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Synthesis and Analysis of the Conformational Preferences of 5-Aminomethyloxazolidine-2,4-dione Scaffolds: First Examples of β^2 - and $\beta^{2,2}$ -Homo-Freidinger Lactam Analogues

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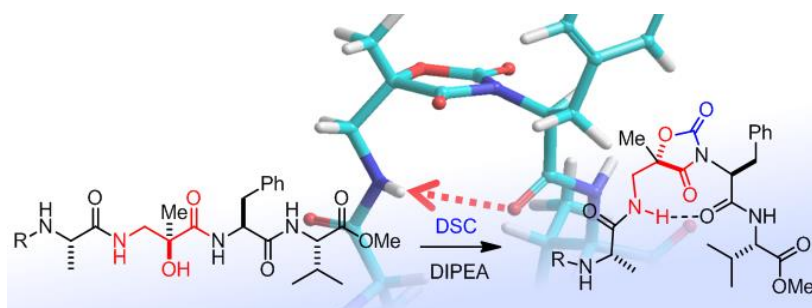


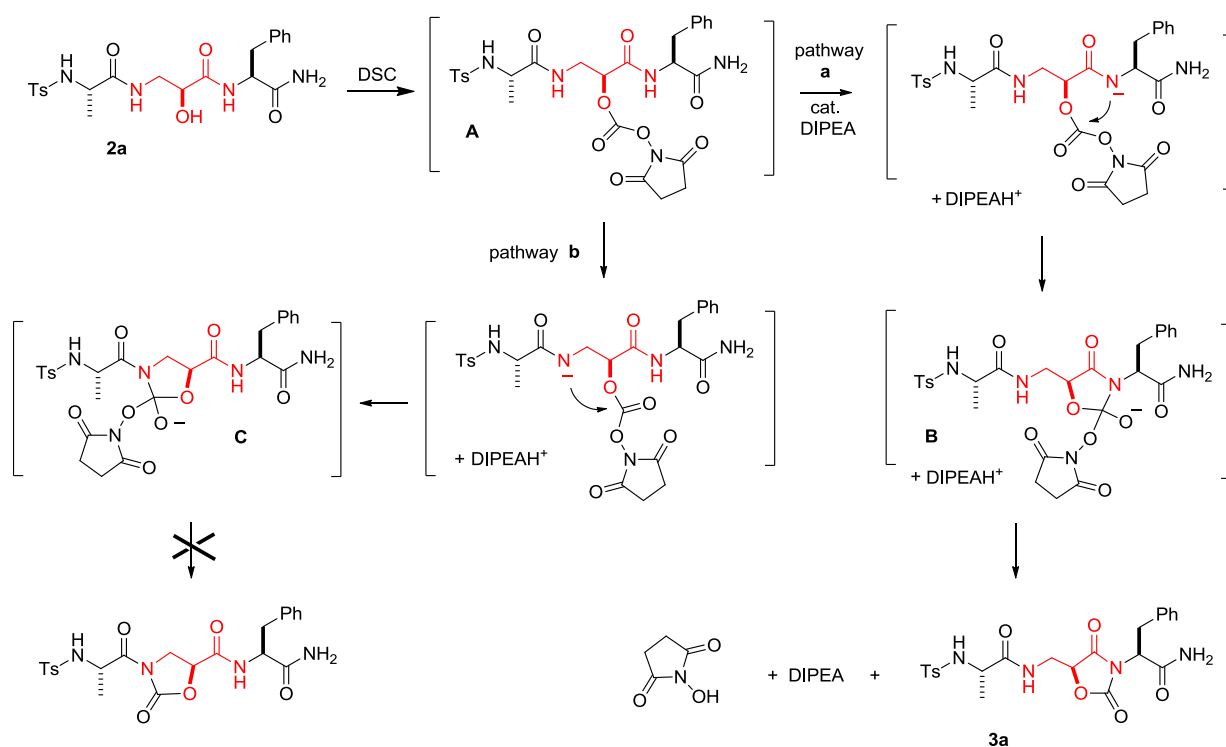
Table of contents

Plausible reaction pathway for the synthesis of Amo-peptides	p. S2
VT-NMR experiments for 3a , 3b , 3c , 3f , 5 , in CDCl ₃	p. S3
Circular dichroism of 3d , 3d , 10a , and 10b in DCM	p. S3
IR analyses of 3d , 3e , 10a , and 10b in DCM	p. S3
Variation of NH chemical shift of 3a , 3c , 3f , in CDCl ₃ /0-8% [D ₆]DMSO	p. S4
ROESY cross peaks for 3d , 3e , 10a , 10b in 8:2 [D ₆]DMSO/H ₂ O	p. S4
Unrestrained MD simulation of 10a in TIP3P water	p. S8
Color versions of Fig. 3, 5, and 6	p. S9
¹ H and ¹³ C-NMR spectra of compounds 3a-h , 5 , 10a , 10b	p. S10

Plausible reaction pathway for the synthesis of Amo-peptides.

A plausible reaction pathway for the cyclization of the model peptide Ts-Ala-*iso*Ser-PheNH₂ to Ts-Ala-Amo-PheNH₂ with DSC and catalytic DIPEA is depicted in Scheme S1. In the proposed mechanism, the cyclization to Amo proceeds *via* the *iso*Ser-O-succinimidyl carbonate intermediate **A**. The intermediate **A** is deprotonated by DIPEA at PheNH, giving the 5-membered anionic intermediate **B** with endocyclic C=O (path **a**). The loss of 2,5-dioxopyrrolidin-1-olate leaving group, rapidly protonated by DIPEAH⁺, leads to the