Investigation on the Role Played
13
14 540 by Fermentation Esters in the Aroma-of Young Spanish Wines by Multivariate Analysis.
15
16 541 *J Sci Food Agric* 67:381–392.
17
18
19 542 31. Saerens SMG, Delvaux FR, Verstrepen KJ, Thevelein JM (2010) Production and
20
21 543 biological function of volatile esters in *Saccharomyces cerevisiae*. *Microb Biotechnol*
22
23 544 3:165–177.
24
25
26 545 32. Boido E, Medina K, Farina L, Carrau F, Versini G, Dellacassa E (2009) The Effect of
27
28 546 Bacterial Strain and Aging on the Secondary Volatile Metabolites Produced during
29
30 547 Malolactic Fermentation of Tannat Red Wine. *J Agric Food Chem* 57:6271–6278.
31
32
33 548 33. Guth H (1997) Quantitation and Sensory Studies of Character Impact Odorants of
34
35 549 Different White Wine Varieties. *J. Agric. Food Chem* 43:35–37.
36
37
38 550 34. Voce S, Pizzamiglio G, Comuzzo P, Sivilotti S, Mosetti D, Bigot G, Lonardi A (2018)
39
40 551 Defogliazione e aromi della Ribolla Gialla. *Vigne, Vini Qual* 38–41.
41
42
43 552 35. Mateo JJ, Jimenez M (2000) Monoterpenes in grape juice and wines. *J Chromatogr A*
44
45 553 881:557–567.
46
47 554 36. Sefton MA, Skouroumounis GK, Elsey GM, Taylor DK (2011) Occurrence, sensory
48
49 555 impact, formation, and fate of damascenone in grapes, wines, and other foods and
50
51 556 beverages. *J Agric Food Chem* 59:9717–9746.
52
53
54 557 37. Escudero A, Campo E, Fariña L, Cacho J, Ferreira V (2007) Analytical Characterization
55
56 558 of the Aroma of Five Premium Red Wines . Insights into the Role of Odor Families and
57
58 559 the Concept of Fruitiness of Wines. *J Agric Food Chem* 55:4501–4510.
59
60 560 38. Belitz HD, Grosch W, Schieberle P (2004) *Food Chemistry*, 3rd ed. Springer, Berlin.

- 1
2
3 561 39. Black CA, Parker M, Siebert TE, Capone DL, Francis IL (2015) Terpenoids and their
4
5 562 role in wine flavour: Recent advances. *Aust J Grape Wine Res* 21:582–600.
6
7 563 40. Mattivi F, Arapitsas P, Perenzoni D, Guella G (2015) In: Ebeler SB, Sacks G, Vidal S,
8
9 564 Winterhalter P (eds) *Advances in Wine Research*. ACS Publications, Washington (USA)
10
11 565 41. Hoenicke K, Borchert O, Ning K G, Simat TJ (2002) Untypical Aging Off-Flavor "
12
13 566 in Wine: Synthesis of Potential Degradation Compounds of Indole-3-acetic Acid and
14
15 567 Kynurenine and Their Evaluation as Precursors of 2-Aminoacetophenone. *J Agric Food*
16
17 568 *Chem* 50:4303–4309.
18
19 569 42. Ramsay AM, Douglas LJ (1979) Effects of Phosphate Limitation of Growth on the Cell-
20
21 570 wall and Lipid Composition of *Saccharomyces cerevisiae*. *J Gen Microbiol* 110:185–
22
23 571 191.
24
25 572 43. Piton F, Charpentier M, Troton D (1988) Cell Wall and Lipid Changes in *Saccharomyces*
26
27 573 *cerevisiae* during Aging of Champagne Wine. *Am J Enol Vitic* 39:221–225.
28
29 574 44. Pueyo E, Martin-Alvarez P, Polo MC (1995) Relationship Between Foam Characteristics
30
31 575 and Chemical Composition in Wines and Cavas (Sparkling Wines). *Am J Enol Vitic*
32
33 576 46:518–524.
34
35 577 45. Vilanova M, Sieiro C (2006) Determination of free and bound terpene compounds in
36
37 578 Albariño wine. *J Food Compos Anal* 19:694–697.
38
39 579 46. Francis IL, Newton JL (2005) Determining wine aroma from compositional data. *Aust J*
40
41 580 *Grape Wine Res* 11:114–126
42
43 581 47. Sansone-Land A, Takeoka GR, Shoemaker CF (2014) Volatile constituents of
44
45 582 commercial imported and domestic black-ripe table olives (*Olea europaea*). *Food Chem*
46
47 583 149:285–295.
48
49 584 48. Zhu J, Xiao Z (2018) Characterization of the Major Odor-Active Compounds in Dry
50
51 585 Jujube Cultivars by Application of Gas Chromatography–Olfactometry and Odor
52
53 586 Activity Value. *J Agric Food Chem* 66:7722–7734.
54
55
56
57
58
59
60

