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| Published<br>DOI:10.1104/pp.111.185710  |
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# Toward the Storage Metabolome: Profiling the Barley Vacuole<sup>1[W][OA]</sup>

#### Takayuki Tohge<sup>2</sup>, Magali Schnell Ramos<sup>2</sup>, Adriano Nunes-Nesi, Marek Mutwil, Patrick Giavalisco, Dirk Steinhauser, Maja Schellenberg, Lothar Willmitzer, Staffan Persson, Enrico Martinoia, and Alisdair R. Fernie<sup>\*</sup>

Max-Planck-Institute for Molecular Plant Physiology, 14476 Potsdam, Germany (T.T., A.N.-N., M.M., P.G., D.S., L.W., S.P., A.R.F.); Institute of Plant Biology, University of Zürich, 8008 Zurich, Switzerland (M.S.R., M.S., E.M.); Institut des Sciences du Végétal, CNRS, 91198 Gif-sur-Yvette, France (M.S.R.); and King Abdulaziz University, Jeddah 21589, Saudi Arabia (L.W.)

While recent years have witnessed dramatic advances in our capacity to identify and quantify an ever-increasing number of plant metabolites, our understanding of how metabolism is spatially regulated is still far from complete. In an attempt to partially address this question, we studied the storage metabolome of the barley (*Hordeum vulgare*) vacuole. For this purpose, we used highly purified vacuoles isolated by silicon oil centrifugation and compared their metabolome with that found in the mesophyll protoplast from which they were derived. Using a combination of gas chromatography-mass spectrometry and Fourier transform-mass spectrometry, we were able to detect 59 (primary) metabolites for which we know the exact chemical structure and a further 200 (secondary) metabolites for which we have strong predicted chemical formulae. Taken together, these metabolites comprise amino acids, organic acids, sugars, sugar alcohols, shikimate pathway intermediates, vitamins, phenylpropanoids, and flavonoids. Of the 259 putative metabolites, some 12 were found exclusively in the vacuole and 34 were found exclusively in the protoplast, while 213 were common in both samples. When analyzed on a quantitative basis, however, there is even more variance, with more than 60 of these compounds being present above the detection limit of our protocols. The combined data were also analyzed with respect to the tonoplast proteome in an attempt to infer specificities of the transporter proteins embedded in this membrane. Following comparison with recent observations made using nonaqueous fractionation of Arabidopsis *thaliana*), we discuss these data in the context of current models of metabolic compartmentation in plants.

Technological developments in the last decade have rendered metabolomics, the comprehensive study of small molecule metabolites, an important functional genomic tool (Fiehn, 2002; Sumner et al., 2003; Weckwerth, 2003; Farré et al., 2006; Matsuda et al., 2009; Saito and Matsuda, 2010). It is currently widely applied for a variety of purposes, ranging from simple diagnostic applications (Catchpole et al., 2005; Tikunov et al., 2005; Baker et al., 2006; Scherling et al., 2009) and gene functional annotation (Tohge et al., 2005, 2007) to being an important component in systems biology-oriented research (Fernie et al.,

<sup>[W]</sup> The online version of this article contains Web-only data.

2004; Saito and Matsuda, 2010). Metabolomics has additionally found great utility in the characterization of the metabolic response of plant cells to a range of biotic and environmental stresses (Cook et al., 2004; Hirai et al., 2004; Kaplan et al., 2004; Ishizaki et al., 2005; Urbanczyk-Wochniak and Fernie, 2005; Roessner et al., 2006; Yamaguchi-Shinozaki and Shinozaki, 2006; Lehmann et al., 2009). However, despite the many important breakthroughs facilitated by this technology, we remain somewhat hampered in our understanding by our inability to gain highresolution spatial data on metabolism (Stitt and Fernie, 2003; Fernie, 2007; Lunn, 2007).

Generally, metabolism is treated in a tissue-specific manner but cells within a tissue are treated as homogenous. Given that there are around 40 different cell types in plants (Goldberg, 1988; Martin et al., 2001), this is clearly an oversimplification. Recent technical advances and the widespread adoption of the GUS fusion protein expression approach (Sundaresan et al., 1995) as well as the increasing availability of resources such as Genevestigator and the BAR databases (Zimmermann et al., 2004; Brady and Provart, 2009), however, have made the identification of cell type specificity of gene expression facile. These developments have been paralleled by the increasing use of so-

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<sup>&</sup>lt;sup>1</sup> This work was supported by the Alexander von Humboldt Foundation (to T.T.).

<sup>&</sup>lt;sup>2</sup> These authors contributed equally to the article.

<sup>\*</sup> Corresponding author; e-mail fernie@mpimp-golm.mpg.de.

The author responsible for distribution of materials integral to the findings presented in this article in accordance with the policy described in the Instructions for Authors (www.plantphysiol.org) is: Alisdair R. Fernie (fernie@mpimp-golm.mpg.de).

<sup>&</sup>lt;sup>[OA]</sup> Open Access articles can be viewed online without a subscription.

www.plantphysiol.org/cgi/doi/10.1104/pp.111.185710

called single cell techniques, allowing microdissection of tissues into their constituent cell types (Schad et al., 2005; Cai and Lashbrook, 2006; Obel et al., 2009). These advances hold great promise for being able to better untangle metabolism at the cell type level, and a metabolite interface for the BAR gene expression atlas has indeed recently been published (Matsuda et al., 2009).