Amo-peptide **3** and DIPEA, which can be utilized in catalytic amount. The intermediate **A** can be deprotonated at *iso*SerNH as well (path **b**); this could give access to the alternative 5-membered cyclic anionic intermediate **C** with hexocyclic C=O, precursor of a Oxd-peptide.

Preliminary computations were performed for the intermediates **B** and **C** employing Density Functional Theory; a systematic conformational analysis for the structures was done at the B3LYP/6-311++G(d,p) level. Optimization was performed by conjugate gradient algorithm, convergence at 0.001. The results indicate that the intermediate **B** is about 2.0 Kcal mol⁻¹ more stable than the alternative intermediate **C**.



Scheme S1. Cyclization of the model peptide Ts-Ala-*iso*Ser-PheNH₂ **2** to Ts-Ala-Amo-PheNH₂ **3**.

Table S1. $\Delta\delta/\Delta t$ values (p.p.b./°K) for the amide protons of peptides **3a**, **3b**, **3c**, **3f**, **5**, in CDCl₃.

Compd	sequence	AlaNH	AmoNH	ValNH	CONH ₂
3a	Ts-Ala-(S)-Amo-PheNH ₂	-15.1	-8.1	-	-14.3/-9.7
3b	Ts-Ala-(R)-Amo-PheNH ₂	-8.6	-6.2	-	-8.6/-8.0
3c	Ts-Ala-(S)-Amo-(R)-PheNH ₂	-8.4	-6.0	-	-7.7/-9.5
3f	Boc-Ala-(S)-Amo-Phe-ValOMe	-4.5	-5.0	-4.5	-
5	Ts-Ala-(R)-5-hydroxyAmo-PheNH ₂	-14.7	-10.0	-	-13.7/-11.0

*Circular dichroism of **3d**, **3e**, **10a**, and **10b** recorded in DCM.*

ECD spectra were recorded from 200 to 400 nm at 25 °C. 1 mM solutions were made up in spectral grade solvents and run in a 0.1 cm quartz cell (Figure S1). Data are reported in molar ellipticity [θ] (deg cm² dmol⁻¹).