- 1
2
3 587 49. Ferreira V, Lopez R, Cacho JF (2000) Quantitative determination of the odorants of
4
5 588 young red wines from different grape varieties. *J Sci Food Agric* 80:1659–1667.
6
7 589 50. Elsharif SA, Buettner A (2018) Structure – Odor Relationship Study on Geraniol, Nerol,
8
9 and Their Synthesized Oxygenated Derivatives. *J Agric Food Chem* 66:2324–2333.
10 590
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For Peer Review

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4 593 **Figure Captions**
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6 594 **Figure 1.** Geographical position of Friuli Venezia Giulia region. Grey areas mark the
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8 595 municipalities in which the wineries that supplied wine samples are located. The number of
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10 596 samples/wineries for each municipality are also reported.
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602 **Table 1. Chemical composition of Ribolla Gialla commercial wines. SD: standard**
 603 **deviation; MIN: minimum value; MAX: maximum value; CV: coefficient of variation.**

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Parameter	Mean \pm SD	MIN	MAX	CV
alcoholic strength (% v/v)	11.68 \pm 0.64	10.40	12.70	0.05
reducing sugars (g L ⁻¹)	10.64 \pm 2.91	6.00	18.87	0.27
titratable acidity (g L ⁻¹)	5.94 \pm 0.48	5.00	7.25	0.08
volatile acidity (g L ⁻¹)	0.31 \pm 0.06	0.20	0.48	0.21
pH	3.22 \pm 0.12	3.03	3.51	0.04
malic acid (g L ⁻¹)	2.08 \pm 0.93	n.d. ^a	3.79	0.45
lactic acid (g L ⁻¹)	0.26 \pm 0.63	n.d.	2.75	2.45
tartaric acid (g L ⁻¹)	2.83 \pm 0.36	2.30	3.93	0.13

^a n.d.: not detected

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Table 2. Non-variatal aroma compounds detected in Ribolla Gialla commercial sparkling wines by different analytical approaches. Concentrations are expressed in $\mu\text{g L}^{-1}$.