Going to even higher levels of resolution allows the analysis of subcellular and even suborganellar aspects of metabolism. Generally, the spatial distribution of metabolites mirrors that of the proteins involved in their metabolism. Given the large number of reports concerning plant organellar proteomes (Igamberdiev et al., 2001; Millar et al., 2001; Carter et al., 2004; Kleffmann et al., 2004; Pendle et al., 2005; Reumann et al., 2007) and the availability of the SUBA database of protein localization (Heazlewood et al., 2007), it should be relatively easy to predict the location and operation of metabolic pathways. However, the facts of organelle-specific isoforms and dual targeting of proteins often result in the duplication of metabolic reactions within different compartments of the cell (Carrari et al., 2003; Millar et al., 2006). Furthermore, several instances exist in which proteins are found in locations other than where they were thought to operate; for example, hexokinase proteins have been found in the nucleus, while the Suc synthase has been localized to the mitochondria (Subbaiah et al., 2006; Cho et al., 2007). These observations thus render such predictions of metabolite localization, on the basis of protein localization alone, liable to error. Moreover, for studies aimed at understanding the function of the vacuole, which is largely metabolically inert, such an association-based approach is likely to be highly difficult. At present, there are three different approaches for the direct determination of subcellular metabolite contents: the expression of genetically encoded marker proteins (Fehr et al., 2002), the nonaqueous fractionation technique (Gerhardt et al., 1983), and silicon oil centrifugation of protoplasts (Igamberdiev et al., 2001). The first of these requires the construction and expression of several variants of a marker protein for each individual metabolite and thus is not tractable at the metabolome level. The second approach has already been coupled to plant cells and proved to be highly informative in characterizing transgenic potato (Solanum tuberosum) plants exhibiting alterations in the Sucto-starch pathway (Farré et al., 2001, 2006). For the third approach, several proteome studies of the vacuole have been presented for Arabidopsis (Arabidopsis thaliana; Carter et al., 2004; Shimaoka et al., 2004; Jaquinod et al., 2007). Our study, however, is focused on barley (*Hordeum vulgare*), since it is our experience that the purity of vacuoles isolated from this species exceeds that from any other species. To this end, we isolated pure vacuoles from protoplasts derived from barley leaves and compared the metabolome of these organelles with that of the protoplast population from which they were derived using a combination of gas chromatography-mass spectrometry (GC-MS) and liquid chromatography-Fourier transform-mass spectrometry (LC-FT-MS). We were able to provide information concerning 61 metabolites of confirmed chemical structure and a further 135 putative metabolites on the basis of their chemical formulae and assess their partitioning between the vacuolar and extravacuolar parts of the cell. Finally, the data were compared with literature values obtained by the nonaqueous fractionation approach and in relation to tonoplast transporters described in the literature (Martinoia et al., 2000, 2007; Kaspar et al., 2010) as well as with data recently published concerning the vacuolar ionome (Smart et al., 2011).

#### RESULTS

#### Isolation of the Barley Vacuole from Protoplast

To investigate vacuole metabolomics, barley vacuoles were purified from barley protoplast. Barley mesophyll protoplasts were prepared using fresh barley leaves grown in a greenhouse for 8 d. Isolation and purification of mesophyll protoplasts and vacuoles were performed as described previously (Endler et al., 2009). In order to get rid of the isolation solution, an additional centrifugation step, as described in "Materials and Methods," was included. Protoplast samples were prepared as three replicates (P1– P3) and vacuole samples were prepared as six replicates (V1–V6) to evaluate biological and experimental variance. Purified vacuole fractions were assessed by  $\alpha$ -mannosidase enzymatic activity, which is known as vacuole-specific enzyme (Boller and Kende, 1979), and chlorophyll content for evaluation of the contamination of chloroplast. Four independent sets of vacuole preparations resulted in 70  $\pm$  29 milliunits mL<sup>-</sup> (vacuole fractions) and 22  $\pm$  4 milliunits mL<sup>-1</sup> (protoplast fractions) of  $\alpha$ -mannosidase enzymatic activity and no (vacuole fractions) and 293  $\pm$  52  $\mu$ g mL<sup>-</sup> (protoplast fractions) of chlorophyll content, respectively. The liquid samples were lyophilized and extracted by 80% methanol for preliminary check by GC-MS analyses. These preparations rendered high reproducibility of metabolite data between experiments. To obtain enough sample for split use in both MS-based analyses, the fractions derived from a single sample set were frozen overnight, and after defreezing, 45  $\mu$ L of the aqueous solution from each tube (50– 60 tubes per experiment) was recovered and pooled (vacuole fraction, 77 milliunits mL<sup>-1</sup> of  $\alpha$ -mannosidase enzymatic activity, no chlorophyll content; protoplast fraction, 23 milliunits mL<sup>-1</sup> and 272  $\mu$ g mL<sup>-1</sup>, respectively). The pooled fractions were lyophilized and used for the metabolomic analysis. The variance of dry weight between replicates was  $8.07 \pm 0.36$  mg (vacuole samples; n = 6) and 8.13  $\pm$  0.60 mg (protoplast; n = 3). For the metabolomic studies, lyophilized samples were extracted by 80% methanol for split use to both MS-based analyses.

#### Global Difference and Fraction-Specific Peaks

Metabolite profiles were analyzed by GC-time of flight-MS for primary metabolites (Lisec et al., 2006) and LC-MS for secondary metabolites (Tohge and Fernie, 2010). The resultant data were combined into a single data matrix for multivariate analyses. First, we performed principal component analysis to visualize the extent of metabolomic changes between protoplast samples and vacuole samples (Supplemental Fig. S1A). Since a clear separation between the protoplast samples and vacuole samples was observed in PC1 (74.9%), we subsequently performed a Venn diagram analysis to illustrate the levels of commonality and difference between samples (Supplemental Fig. S1A). In total, 259 peaks were detected by GC-MS and LC-MS. Twelve peaks detected by LC-MS were exclusively detected in the vacuole samples (Supplemental Table S1). On the other hand, 34 peaks were observed as protoplast-specific peaks, meaning that 213 peaks were detected in both fractions. Given that the subcellular vacuole volume of barley leaves has been defined to be 60.5% of the total cell volume (Winter et al., 1993), we defined those compounds that were abundant at percentages above 60% in vacuole as "vacuolar accumulated." Ninety-nine peaks were detected at high relative abundance (more than 1.5-fold change [FC]) in the vacuole samples, while 62 peaks were detected at low relative abundance (less than 0.5 FC). It is important to bear in mind here that vacuole-specific peaks could, in principle, also be detected in protoplast samples. For those that were solely detected in the vacuole, it thus seems likely that their levels were under the limit of detection in our protoplast samples. By contrast, protoplast-specific peaks suggest that these metabolites are not transported into the vacuole.

#### Metabolite Profiling of Primary Metabolites by GC-MS

As stated above, measurements of the relative levels of primary metabolites were performed by GC-MS (Lisec et al., 2006), and the absolute levels of 59 compounds were estimated based on concentration curves of authentic standard that were run side by side with the samples (Table I). The results show that the vacuole accumulates a large variety of different compounds: sugars, sugar alcohols, organic acids, and amino acids. A total of 22 amino acids and their related compounds were determined in vacuolar and protoplast samples. Three of them had higher vacuolar than protoplast levels (more than 90% of compound in the vacuole, more than 1.5 FC), seven were invariant (30%–90%, 0.5-1.5 FC), and seven displayed a lower vacuolar level (less than 30%, less than 0.5 FC; Table I; Fig. 1). Interestingly, His and Ala were highly more abundant in the barley vacuole than in the protoplast, followed by Trp (1.53 FC), Met (1.43 FC), and Ser (1.24 FC). The relative amounts of Glu (less than 0.02 FC), Pro (0.09 FC), Gln (0.10 FC), and Asp (0.12 FC) were very low inside the vacuole. The levels of other amino acids, such as Orn, Tyr, Ile, tyramine, Val, Phe, pyro-Glu, and Gly, were invariant. Additionally, we were also able to determine  $\gamma$ -aminobutyric acid amounts in both samples, but at higher levels in the protoplast samples (Table I; Fig. 1).