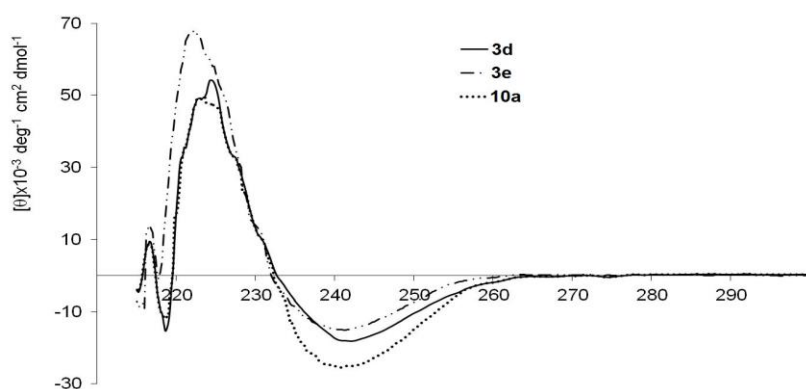


Figure S1. ECD spectra of **3d**, **3e**, **10a**, and **10b** recorded in DCM at r.t.; **3d** and **10b** are practically superimposed.

*IR analyses **3d**, **3e**, **10a**, and **10b** in DCM.*

The compounds were dried in vacuo, and all the sample preparations were performed under nitrogen atmosphere. All infrared spectra were obtained for 1 mM solutions in dry DCM at 297 °K at 2 cm⁻¹ resolution, using a 1 mm NaCl solution cell and a FT-IR spectrometer (64 scans). (Figure S2).

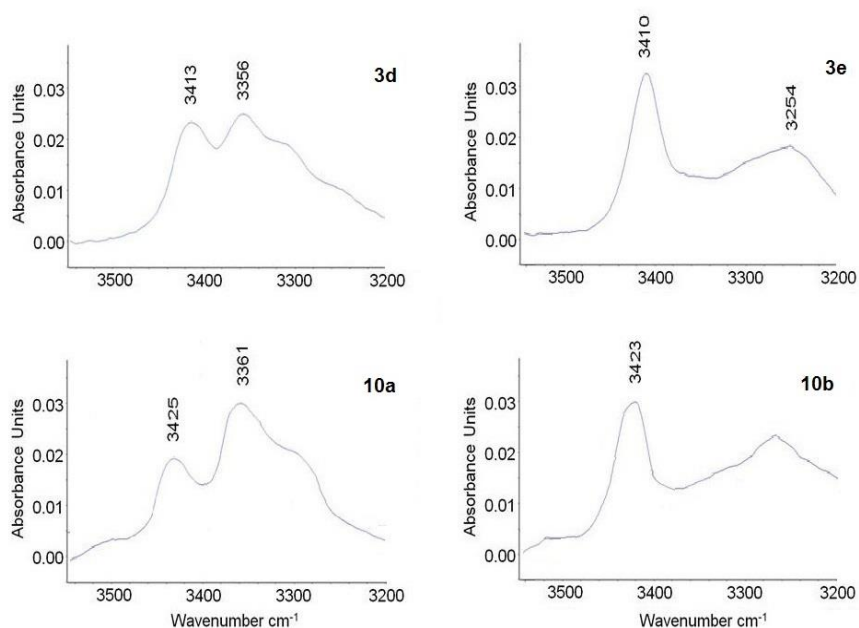


Figure S2. Amide NH stretching regions of the IR absorption spectra for samples of 2 mM **3d**, **3e**, **10a**, and **10b** in DCM at r.t.

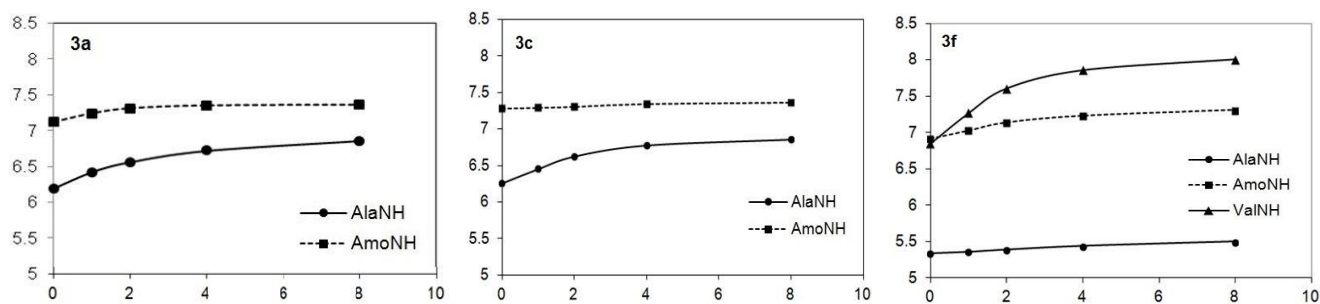


Figure S3. Titration experiments: variation of NH proton chemical shift (p.p.m.) of 2 mM **3a**, **3c**, **3f**, in $CDCl_3$ as a function of increasing $[D_6]DMSO$ (from 0 to 8 % v/v).