Compounds	AM ^a	IM ^b	Mean \pm SD ^c	MIN ^d	MAX ^e	CV ^f	ODT ($\mu\text{g L}^{-1}$) ^g
<i>Acids</i>							
acetic acid	LLE	MS RI S	2624 \pm 1206	1000	6388	0.46	200000 [33]
2-methylpropanoic acid	LLE	MS RI S	376 \pm 168	n.d. ^h	877	0.45	230 [28]
butanoic acid	LLE	MS RI S	423 \pm 126	218	812	0.30	10000 [33]
3-methylbutanoic acid	LLE	MS RI S	228 \pm 242	n.d.	1103	1.06	250 [28]
hexanoic acid	LLE	MS RI S	4196 \pm 1298	2111	7817	0.31	3000 [33]
heptanoic acid	LLE	MS RI	56 \pm 70	n.d.	224	1.24	-
octanoic acid	LLE	MS RI S	6923 \pm 2390	2878	15070	0.35	500 [45]
nonanoic acid	SPME	MS S	454 \pm 96	239	708	0.21	-
decanoic acid	LLE	MS RI S	872 \pm 736	n.d.	4043	0.84	1000 [45]
benzoic acid	SPME	MS S	5 \pm 3	2	17	0.61	-
dodecanoic acid	SPME	MS S	3 \pm 2	1	9	0.67	-
hexadecanoic acid	LLE	MS RI	2065 \pm 2072	n.d.	7714	1.00	-
<i>Alcohols</i>							
2-methyl-1-propanol	LLE	MS RI	8784 \pm 2825	5569	18165	0.32	40000 [33]
1-butanol	LLE	MS RI S	210 \pm 103	n.d.	515	0.49	40000 [46]
2- and 3-methyl-1-butanol	LLE	MS RI S	125666 \pm 24151	85297	190584	0.19	40000 [28]
3-ethoxy-1-propanol	LLE	MS RI	172 \pm 349	n.d.	1651	2.03	-
2-phenylethanol	LLE	MS RI S	23645 \pm 12214	9954	63961	0.52	10000 [33]
<i>C6 compounds</i>							
1-hexanol	LLE	MS RI S	957 \pm 231	603	1443	0.24	2500 [28]

1	<i>trans</i> -3-hexen-1-ol	SPME	MS S	3 ± 1	2	9	0.41	1000	[28]
2	<i>trans</i> -2-hexen-1-ol	SPME	MS S	7 ± 6	n.d.	35	0.84	-	
3	<i>cis</i> -3-hexen-1-ol	LLE	MS RI S	39 ± 118	n.d.	646	3.01	400	[28]
4	hexanal	SPME	MS S	2 ± 7	n.d.	42	3.19	5	[47]
5	<i>trans</i> -2-hexenal	SPME	MS S	4 ± 1	1	8	0.36	82	[48]
6									
7									
8									
9	<i>Diols</i>								
10	2,3-butanediol	LLE	MS RI	2987 ± 1079	1497	5601	0.36	-	
11	1-2 propandiol	LLE	MS RI	674 ± 250	258	1216	0.37	-	
12									
13									
14	<i>Esters</i>								
15	2-methyl-1-propanol acetate	SPME	MS S	0 ± 0	n.d.	1	1.07	-	
16	ethyl butanoate	SPME	MS S	44 ± 12	18	64	0.26	20	[33]
17	3-methyl-1-butanol acetate	LLE	MS RI	1182 ± 1168	n.d.	5908	0.99	30	[33]
18	methyl hexanoate	SPME	MS S	1 ± 0	n.d.	2	0.38	-	
19	ethyl hexanoate	LLE	MS RI S	554 ± 268	132	1165	0.48	65	[45]
20	hexyl acetate	LLE	MS RI	8 ± 32	n.d.	171	3.94	2	[47]
21	ethyl lactate	LLE	MS RI	20976 ± 33732	1589	141330	1.61	60000	[28]
22	methyl octanoate	SPME	MS S	3 ± 1	1	6	0.39	200	[47]
23	ethyl octanoate	LLE	MS RI S	433 ± 326	n.d.	982	0.75	580	[45]
24	3-methylbutyl lactate	SPME	MS S	1 ± 0	n.d.	1	0.30	-	
25	ethyl-2-hydroxy-4-methylpentanoate	SPME	MS S	6 ± 3	1	14	0.46	-	
26	methyl decanoate	SPME	MS S	0 ± 0	n.d.	1	0.54	4	[48]
27	ethyl 3-hydroxybutanoate	LLE	MS RI	42 ± 87	n.d.	295	2.04	-	
28	ethyl decanoate	LLE	MS RI S	36 ± 80	n.d.	307	2.25	200	[45]
29	methyl ethyl succinate	SPME	MS S	96 ± 54	25	283	0.56	-	
30	3-methylbutyl octanoate	SPME	MS S	5 ± 2	2	9	0.35	125	[49]
31	diethyl succinate	LLE	MS RI	2555 ± 1849	263	8391	0.72	100000	[28]
32	ethyl 9-decenoate	SPME	MS S	9 ± 11	1	50	1.15	-	
33	methyl salicylate	SPME	MS S	5 ± 6	1	29	1.31	40	[47]
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1	2-phenylethyl acetate	LLE	MS RI S	136 ± 341	n.d.	1707	2.50	250	[33]
2	ethyl dodecanoate	LLE	MS RI	57 ± 125	n.d.	456	2.20	-	
3	diethyl malate	LLE	MS RI	5544 ± 4199	679	18639	0.76	760000	[28]
4	ethyl tetradecanoate	LLE	MS RI	57 ± 140	n.d.	504	2.48	-	
5	diethyle 2-hydroxypentanedioate	LLE	MS RI	605 ± 406	n.d.	1472	0.67	-	
6	ethyl hexadecanoate	LLE	MS RI S	1069 ± 1764	n.d.	6185	1.65	-	
7	ethyl hydrogen succinate	SPME	MS S	57 ± 30	19	125	0.52	1000000	[28]
8									
9									
10									
11									
12	<u>Carbonyl compounds</u>								
13	3-hydroxy-2-butanone (acetoin)	LLE	MS RI S	599 ± 311	162	1230	0.52	150000	[49]
14	furfural	SPME	MS S	29 ± 25	7	138	0.86	770	[28]
15	benzaldehyde	SPME	MS S	14 ± 28	4	164	2.05	350	[47]
16	3,4-dimethyl benzaldehyde	SPME	MS S	1 ± 0	n.d.	2	0.34	-	
17									
18									
19									
20	<u>Others</u>								
21	dihydro-2-methyl-3(2 <i>H</i>)-thiophenone	SPME	MS S	4 ± 3	1	11	0.59	-	
22	dihydro-2(3 <i>H</i>)-furanone (γ -butyrolactone)	LLE	MS RI S	691 ± 204	316	1237	0.30	1000	[28]
23	methionol	LLE	MS RI	232 ± 145	n.d.	623	0.62	1000	[49]