Several of the organic acids measured had higher vacuolar than protoplast levels (Table I). Higher levels of the tricarboxylic acid (TCA) cycle intermediates isocitrate, malate, succinate, and gluconate were observed in vacuoles. The levels of aconitate, glycerate, 2-oxoglutarate, galacturonate, shikimate, maleate, glutarate, pyruvate, benzoate, fumarate, aconitate, and threonate, however, were lower in the vacuole samples (Table I; Fig. 1).

We were additionally able to detect several sugars in the vacuole samples (Table I; Fig. 1). Monosaccharides such as Ara exhibited higher levels in vacuolar than protoplast samples. Interestingly, the level of Suc was less abundant in vacuole samples. Several other sugars, namely Glc, Xyl, Fru, Gal, and maltose, were also found at abundant levels within the vacuole.

Additional nitrogen-containing metabolites were also detected in the vacuole samples, including putrescine (Table I; Fig. 1). The purine derivative adenine displayed higher levels in protoplast samples, whereas putrescine and urea levels were invariant between sample types. Among the other compounds analyzed, it is interesting that the levels of intermediates involved in membrane biogenesis, such as C16:0 and C18:0 fatty acids, were lower in vacuole samples (Table II).

#### Metabolite Profiling of Secondary Metabolites by LC-MS

Next, ultra-performance liquid chromatography (UPLC)-FT-MS analysis was performed for the determination of secondary metabolite contents in the different samples. Detected peaks were identified and annotated by a combined usage of standard compounds, database surveys on the basis of accurate mass-to-charge ratio (m/z) values as analyzed by FT-MS (Giavalisco et al., 2009; Horai et al., 2010), and comparison between coelution analysis with wellanalyzed plant extracts such as Arabidopsis (Tohge et al., 2005; Nakabayashi et al., 2010) and tomato (Solanum lycopersicum) fruits (Moco et al., 2006; Iijima et al., 2008). By UPLC-FT-MS analysis, 200 peaks were detected in total. Twelve of these peaks were undetected in the protoplast samples, whereas 34 peaks were undetected in the vacuolar samples (Supplemental Table S1). Both Phe and Trp displayed similar relative levels as analyzed by GC-MS. Of 12 vacuolespecific peaks, four peaks were annotated as glycoflavone derivatives (Table II). The major flavonoid isovitexin (Kaspar et al., 2010), which was identified by standard compound confirmation, was also highly detected in vacuolar samples. In addition, analysis of the other six glycoflavone derivatives annotated by UPLC-FT-MS revealed that all putative glycoflavone derivatives were more highly abundant in the vacuo-

#### Table I. Metabolite contents of barley protoplasts and vacuole samples

Values are presented as means  $\pm$  sp of determinations on six replicates of vacuole fractions and three protoplast fractions. Percentage of content levels in vacuole was calculated by  $\alpha$ -mannosidase activity. Given that the subcellular vacuole volume of barley leaves has been defined to be 60.5% of the total cell volume (Winter et al., 1993), we defined those compounds that were abundant at percentages above 60% in vacuole as vacuolar accumulated.