Table S2. Non-obvious ROESY cross-peaks observed for **3d** in 8:2 $[D_6]DMSO/H_2O$.^a

Cross-peak	intensity	Cross-peak	Intensity
ValNH-ValMe	vs	PheArH _{3,5} -AmoH β _{up}	m
ValNH-AlaMe	w	PheArH _{3,5} -PheH β _{up}	m
ValNH-ValH β	s	PheArH _{3,5} -PheH β _{dw}	w
ValNH-TsMe	w	PheArH _{3,5} -ValH α	w
ValNH-AmoH β _{up}	w	PheArH _{3,5} -PheH α	m
ValNH-PheH β _{up}	w	PheArH ₄ -ValMe	w
ValNH-PheH β _{dw}	s	PheArH ₄ -TsMe	w
ValNH-COOMe	w	PheArH _{2,6} -ValMe	w
ValNH-ValH α	m	PheArH _{2,6} -AlaMe	w
ValNH-PheH α	vs	PheArH _{2,6} -TsMe	w
ValNH-AlaNH	w	PheArH _{2,6} -AmoH β _{up}	m
ValNH-AmoNH	w	PheArH _{2,6} -PheH β _{up}	vs
AmoNH-ValMe	w	PheArH _{2,6} -PheH β _{dw}	s
AmoNH-AlaMe	m	PheArH _{2,6} -PheH α	vs
AmoNH-AmoH β _{up}	s	AmoH α -AlaMe	m
AmoNH-AmoH β _{dw}	m	AmoH α -AmoH β _{up}	m
AmoNH-AlaH α	vs	AmoH α -AmoH β _{dw}	vs
AmoNH-AmoH α	m	PheH α -ValMe	w
AmoNH-TsArH _{2,6}	m	PheH α -ValH β	w
AmoNH-AlaNH	m	PheH α -TsMe	w
AlaNH-ValMe	w	PheH α -PheH β _{up}	m
AlaNH-AlaMe	vs	PheH α -PheH β _{dw}	s
AlaNH-PheH β _{dw}	w	PheH α -COOMe	w
AlaNH-AlaH α	m	PheH α -ValH α	w
AlaNH-PheH α	w	ValH α -AlaMe	w
AlaNH-TsArH _{2,4}	w	ValH α -ValH β	vs
TsArH _{2,6} -ValMe	w	ValH α -TsMe	w
TsArH _{2,6} -AlaMe	s	ValH α -PheH β _{up}	w

TsArH _{2,6} -AmoHβ _{up}	w	ValHα-PheHβ _{dw}	w
TsArH _{2,6} -AmoHβ _{dw}	m	COOMe-ValHβ	m
TsArH _{2,6} -AlaHα	s	PheHβ _{dw} -AlaMe	w
TsArH _{2,6} -ValHα	w	PheHβ _{up} -ValMe	m
TsArH _{2,6} -AmoHα	w	PheHβ _{up} -ValHβ	w
TsArH _{3,5} -ValMe	w	AmoHβ _{dw} -AlaMe	w
TsArH _{3,5} -AlaMe	w	TsMe-ValMe	w
TsArH _{3,5} -AmoHβ _{dw}	w	TsMe-AlaMe	w
TsArH _{3,5} -Val Hα	w	TsMe-ValHβ	w
TsArH _{3,5} -Amo Hα	w	ValHβ-AlaMe	w
PheArH _{3,5} -AlaMe	w	ValMe-AlaMe	w

^a Stereochemistry has been omitted; ^b up = upfield, dw = downfield; ^c vs = very strong, s = strong, m = medium, w = weak.