^a AM: analytical method; ^b IM: identification method (S comparison of mass spectra and retention time with those of standard compounds; RI comparison of order of elution with those reported in literature; MS comparison of mass spectra with those reported in mass spectrum libraries); ^c SD: standard deviation; ^d MIN: minimum value; ^e MAX: maximum value; ^f CV: coefficient of variation; ^g ODT: odor detection threshold; ^h n.d.: not detected

Table 3. Free and bound terpenes and norisoprenoids detected in Ribolla Gialla commercial sparkling wines by different analytical approaches.Concentrations are expressed in $\mu\text{g L}^{-1}$.

Compounds	AM ^a	IM ^b	Mean \pm	SD ^c	MIN ^d	MAX ^e	CV ^f	ODT ($\mu\text{g L}^{-1}$) ^g	
<i>Free terpenes</i>									
β -myrcene	SPME	MS S	0 \pm 1		n.d. ^h	7	3.56	-	
limonene	SPME	MS S	2 \pm 1		1	4	0.33	200 [47]	
<i>cis</i> -linalool oxide (furanic)	SPE	MS RI S	4 \pm 4		n.d.	12	0.88	6000 [28]	
<i>trans</i> -linalool oxide (furanic)	SPE	MS RI S	1 \pm 2		n.d.	9	1.61	6000 [28]	
linalool	SPE	MS RI S	10 \pm 15		n.d.	80	1.51	50 [28]	
terpinen-4-ol	SPME	MS S	1 \pm 0		1	2	0.34	340 [47]	
α -terpineol	SPE	MS RI S	23 \pm 34		n.d.	194	1.49	250 [49]	
β -citronellol	SPE	MS RI S	2 \pm 4		n.d.	18	1.81	100 [33]	
nerol	SPME	MS S	1 \pm 1		n.d.	6	0.80	60 [50]	
geraniol	SPE	MS RI S	4 \pm 10		n.d.	39	2.87	30 [49]	
geranic acid	SPME	MS S	7 \pm 4		2	23	0.65	-	
<i>Free C₁₃-norisoprenoids</i>									
β -damascenone	SPE	MS RI	5 \pm 6		n.d.	28	1.27	7 [36]	
3-oxo- α -ionol	SPE	MS RI	10 \pm 11		n.d.	42	1.11	-	
riesling acetal	SPME	MS S	4 \pm 2		n.d.	7	0.49	-	
vitispirane (isomer 1)	SPME	MS S	4 \pm 3		2	14	0.68	800 [38]	
1,1,6-trimethyl-1,2-dihydronaphthalene (TDN)	SPME	MS S	5 \pm 6		1	28	1.19	20 [38]	
vitispirane (isomer 2)	SPME	MS S	14 \pm 9		3	37	0.61	800 [38]	
<i>Bound terpenes</i>									
nerol	SPE	MS RI S	3 \pm 4		n.d.	20	1.54		