| Compound Name               | FC     | In Vacuole | Protoplast                       | Vacuole                           |
|-----------------------------|--------|------------|----------------------------------|-----------------------------------|
|                             |        | %          | nmol $L^{-1}$ protoplast         |                                   |
| Amino acids and derivatives |        |            |                                  |                                   |
| His                         | >1.65  | >100       | $5.6 \pm 1.7$                    | $21.9 \pm 9.8$                    |
| Ala                         | >1.65  | >100       | $1,981.6 \pm 127.6$              | $4,789.2 \pm 1,415.7$             |
| Trp                         | 1.53   | 92.4       | $47.5 \pm 3.9$                   | $43.9 \pm 7.6$                    |
| Met                         | 1.43   | 86.7       | $22.4 \pm 2.6$                   | $19.4 \pm 2.1$                    |
| Ser                         | 1.24   | 75.2       | $4,190.2 \pm 392.4$              | $3,150.2 \pm 1069.6$              |
| Orn                         | 0.79   | 47.6       | $15.9 \pm 0.6$                   | $7.6 \pm 1.6$                     |
| Tyr                         | 0.78   | 47.4       | $1,612.4 \pm 46.5$               | $765.0 \pm 92.7$                  |
| lle                         | 0.78   | 46.9       | $1,848.5 \pm 111.8$              | 867.5 ± 150.4                     |
| Tyramine                    | 0.65   | 39.4       | $12.8 \pm 1.4$                   | $5.0 \pm 0.8$                     |
| Val                         | 0.56   | 33.6       | $4,063.4 \pm 442.1$              | $1,364.0 \pm 304.9$               |
| Phe                         | 0.55   | 33.1       | $734.0 \pm 29.0$                 | $242.9 \pm 38.8$                  |
| Pyro-Glu                    | 0.51   | 30.6       | $1,583.2 \pm 160.2$              | $484.6 \pm 63.5$                  |
| Gly                         | 0.43   | 26.2       | $1,949.2 \pm 258.2$              | $404.0 \pm 03.3$<br>509.8 ± 131.9 |
|                             |        |            |                                  |                                   |
| Asn                         | 0.35   | 21.4       | $980.2 \pm 176.1$                | $209.4 \pm 103.4$                 |
| Thr                         | 0.33   | 20.2       | $4,169.7 \pm 233.1$              | $842.4 \pm 99.1$                  |
| Lys                         | 0.31   | 18.9       | $1,028.3 \pm 54.6$               | $193.9 \pm 42.7$                  |
| β-Ala                       | 0.31   | 18.6       | $143.6 \pm 10.0$                 | $26.6 \pm 2.4$                    |
| Asp                         | 0.12   | 7.5        | $1,358.7 \pm 882.7$              | $102.5 \pm 17.9$                  |
| Gln                         | 0.10   | 5.8        | $990.8 \pm 97.8$                 | $57.1 \pm 9.5$                    |
| Pro                         | 0.09   | 5.2        | $62.2 \pm 11.3$                  | $3.3 \pm 0.8$                     |
| γ-Aminobutyric acid         | < 0.02 | <1         | $1,3381.1 \pm 784.5$             | $101.4 \pm 7.9$                   |
| Glu                         | < 0.02 | <1         | $98.8 \pm 15.8$                  | $0.4 \pm 0.1$                     |
| Organic acids               |        |            |                                  |                                   |
| Isocitric acid              | >1.65  | >100       | $114.0 \pm 1.9$                  | $1,080.0 \pm 900.3$               |
| Malic acid                  | >1.65  | >100       | $2,974.1 \pm 78.5$               | 6,352.5 ± 834.2                   |
| Succinic acid               | >1.65  | >100       | $3,229.1 \pm 126.5$              | $3,709.9 \pm 1,126.6$             |
| Gluconic acid               | 1.52   | 92.2       | $2,911.7 \pm 388.1$              | $2,685.9 \pm 534.6$               |
| Citric acid                 | 0.56   | 33.9       | $126.1 \pm 16.6$                 | $42.7 \pm 7.3$                    |
| Threonic acid               | 0.46   | 27.8       | $20.5 \pm 2.4$                   | $5.7 \pm 1.3$                     |
| Aconitic acid               | 0.44   | 26.6       | $9.5 \pm 1.6$                    | $3.7 \pm 1.3$<br>$2.5 \pm 0.6$    |
|                             | 0.40   | 23.9       |                                  |                                   |
| Fumaric acid                |        |            | $179.4 \pm 35.4$                 | $42.9 \pm 12.2$                   |
| Benzoic acid                | 0.39   | 23.6       | $61.5 \pm 14.2$                  | $14.5 \pm 4.9$                    |
| Pyruvic acid                | 0.39   | 23.3       | $23.5 \pm 3.0$                   | $5.5 \pm 0.9$                     |
| Glutaric acid               | 0.31   | 18.9       | $0.5 \pm 0.2$                    | $0.1 \pm 0.1$                     |
| Maleic acid                 | 0.23   | 14.0       | $65.4 \pm 17.3$                  | $9.1 \pm 2.4$                     |
| Shikimic acid               | 0.21   | 12.9       | $2.2 \pm 0.3$                    | $0.3 \pm 0.1$                     |
| GalUA                       | 0.12   | 7.3        | $351.5 \pm 54.6$                 | $25.6 \pm 7.2$                    |
| 2-Oxoglutarate              | 0.11   | 6.9        | $3.3 \pm 1.1$                    | $0.2 \pm 0.1$                     |
| Glyceric acid               | 0.04   | 2.7        | $544.7 \pm 58.8$                 | $14.8 \pm 1.6$                    |
| Sugars                      |        |            |                                  |                                   |
| Āra                         | 1.44   | 87.2       | $7.9 \pm 0.9$                    | $6.9 \pm 1.2$                     |
| Glc                         | 1.01   | 61.4       | $385.1 \pm 46.7$                 | $236.4 \pm 41.2$                  |
| Xyl                         | 0.94   | 57.0       | $6.4 \pm 0.8$                    | $3.6 \pm 0.6$                     |
| Fru                         | 0.84   | 51.0       | $28.3 \pm 3.4$                   | $14.4 \pm 2.7$                    |
| Gal                         | 0.65   | 39.1       | $4.7 \pm 1.7$                    | $1.9 \pm 0.9$                     |
| Maltose                     | 0.50   | 30.5       | $4.7 \pm 1.7$<br>13.9 ± 1.8      | $4.2 \pm 0.6$                     |
| Gentiobiose                 | 0.43   | 26.2       | $13.9 \pm 1.0$<br>$11.9 \pm 2.9$ | $4.2 \pm 0.6$<br>$3.1 \pm 0.5$    |
|                             |        |            |                                  |                                   |
| Fuc                         | 0.42   | 25.5       | $15.9 \pm 2.2$                   | $4.0 \pm 0.8$                     |
| Man                         | 0.39   | 23.3       | $1.4 \pm 0.3$                    | $0.3 \pm 0.0$                     |
| Isomaltose                  | 0.38   | 23.1       | $3.9 \pm 0.4$                    | $0.9 \pm 0.2$                     |
| Rib                         | 0.37   | 22.3       | $6.0 \pm 0.9$                    | $1.3 \pm 0.2$                     |
| Suc                         | 0.13   | 7.8        | $54.9 \pm 6.2$                   | $4.3 \pm 0.7$                     |
| Fatty acids                 |        |            |                                  |                                   |
| 18:00                       | 0.57   | 34.7       | $316.9 \pm 81.4$                 | $110.1 \pm 43.0$                  |
| 16:00                       | 0.53   | 32.0       | $305.3 \pm 82.7$                 | $97.6 \pm 40.6$                   |
|                             |        |            | (Table co                        | ntinues on following page.)       |

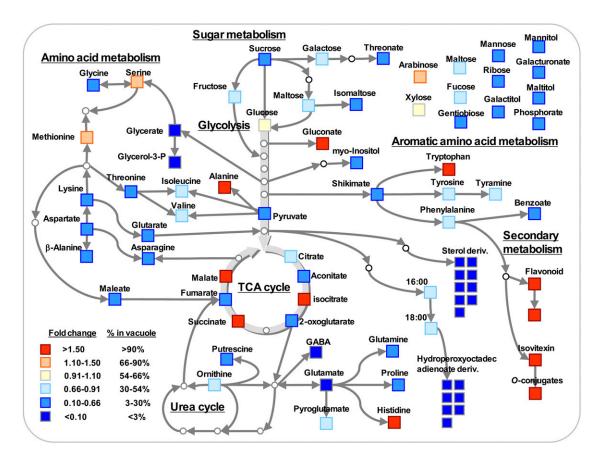
(Table continues on following page.)

| Compound Name        | FC   | In Vacuole | Protoplast          | Vacuole        |
|----------------------|------|------------|---------------------|----------------|
| Others               |      |            |                     |                |
| Mannitol             | 0.48 | 29.0       | $106.8 \pm 21.4$    | $30.9 \pm 6.5$ |
| Phosphoric acid      | 0.48 | 28.8       | $1,687.5 \pm 283.4$ | 485.8 ± 170.4  |
| Sorbitol/galactitol  | 0.44 | 26.9       | $1,698.3 \pm 443.6$ | 456.2 ± 174.9  |
| Maltitol             | 0.44 | 26.9       | $1.4 \pm 0.5$       | $0.4 \pm 0.2$  |
| Putrescine           | 0.18 | 11.0       | $20.5 \pm 3.1$      | $2.3 \pm 0.5$  |
| Myoinositol          | 0.08 | 4.7        | $466.7 \pm 18.3$    | $21.8 \pm 2.8$ |
| Glycerol-3-phosphate | 0.04 | 2.2        | $35.6 \pm 2.7$      | $0.8 \pm 0.1$  |

lar than the protoplast samples. Sterol derivatives represent one of the most important (and abundant) classes of secondary metabolite in barley (Bush et al., 1971; Hübke et al., 2005). In total, nine peaks were annotated as sterol derivatives, including both sitosterol and cholesterol derivatives (Table II). These putative sterol derivatives (Bush et al., 1971; Konig and Seifert, 1998), however, were not detected in vacuolar samples. In addition, seven putative hydroperoxyoctadecadienoic acid (HODE)-related peaks (Hübke et al., 2005) were also only detected in protoplast samples. Indeed, sterol derivatives and HODE were not even detected in the vacuolar samples in their glycosidic forms.