Table S3. Non-obvious ROESY cross-peaks observed for **3e** in 8:2 [D₆]DMSO/H₂O.^a

Cross-peak	Intensity	Cross-peak	Intensity
ValNH-ValMe	vs	PheArH _{3,5} -PheHβ _{up}	w
ValNH-ValHβ	s	PheArH _{3,5} -PheHβ _{dw}	w
ValNH-AmoHα	w	PheArH _{3,5} -Phe Hα	w
ValNH-PheHβ _{up}	m	PheArH _{2,6} -AlaMe	w
ValNH-PheHβ _{dw}	w	PheArH _{2,6} -AmoHβ _{dw}	m
ValNH-COOMe	w	PheArH _{2,6} -PheHβ _{up}	s
ValNH-ValHα	m	PheArH _{2,6} -PheHβ _{dw}	vs
ValNH-PheHα	vs	PheArH _{2,6} -PheHα	vs
ValNH-AlaNH	w	AmoHα-ValMe	w
ValNH-AmoNH	w	AmoHα-AmoHβ _{up}	m
AmoNH-ValMe	w	AmoHα-AmoHβ _{dw}	m
AmoNH-AlaMe	m	AmoHα-PheHβ _{dw}	w
AmoNH-AmoHβ _{up}	m	AmoHα-ValHα	w
AmoNH-AmoHβ _{dw}	s	PheHα-ValMe	w
AmoNH-AlaHα	vs	PheHα-ValHβ	w
AmoNH-AmoHα	w	PheHα-PheHβ _{up}	s
AmoNH-AlaNH	w	PheHα-PheHβ _{dw}	m
AlaNH-ValMe	w	PheHα-ValHα	w
AlaNH-AlaMe	vs	ValHα-AlaMe	m
AlaNH-AmoHα	w	ValHα-ValHβ	m
AlaNH-AlaHα	s	ValHα-ValMe	vs

AlaNH-TsArH _{2,6}	s	ValH α -AmoH β _{dw}	w
AlaNH-TsArH _{3,5}	s	ValH α -PheH β _{up}	w
TsArH _{2,6} -ValMe	w	ValH α -PheH β _{dw}	w
TsArH _{2,6} -AlaMe	m	COOMe-ValMe	w
TsArH _{2,6} -AlaH α	m	COOMe-ValH β	w
TsArH _{2,6} -ValH α	w	COOMe-PheH β _{up}	s
TsArH _{2,6} -AmoH α	w	AlaH α -TsMe	w
TsArH _{2,6} -PheH α	w	TsMe-AlaMe	w
TsArH _{3,5} -ValMe	w	TsMe-ValH β	m
TsArH _{3,5} -AlaMe	w	ValH β -AlaMe	w
PheArH _{3,5} -AmoH β _{dw}	w	ValMe-AlaMe	w

^a Stereochemistry has been omitted; ^b up = upfield, dw = downfield; ^c vs = very strong, s = strong, m = medium, w = weak.

Table S4. Non-obvious ROESY cross-peaks observed for **10a** in 8:2 [D₆]DMSO/H₂O.^a

Cross-peak	intensity	Cross-peak	Intensity
ValNH-ValMe	vs	PheArH _{3,5} -AlaMe	w
ValNH-AlaMe	m	PheArH _{3,5} -AmoH β _{up}	m
ValNH-ValH β	s	PheArH _{3,5} -PheH β _{up}	m
ValNH-TsMe	w	PheArH _{3,5} -PheH β _{dw}	w
ValNH-AmoH β _{up}	w	PheArH _{3,5} -ValH α	w
ValNH-PheH β _{up}	w	PheArH _{3,5} -PheH α	m
ValNH-PheH β _{dw}	s	PheArH ₄ -ValMe	w
ValNH-COOMe	w	PheArH ₄ -TsMe	w
ValNH-ValH α	m	PheArH _{2,6} -ValMe	w
ValNH-PheH α	vs	PheArH _{2,6} -AlaMe	w
ValNH-AlaNH	w	PheArH _{2,6} -TsMe	w
ValNH-AmoNH	w	PheArH _{2,6} -AmoH β _{up}	m
AmoNH-ValMe	w	PheArH _{2,6} -PheH β _{up}	vs
AmoNH-AlaMe	m	PheArH _{2,6} -PheH β _{dw}	s
AmoNH-AmoMe	m	PheArH _{2,6} -PheH α	vs
AmoNH-ValH β	m	PheH α -ValMe	w
AmoNH-AmoH β _{up}	s	PheH α -ValH β	w
AmoNH-AmoH β _{dw}	m	PheH α -TsMe	w
AmoNH-AlaH α	vs	PheH α -PheH β _{up}	m
AmoNH-ValH α	m	PheH α -PheH β _{dw}	s
AmoNH-PheH α	m	PheH α -COOMe	w
AmoNH-TsArH _{2,6}	m	PheH α -ValH α	w
AmoNH-AlaNH	m	ValH α -AlaMe	m