1	geraniol		SPE	MS RI S	13 ± 13	n.d.	51	1.03
2	1-hydroxylinalool		SPE	MS RI	2 ± 4	n.d.	15	1.99
3	geranic acid		SPE	MS RI	8 ± 9	n.d.	36	1.13
4								
5								
6								
7	<u>Bound C₁₃-norisoprenoids</u>							
8	3-hydroxy-β-damascone		SPE	MS RI	6 ± 10	n.d.	50	1.83
9	<i>trans,trans</i> -2,6-dimethyl-2,6-octadiene-1,8-diol	(Z8-	SPE	MS RI	11 ± 13	n.d.	56	1.15
10	hydroxygeraniol)							
11	3-oxo-α-ionol		SPE	MS RI	40 ± 32	n.d.	135	0.81
12	3-oxo-7,8-dihydro-α-ionol (blumenol C)		SPE	MS RI	16 ± 22	n.d.	103	1.38
13	3-hydroxy-7,8-dihydro-β-ionol		SPE	MS RI	1 ± 2	n.d.	7	2.02
14								
15								

^a AM: analytical method; ^b IM: identification method (S comparison of mass spectra and retention time with those of standard compounds; RI comparison of order of elution with those reported in literature; MS comparison of mass spectra with those reported in mass spectrum libraries); ^c SD: standard deviation; ^d MIN: minimum value; ^e MAX: maximum value; ^f CV: coefficient of variation; ^g ODT: odor detection threshold; ^h n.d.: not detected

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607 **Table 4. Tryptophan metabolites detected in Ribolla Gialla commercial sparkling wines.**608 **Concentrations are expressed in mg L⁻¹. SD: standard deviation; MIN: minimum value;**609 **MAX: maximum value; CV: coefficient of variation.**

Compound ^a	Mean	±	SD	MIN	MAX	CV
ILA-SO ₃ H	8.53	±	10.31	0.66	54.20	1.21
TOL-SO ₃ H	3.73	±	2.55	0.23	12.49	0.68
TYL	2.55	±	0.97	0.93	4.79	0.38
TYR	1.13	±	0.49	0.08	1.98	0.43
ILA-GLU-SO ₃ H	0.53	±	0.37	0.12	1.79	0.70
PHE	0.51	±	0.26	0.03	0.99	0.51
TRP-EE	0.44	±	0.21	0.11	0.89	0.48
TOL	0.29	±	0.27	0.01	1.08	0.93
ABA	0.22	±	0.11	0.02	0.46	0.50
IAA-SO ₃ H	0.20	±	0.07	0.13	0.39	0.35
ABA-GLU	0.13	±	0.05	0.04	0.26	0.38
TRP	0.13	±	0.10	0.01	0.39	0.77
ILA-GLU	0.10	±	0.05	0.01	0.25	0.50
TYR-EE	0.08	±	0.03	0.02	0.13	0.38
ILA	0.03	±	0.03	n.d. ^b	0.12	1.00
KYNA	0.02	±	0.01	n.d.	0.04	0.50
AA	0.01	±	0.06	n.d.	0.33	6.00
IAA-ASP	4.41	±	2.87	n.d.	12.35	0.65
N-TYR-EE	3.86	±	2.11	n.d.	9.63	0.55
KYN	2.42	±	1.98	n.d.	9.24	0.82
IAA	1.75	±	1.63	n.d.	8.84	0.93
N-TRP-EE	1.08	±	0.81	n.d.	3.47	0.75