#### Integration of Metabolomic Data Sets with a Subcellarly Targeted Coexpression Analysis

In an attempt to predict the underlying cytosolicvacuolar transport system, we next evaluated tonoplast proteins predicted to have a transport function within the context of the data collected here. This integration analysis was performed by comparing the data obtained here with published data. For this



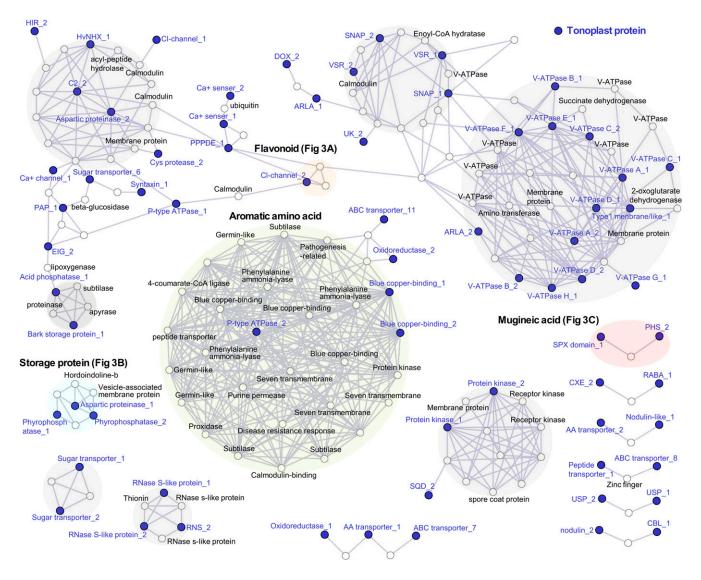
**Figure 1.** Ratio of metabolite abundance in vacuole to protoplast represented on metabolic pathways. The metabolic level in the vacuole fraction was calculated against the level in the protoplast fraction. Values used for both sample types were normalized by  $\alpha$ -mannosidase activity. The intensity of differential abundance is illustrated by a false-color scale as described in the key. GABA,  $\gamma$ -Aminobutyric acid.

| Table II. Comparison of the content of secondary metabolites detected and annotated by UPLC-FT-MS between protoplasts and purified vacuole |
|--|
| fractions  |

| nctions        |          |            |                      |        | ,          |   |
|----------------|----------|------------|----------------------|--------|------------|---|
| Retention Time | m/z      | Formula    | Molecular<br>Weight  | FC     | In Vacuole | Peak<br>Annotation                      |
| min            |          |            |                      |        | %          |   |
| Rt5.12         | 609.1455 | C27H30O16  | 610.1534             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt9.62         | 631.2768 | C33H38N4O6 | 586.2791             | >1.65  | High in V  | Putative tetrapyrrole related           |
| Rt9.77         | 631.2768 | C33H38N4O6 | 586.2791             | >1.65  | High in V  | Putative tetrapyrrole related           |
| Rt5.12         | 655.1513 | C28H32O18  | 656.1589             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt7.22         | 665.2819 | C32H44O12  | 620.2833             | >1.65  | High in V  | Putative phenylpropanoid                |
| Rt4.59         | 759.2351 | C32H42O18  | 714.2371             | >1.65  | High in V  | Putative phenylpropanoid                |
| Rt6.71         | 799.2083 | C38H40O19  | 800.2164             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt6.27         | 815.2038 | C38H40O20  | 816.2113             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt7.36         | 431.0979 | C21H20O10  | 432.1057             | >1.65  | High in V  | Flavonoid:apigenin-6-C-glucoside        |
| Rt4.07         | 693.2353 | C40H38O11  | 694.2434             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt7.52         | 475.1814 | C21H32O12  | 476.1894             | >1.65  | High in V  | Putative phenyl diglycoside             |
| Rt6.78         | 875.2251 | C40H44O22  | 876.2324             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt8.43         | 455.1916 | C21H30O8   | 410.1941             | >1.65  | High in V  | Putative sesquiterpene glycoside        |
| Rt7.97         | 475.2113 | C29H32O6   | 476.2199             | >1.65  | High in V  | Putative sesquiterpene glycoside        |
| Rt3.96         | 407.1719 | C21H28O8   | 408.1784             | >1.65  | High in V  | Putative sesquiterpene glycoside        |
| Rt6.19         | 523.1659 | C21H32O15  | 524.1741             | >1.65  | High in V  | Putative iridoide                       |
| Rt11.14        | 637.3587 | C33H52O9   | 592.3611             | >1.65  | High in V  | Putative triterpene derivatives related |
| Rt11.88        | 649.3587 | C34H52O9   | 604.3611             | >1.65  | High in V  | Putative triterpene derivatives related |
| Rt6.63         | 523.1658 | C21H32O15  | 524.1741             | >1.65  | High in V  | Putative iridoide                       |
| Rt7.48         | 631.2237 | C27H38O14  | 586.2262             | >1.65  | High in V  | Putative iridoide                       |
| Rt5.31         | 459.1501 | C20H28O12  | 460.1581             | >1.65  | High in V  | Putative phenylpropanoid                |
| Rt5.78         | 507.2073 | C21H34O11  | 462.2101             | >1.65  | High in V  | Putative iridoide                       |
| Rt7.05         | 845.2144 | C39H42O21  | 846.2219             | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt6.55         | 222.0402 | C10H9NO5   | 223.0481             | >1.65  | High in V  | Putative Trp derivative                 |
| Rt7.77         | 485.1658 | C21H28O10  | 440.1683             | >1.65  | High in V  | Putative sesquiterpene glycoside        |
| Rt7.10         | 609.1454 | C27H30O16  |                      | >1.65  | High in V  | Putative flavonoid glycoside            |
| Rt6.95         | 769.1975 | C37H38O18  | 610.1534<br>770.2058 | 1.59   | 95.6       |   |
| Rt5.88         |          |            |                      | 1.59   | 92.6       | Putative flavonoid glycoside            |
|                | 669.1665 | C29H34O18  | 670.1745             |        |            | Putative flavonoid glycoside            |
| Rt3.96         | 249.0875 | C11H12N2O2 | 204.0899             | 1.24   | 74.7       | Trp                                     |
| Rt6.36         | 307.1394 | C12H22O6   | 262.1416             | 1.04   | 62.7       | Putative fatty acid glycoside           |
| Rt2.97         | 210.0767 | C9H11NO2   | 165.0790             | 0.59   | 35.8       | Phe                                     |
| Rt5.66         | 593.1504 | C27H30O15  | 594.1585             | 0.59   | 35.8       | Sapnarin, flavonoid                     |
| Rt3.59         | 259.1293 | C10H18N2O3 | 214.1317             | < 0.02 | <1.0       | Putative dethiobiotin related           |
| Rt2.69         | 283.0678 | C10H12N4O6 | 284.0757             | < 0.02 | No in V    | Putative xanthosine                     |
| Rt13.50        | 293.2116 | C18H30O3   | 294.2195             | < 0.02 | No in V    | Cyclopentaneoctanoic acid related       |
| Rt13.74        | 293.2116 | C18H30O3   | 294.2195             | < 0.02 | No in V    | Cyclopentaneoctanoic acid related       |
| Rt13.81        | 293.2116 | C18H30O3   | 294.2195             | < 0.02 | No in V    | Cyclopentaneoctanoic acid related       |
| Rt12.97        | 311.2222 | C18H32O4   | 312.2301             | < 0.02 | No in V    | HODE related                            |
| Rt13.48        | 311.2222 | C18H32O4   | 312.2301             | < 0.02 | No in V    | HODE related                            |
| Rt11.30        | 327.2171 | C18H32O5   | 328.2250             | < 0.02 | No in V    | HODE related                            |
| Rt11.64        | 327.2171 | C18H32O5   | 328.2250             | < 0.02 | No in V    | HODE related                            |
| Rt11.94        | 327.2171 | C18H32O5   | 328.2250             | < 0.02 | No in V    | HODE related                            |
| Rt10.86        | 327.2172 | C18H32O5   | 328.2250             | < 0.02 | No in V    | HODE related                            |
| Rt12.18        | 329.2328 | C18H34O5   | 330.2406             | < 0.02 | No in V    | HODE related                            |
| Rt15.37        | 477.3215 | C28H46O6   | 478.3294             | < 0.02 | No in V    | Sterol glycoside                        |
| Rt15.53        | 491.3372 | C29H48O6   | 492.3451             | < 0.02 | No in V    | Sterol glycoside                        |
| Rt12.81        | 619.4210 | C35H58O6   | 574.4233             | < 0.02 | High in V  | Putative flavonoid glycoside            |
| Rt13.09        | 619.4211 | C35H58O6   | 574.4233             | < 0.02 | High in V  | Putative tetrapyrrole related           |
| Rt12.81        | 621.4371 | C36H62O8   | 622.4445             | < 0.02 | High in V  | Putative tetrapyrrole related           |