AlaNH-ValMe	m	ValH α -ValH β	vs
AlaNH-AlaMe	vs	ValH α -TsMe	w
AlaNH-ValH β	w	ValH α -PheH β _{up}	w
AlaNH-PheH β _{dw}	w	ValH α -AlaH α	m
AlaNH-AlaH α	m	ValH α -COOMe	m
AlaNH-ValH α	m	ValH α -PheH β _{dw}	w
AlaNH-PheH α	w	COOMe-ValH β	m
AlaNH-TsArH _{2,4}	w	PheH β _{dw} -AlaMe	w
TsArH _{2,6} -ValMe	w	PheH β _{up} -ValMe	m
TsArH _{2,6} -AlaMe	s	PheH β _{up} -ValH β	w
TsArH _{2,6} -AmoMe	w	AmoH β _{dw} -AlaMe	w
TsArH _{2,6} -AmoH β _{up}	w	TsMe-ValMe	w
TsArH _{2,6} -AmoH β _{dw}	m	TsMe-AlaMe	w
TsArH _{2,6} -AlaH α	s	TsMe-ValH β	w
TsArH _{2,6} -ValH α	w	ValH β -AlaMe	m
TsArH _{3,5} -ValMe	w	AmoMe-AlaMe	w
TsArH _{3,5} -AlaMe	w	AmoMe-AmoH β _{up}	m
TsArH _{3,5} -AmoMe	w	AmoMe-AmoH β _{dw}	vs
TsArH _{3,5} -AmoH β _{dw}	w	AlaMe-ValMe	m
TsArH _{3,5} -ValH α	w		

^a Stereochemistry has been omitted; ^b up = upfield, dw = downfield; ^c vs = very strong, s = strong, m = medium, w = weak.

Table S5. Non-obvious ROESY cross-peaks observed for **10b** in 8:2 [D₆]DMSO/H₂O.^a

Cross-peak	Intensity	Cross-peak	Intensity
ValNH-ValMe	vs	PheArH _{3,5} -PheH β _{up}	w
ValNH-AmoMe	m	PheArH _{3,5} -PheH β _{dw}	w
ValNH-ValH β	s	PheArH _{3,5} -PheH α	w
ValNH-PheH β _{up}	m	PheArH _{2,6} -AlaMe	w
ValNH-PheH β _{dw}	w	PheArH _{2,6} -AmoH β _{dw}	m
ValNH-COOMe	w	PheArH _{2,6} -PheH β _{up}	s
ValNH-ValH α	m	PheArH _{2,6} -PheH β _{dw}	vs
ValNH-PheH α	vs	PheArH _{2,6} -PheH α	vs
ValNH-AlaNH	w	PheH α -ValMe	w
ValNH-AmoNH	w	PheH α -ValH β	w
AmoNH-ValMe	w	PheH α -PheH β _{up}	s
AmoNH-AlaMe	m	PheH α -PheH β _{dw}	m
AmoNH-AmoMe	m	PheH α -ValH α	w