^a ILA-SO₃H: indole-lactic acid-2-sulfonate; TOL-SO₃H: tryptophol-2-sulfonate; TYL: tryptophol; TYR: tyrosine; ILA-GLU-SO₃H: sulfonated indole-3-lactic acid; PHE: phenylalanine; TRP-EE: tryptophan-ethyl ester; TOL: tryptophol; ABA: abscisic acid; IAA-SO₃H: sulfonated indole 3-acetic acid; ABA-GLU: glucoside of abscisic acid; TRP: tryptophan; ILA-GLU: indole 3-lactic acid glucoside; TYR-EE: tyrosine-ethyl ester; ILA: indole 3-lactic acid; KYNA: kynurenic acid; AA: anthranilic acid; IAA-ASP: indole 3-acetic acid conjugate with aspartic acid; N-TYR-EE: N-acetyl-tyrosine-ethyl ester; KYN: kynurenine; IAA: indole 3-acetic acid; N-TRP-EE: N-acetyl-tryptophan-ethyl ester.

^b n.d.: not detected

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612 **Table 5. Lipid compounds detected in Ribolla Gialla commercial sparkling wines.**613 **Concentrations are expressed in mg L⁻¹. SD: standard deviation; MIN: minimum value;**614 **MAX: maximum value; CV: coefficient of variation.**

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Compounds	Mean	±	SD	MIN	MAX	CV
<u>Fatty acids</u>						
palmitic acid	9.24	±	13.52	5.21	11.49	1.46
stearic acid	6.49	±	9.32	4.43	8.40	1.44
myristic acid	0.31	±	0.84	0.11	0.50	2.68
oleic acid + <i>cis</i> -vaccenic acid	0.19	±	0.32	0.12	0.26	1.72
arachidic acid	0.15	±	0.40	0.08	0.26	2.70
<i>cis</i> -11-eicosanoic acid	0.11	±	0.03	0.10	0.12	0.28
heptadecanoic acid	0.06	±	0.14	0.04	0.10	2.19
myristoleic acid	0.05	±	0.04	0.05	0.06	0.78
linoleic acid	0.04	±	0.08	0.02	0.06	2.29
lignoceric acid	0.03	±	0.16	n.d. ^a	0.09	5.52
behenic acid	0.02	±	0.08	0.01	0.05	4.21
palmitoleic acid	0.02	±	0.36	n.d.	0.14	20.00
linolenic acid	0.01	±	0.04	0.00	0.03	8.00
<u>Sterols</u>						
ergosterol	0.18	±	0.97	0.03	0.35	5.54
lupeol	0.10	±	1.10	0.01	0.48	10.89
<u>Glycerolipids</u>						
1-linoleoyl-rac-glycerol	0.02	±	0.02	0.01	0.02	1.18
1-oleoyl-rac-glycerol	0.01	±	0.02	0.01	0.02	1.43
glyceryl tripalmitoleate	0.01	±	0.09	n.d.	0.04	6.43
<u>Fatty acid esters</u>						
ethyl stearate	0.14	±	0.18	0.10	0.17	1.31
ethyl palmitate	0.13	±	0.22	0.07	0.17	1.64
ethyl oleate	0.03	±	0.05	0.02	0.05	1.47
ethyl linoleate	0.01	±	0.01	0.00	0.01	2.00
methyl palmitate	0.08	±	1.20	n.d.	0.50	15.19
methyl stearate	0.04	±	0.36	n.d.	0.15	9.73
methyl oleate	0.01	±	0.06	n.d.	0.03	6.67
<u>Triacylglycerols</u>						
tripentadecanoin	0.01	±	0.04	n.d.	0.02	5.00