purpose, we evaluated 88 proteins reported to be tonoplast proteins of barley (Endler et al., 2006). As an initial step, we bioinformatically evaluated the possibility that individuals within either set of proteins represented transporter proteins. In order to interpret these combined data sets, an integrative analysis was performed utilizing coexpression analysis of publicly available microarray data. In order to convert GenBank identifiers to those of the Affymetrix probe, an amino acid BLAST search was performed using PLEXdb BLAST (http://www.plexdb. org/modules/tools/plexdb\_blast.php). In this process, the 88 proteins were converted to 128 probe sets (since the numbers are not identical, comments on cross-hybridization are given in the list in Supplemental Table S1).

To classify gene expression clusters, we performed coexpression analyses with the 128 probe sets using PlaNet for Barley (Mutwil et al., 2011; http://aranet. mpimp-golm.mpg.de/barnet) to facilitate this process (Fig. 2). The framework of the coexpression network was constructed by the connections between vacuolar probe sets using the PlaNet database. Coexpressed network genes encoding tonoplast proteins were separated into 13 subgroups (Fig. 2; Supplemental Fig. S2), which were constructed by 70 vacuolar probe sets and 112 genes (Supplemental Table S2). The dense cluster group (middle group in Fig. 2) is well correlated with aromatic amino acid-related genes such as Phe ammonia lyase and generally expressed in photosynthetic tissues (PlaNet barley expression profiling; http://aranet.mpimp-golm.mpg.de/barnet). By contrast, a second dense cluster (top right group in Fig. 2) showed a cluster of vacuolar ATP synthase proteins



**Figure 2.** Frameworks of the coexpression network of barley vacuolar probe sets. The quality-checked coexpression network for barley was obtained from the PlaNet server (http://aranet.mpimp-golm.mpg.de/download/). Genes (probe sets) are depicted as nodes, while significant coexpression similarity between any two probe sets is represented by an edge. Probe sets corresponding to tonoplast-located proteins (depicted by blue nodes) were used as queries for searching the coexpression networks (Supplemental Table S2). The analysis extracted all probe sets connected to at least two vacuolar probe sets (i.e. strongly associated with vacuolar processes). Annotation of the probe sets and PageMan analysis of ontology terms can be found in Supplemental Table S2. ARLA, ADP-ribosylation factor; CBL, calcineurin B; CXE, carboxyesterase; DOX, lipoxygenase; EIG, elicitor-inducible protein; GST, glutathione *S*-transferase; PAP, purple acid phosphatase; PHS, propyzamide-hypersensitive protein; RNS, RNase; SNAP, soluble *N*-ethylmaleimide-sensitive factor attachment protein; SQD, sulfoquinovosyldiacylglycerol glycosyltransferase; UK, unknown protein; USP, universal stress protein; V-ATPase, vacuolar ATP synthase; VSR, vacuolar targeting receptor.

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and TCA cycle-related genes, such as 2-oxoglutarate dehydrogenase and succinate dehydrogenase.