AmoNH-AmoH β _{up}	m	ValH α -AlaMe	m
AmoNH-AmoH β _{dw}	s	ValH α -ValH β	m
AmoNH-AlaH α	vs	ValH α -ValMe	vs
AmoNH-AlaNH	w	ValH α -AmoH β _{dw}	w
AlaNH-ValMe	w	ValH α -PheH β _{up}	w
AlaNH-AlaMe	vs	ValH α -PheH β _{dw}	w
AlaNH-AmoMe	w	COOMe-ValMe	w
AlaNH-AlaH α	s	COOMe-ValH β	w
AlaNH-TsArH _{2,6}	s	COOMe-PheH β _{up}	s
AlaNH-TsArH _{3,5}	s	AlaH α -TsMe	w
TsArH _{2,6} -ValMe	w	TsMe-AlaMe	w
TsArH _{2,6} -AlaMe	m	TsMe-ValH β	m
TsArH _{2,6} -AmoMe	w	ValH β -AlaMe	w
TsArH _{2,6} -AlaH α	m	AmoMe-ValMe	m
TsArH _{2,6} -ValH α	w	AmoMe-AmoH β _{up}	m
TsArH _{2,6} -PheH α	w	AmoMe-AmoH β _{dw}	m
TsArH _{3,5} -ValMe	w	AmoMe-PheH β _{dw}	w
TsArH _{3,5} -AlaMe	w	AmoMe-ValH α	m
PheArH _{3,5} -AmoH β _{dw}	w	AlaMe-ValMe	m

^a Stereochemistry has been omitted; ^b up = upfield, dw = downfield; ^c vs = very strong, s = strong, m = medium, w = weak.

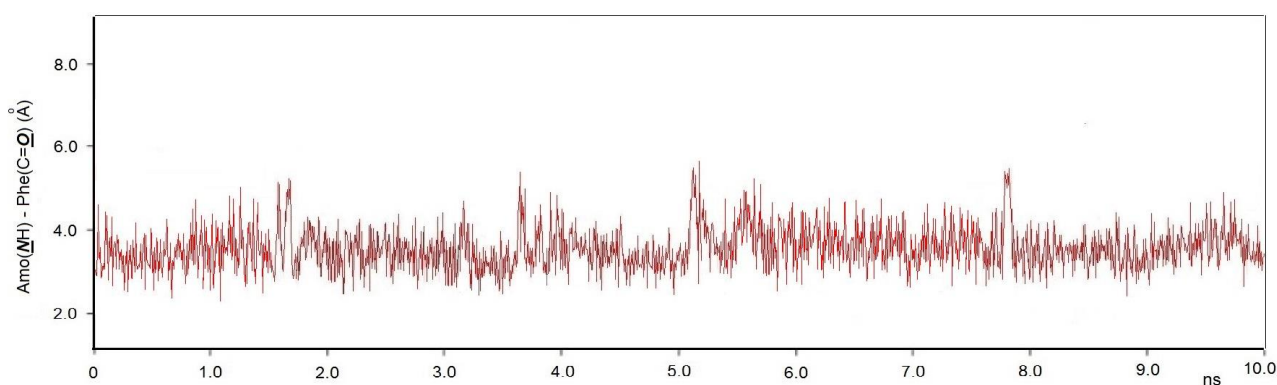


Figure S4. Distances (Å) between the amide nitrogen of Amo² and the carbonyl oxygen of Phe³ sampled from a 10 ns unrestrained Molecular Dynamics simulation of **10a** calculated in a 30x30x30 Å box of equilibrated standard TIP3P water molecules, using the ROESY-derived geometry as starting structure.

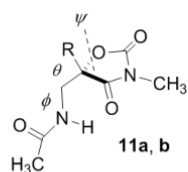


Figure 3. Minimum-energy conformations and relative energies of the *+g*, *trans*, and *-g* rotamers around the central backbone dihedral angle θ of model compounds **11a** and **11b**. A systematic conformational analysis around ϕ and θ was performed in gas-phase employing DFT; ΔE are given in kcal/mol; ϕ , θ , and ψ , are given into brackets in degrees; Amo is rendered in balls and cylinders, the rest in sticks.