616 ^a n.d.: not detected

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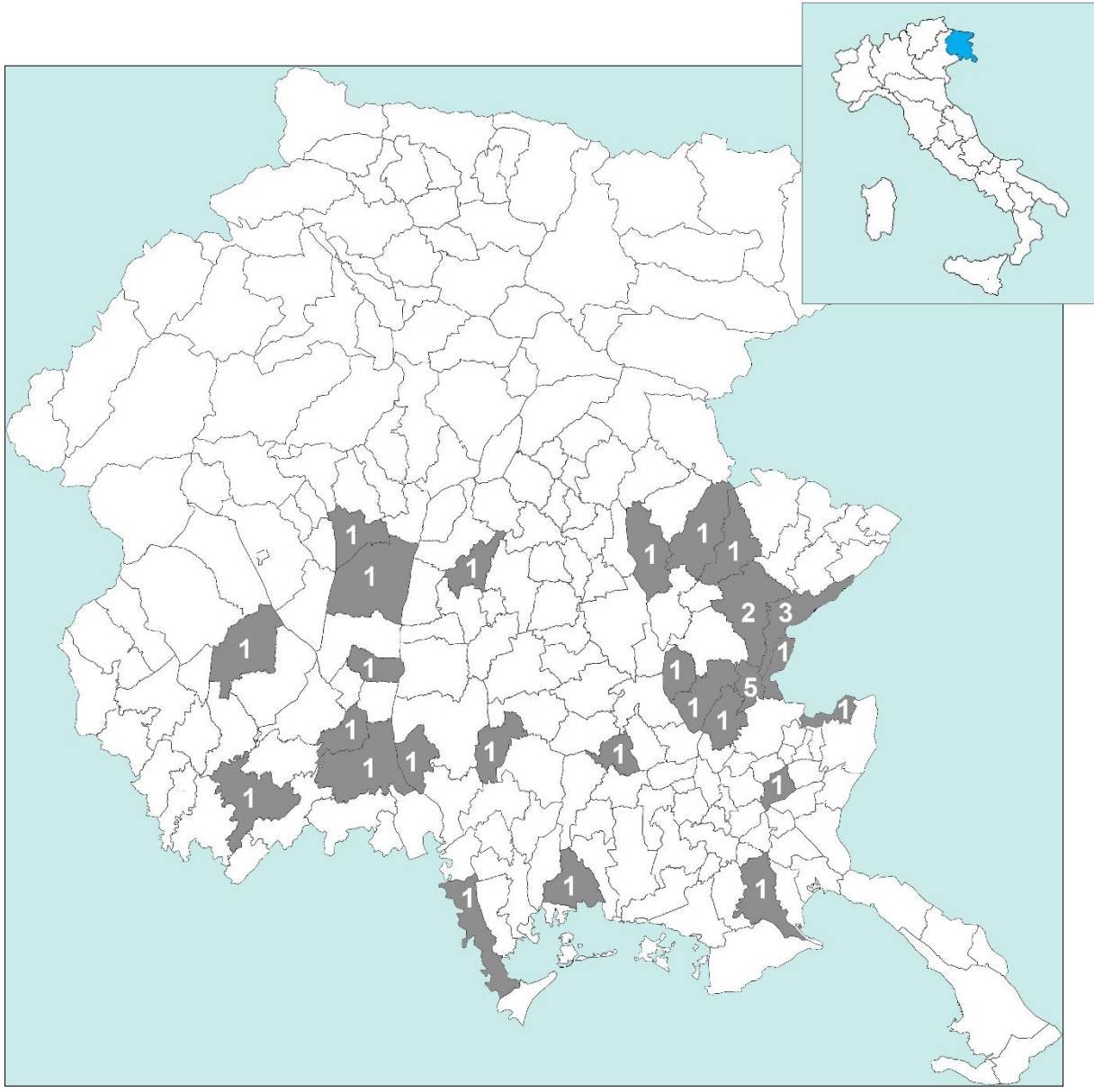


Fig. 1.