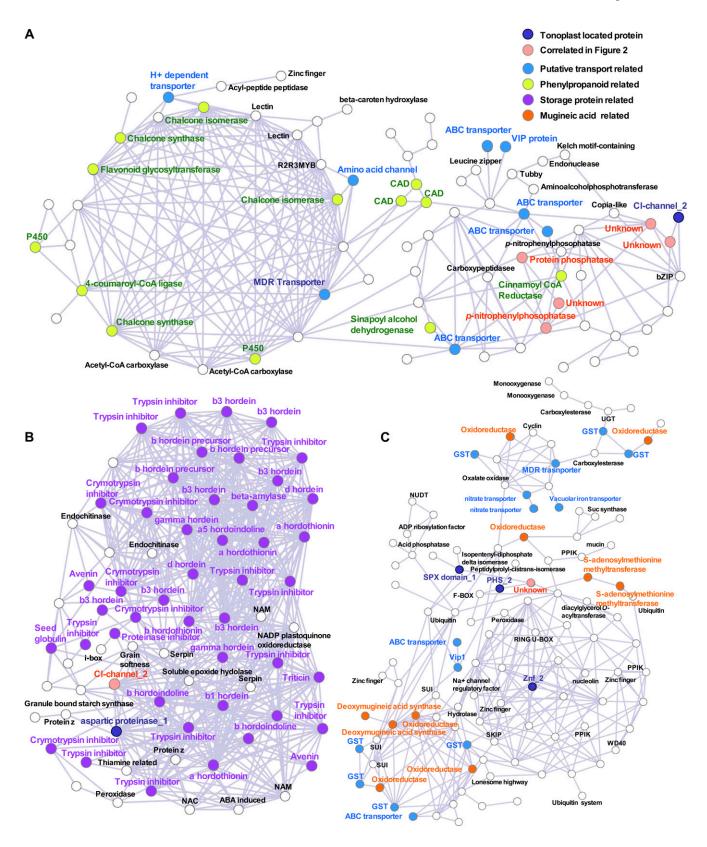
As a next step, we evaluated the barley coexpressed genes in each cluster group independently to facilitate the prediction of related metabolic pathways. In order to analyze these networks in more detail, some subclusters were focused using the PlaNet database. A gene of interest was queried to search a specific subcluster classified by PlaNet based on the framework described in Figure 2. We were able to identify three examples (Fig. 3). First, we focused on a subcluster (orange in Fig. 2) between groups, because this is a hub cluster to connect to the vacuolar ATP synthase cluster to calmodulin clusters (Fig. 2). Figure 3A shows a correlation network of a gene encoding a chloride channel protein (Contig6783\_at) and unknown proteins (Contig3508\_at and Contig3508\_s\_at). This network revealed a network connection to stressinducible secondary metabolites such as flavonoid, because chalcone synthase (Contig7356\_at and Contig7358\_at), chalcone isomerase (Contig9047\_at and Contig9048\_ at), 4-coumaroyl-CoA ligase (Contig4677\_at), P450 (HVSCEb0006P14r2\_at and HVSCEb0006O20r2\_at), and flavonoid glycosyltransferase (Contig11602\_at) were observed in this cluster. In this network, analysis showed the candidates for transporter genes that are putatively related to secondary metabolism, such as multidrugresistant-type transporter (HVSMEn0015M0015Mo1r2\_ s\_at) and  $H^+$ -dependent transporter (Contig26311\_at). Other examples indicate a subcluster that showed a small dense network with many genes encoding storage proteins (Fig. 3B). The network in Figure 3B was correlated with many storage protein genes. In this network, many genes encoding storage proteins and related proteins, such as  $\alpha$ - $\beta$ - $\gamma$ -hordein, hordoindoline, prolamine, avenin, and a trypsin inhibitor, were observed. These proteins were predicted to be transporters for storage protein, as mentioned above, and were highly expressed during seed development and correlated with the genes hordein and avenin. The tonoplast-located proteins aspartic protease (Contig3144\_at) and Cl<sup>-</sup> channel protein (Contig7078\_at) were strongly connected to the storage protein cluster. Again, we next performed gene expression tree analysis using only the small cluster that was characterized by "contains similarity to a tetracycline resistance efflux protein" (Contig13913\_at) and putative protein phosphatase (Contig9964\_at). The network showed a correlation network with mugineic acid biosynthesis, such as deoxymugineic acid synthase (Contig3519\_ at and HVSMEa0014J15r2\_x\_at) and oxidoreductase (Contig6722 at, Contig9968 at, Contig9968 at, Contig10877 at, Contig11890\_at, and HV\_CEa0006E02r2\_s\_at; Fig. 3C). In addition, this network also showed strong connection with many genes encoding transport-related protein, such as ATP-binding cassette (ABC) transporter (Contig2625\_at and Contig7226\_at) and glutathione S-transferase (Contig4137\_at, Contig9632\_at, HA03B19u\_ s\_at, and HVSMEa0014H14r2\_s\_at), because mugineic acid is a phytosiderophore and a ligand compound that is a metal chelator in graminaceous plants.

#### DISCUSSION

The process of homeostasis requires that the levels of certain metabolite pools be maintained at nearconstant levels. One mechanism to achieve this is the sequestration of metabolites into a relatively metabolically inert environment such as the vacuole. Our current knowledge of the vacuole comes largely from studies of isolated vacuoles as well as cell biological studies of exocytosis, endocytosis, and autophagy (Phan et al., 2008; Liu et al., 2009). These approaches have revealed a wide range of metabolites including sugars, organic acids, amino acids as well as chlorophyll catabolites, glutathione conjugates, bile acids, and sulfate conjugates (Martinoia et al., 2000; Farré et al., 2001); however, until recently, this was rather fragmentary. The application of the nonaqueous fractionation method in potato tubers when coupled to GC-MS analysis represented a first attempt to address this more systematically (Farré et al., 2001). This has been further refined by improvements in the resolution of the nonaqueous technique due to the ability to assess ever-smaller sample volumes in addition to the additional use of FT-ion cyclotron resonance (ICR)-MS, which has a far greater mass accuracy and thus potential for compound prediction. This combined approach was recently applied to Arabidopsis (Krueger et al., 2011) and produced an unprecedented amount of information concerning spatial resolution. Here, we utilized both GC-MS and FT-ICR-MS but instead applied them to silicon oil-centrifuged vacuoles from protoplast preparations of barley. Following this approach, we were able to detect some 259 metabolites and to define that some 12 of them were preferentially localized to the vacuole, 34 were preferentially localized outside of the vacuole, and 52 were distributed equally in the cell.