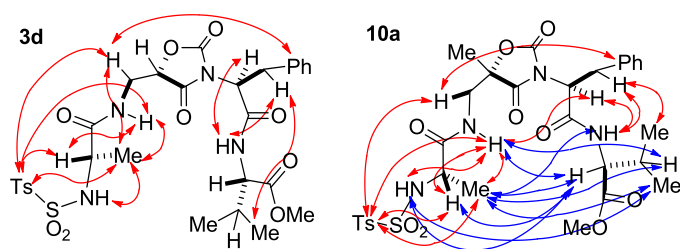
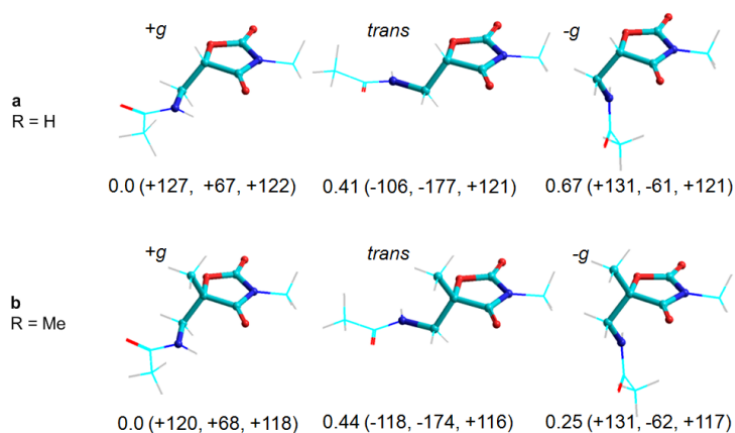


Figure 5. Sketches of the structures of **3d** and **10a**, and short-range (distances ≤ 3 Å) proton-proton ROESY correlations, indicated by arrows. Intra-residue and long-range (> 3 Å) correlations are not shown. Red arrows connect protons belonging to consecutive residues ($i-i+1$), while blue arrows connect protons of non-consecutive residues ($i-i+2$ or $i-i+3$).

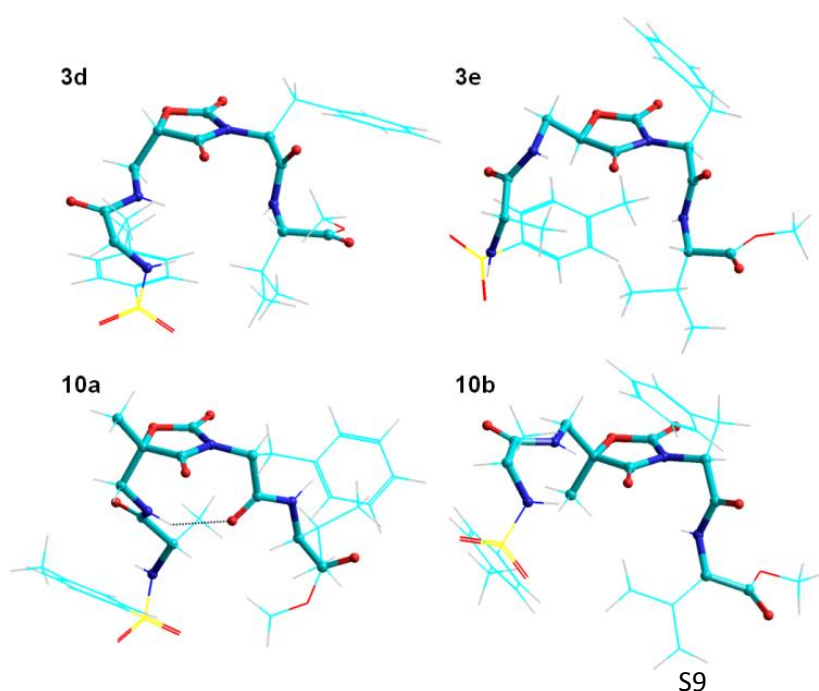


Figure 6. Top, representative lowest energy structures **3d**, **3e**, **10a**, and **10b**, calculated by restrained MD in a $30 \times 30 \times 30$ Å box of standard TIP3P water molecules. Backbones and Amo rings are rendered in balls and cylinders, the rest in sticks.