Primary metabolites generally displayed the partition that would be anticipated with organic acids and sugars as well as a few amino acids clearly being stored here. Of particular note are the high accumulations of isocitrate, citrate, and aconitate as well as Glc and Fru. That said, there were some clear exceptions to that statement, with the organic acids fumarate and 2-oxoglutarate being essentially absent in the vacuole, as were Suc and shikimate. Barley was additionally characterized by the accumulation of His, Ala, Trp, Met, and Ser in the vacuole; however, many of the other amino acids preferentially accumulate outside the vacuole. The pattern of organic acid partitioning suggests that 2-oxoglutarate and fumarate do not represent storage compounds in barley. In the case of 2-oxoglutarate, this is perhaps unsurprising given the crucial role of this metabolite in nitrate assimilation and the TCA cycle (Araújo et al., 2008, 2011). The fact that fumarate appears not to be a storage metabolite clearly contrasts with the situation in Arabidopsis, where, like malate, it appears to play a crucial role as a carbon store (Fahnenstich et al., 2007; Zell et al., 2010). The difference between these species may well be

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**Figure 3.** Subclusters of coexpression network analysis to predict the functions of tonoplast-located proteins. Subclusters of the frameworks of the coexpression network in Figure 2 indicated flavonoid- and phenylpropanoid-related biosynthesis (A), storage protein (B), and mugineic acid biosynthesis (C). Genes (probe sets) are depicted as nodes, while significant coexpression similarity between any two probe sets is represented by an edge. Probe sets corresponding to tonoplast-located proteins

explained by the presence of a cytosolic fumarase (Pracharoenwattana et al., 2010) and the fact that fumarate is an important transport form of carbon in Arabidopsis (Chia et al., 2000). The relative levels of sugars is consistent with a very high vacuolar invertase activity in barley, as has been described for other species including potato, maize (*Zea mays*), carrot (*Daucus carota*), and tobacco (*Nicotiana tabacum*; Büssis et al., 1997; Tang et al., 1999; Trouverie et al., 2004; Junker et al., 2006). Furthermore, it is consistent with the presence of a large number of sugar-conjugated secondary metabolites in this compartment.

Perhaps surprisingly, a large number of the secondary metabolites we detected were not found to be preferentially accumulated in the vacuole; however, there are some clear exceptions to this. Among the 49 peaks that were annotated, a chemical formula by accurate m/z value, 26 of the secondary metabolites were detected to be at higher abundance in the vacuole. Characterization of the chemical composition of secondary metabolites in barley has not been extensively reported, although there are some publications concerning flavonoids (Norbaek et al., 2000; Kaspar et al., 2010), sterol derivatives (Bush et al., 1971; Konig and Seifert, 1998), and HODE (Hamberg and Hamberg, 1996; Hübke et al., 2005). In our metabolite profiling by FT-MS with accurate m/z value for chemical formula prediction, we confirmed a formula for these compounds. All peaks we listed in Table II except Phe and Trp were not observed in Arabidopsis or tomato tissues, but similar peaks such as glycoflavone glycosides were detected in monocot plants such as rice (Oryza sativa) and maize (Besson et al., 1985; Snook et al., 1995). In our measurement of barley secondary metabolites, 12 putative flavonoids, seven putative HODE-related compounds, three cyclopentaneoctanoic acid-related compounds, and nine putative sterol derivatives were annotated. The compounds in each class displayed similar behavior; for example, all flavonoids were higher in vacuole, whereas HODErelated compounds and sterol-related compounds were not detected in vacuole samples. The sterolrelated compounds are likely located at the tonoplast membrane (Yoshida and Uemura, 1986). In addition, the cycloartenol synthase protein (AAT38887 homolog, Contig11705 at; Supplemental Table S2), which is involved in sterol biosynthesis, was previously identified as a "membrane fusion and remodeling protein" (Endler et al., 2006).

In the case of dicot plants such as Arabidopsis leaves, flavonol-O-glycoside, anthocyanin, and glucosinolates have been identified as major secondary metabolites (Tohge et al., 2005; Maruyama-Nakashita et al., 2006; Hirai et al., 2007; Matsuda et al., 2009), whereas other dicots such as Solanaceae species, including tomato and tobacco, accumulate glycoalkaloids, quinate derivatives, and acylated sugars (Moco et al., 2006; Butelli et al., 2008; Iijima et al., 2008; Adato et al., 2009). On the other hand, in leaves of monocot plant species such as rice, maize, and barley, flavonol-O-glycoside and flavonol-C-glycoside (glycoflavone) are reported (Norbaek et al., 2000; Kaspar et al., 2010). Typical flavonoids such as flavonol and anthocyanin are transported to the vacuole after their conversion to glycosides and in some cases acylates, either in exchange for protons or by direct energization by ABC transporters. In Arabidopsis research, multidrug and toxic compound extrusion-type transporter (Debeaujon et al., 2001; Marinova et al., 2007) and the soluble glutathione S-transferase proteins (Kitamura et al., 2004, 2010) are characterized as key players of flavonoid transport into the vacuole. In maize, genetic evidence was provided that ABC transporters are implicated in anthocyanin transport (Goodman et al., 2004). Based on this assumption, an integration analysis was performed utilizing coexpression analysis of publicly available microarray data. Generally, coexpression analysis does not work well for transportrelated proteins, since the expression of genes encoding transport-related proteins such as ABC transporters is generally low, and they are not usually stress responsive (Klein et al., 2006). Despite this, gene coexpression analysis using data obtained for different developmental stages has previously revealed that TT12 (a multidrug and toxic compound extrusion transporter for seed pigment proanthocyanidin in Arabidopsis) was well correlated with enzymatic genes of proanthocyanidin metabolism, such as BAN leucocyanidin leucopelargonidin (At1g61720) and laccase-like polyphenol oxidase (AtLAC15; At5g48100; Tohge and Fernie, 2010). In the other example of transporters for glucosinolate that accumulated in the vacuole of *Brassica* species, BAT5 (At4g12030), which is required for the biosynthesis of Met-derived glucosinolates, was well correlated in the coexpression network analysis (Hirai et al., 2007; Gigolashvili et al., 2009). Our trial of coexpression network analysis gave us many hints that may aid in understanding of the vacuolar transport system for flavonoid, storage proteins, and mugineic acid. The case of flavonoid acts as a positive control, given that this accumulation in the vacuole was experimentally annotated in several plant species (Song et al., 2010). Intriguingly, it also suggested a transport system for phytosiderophores, and it would

Figure 3. (Continued.)

<sup>(</sup>depicted by blue nodes; Supplemental Figure S3) were used as queries for searching the coexpression networks (Supplemental Table S2). Annotation of the probe sets and PageMan analysis of ontology terms can be found in Supplemental Table S2. CAD, Cinnamoyl-alcohol dehydrogenase; GST, glutathione *S*-transferase; MDR, multidrug-resistant-type transporter; NAM, no apical meristem; NUDT, nudix hydrolase homolog; PHS, propyzamide-hypersensitive protein; PPIK, phosphatidylinositol 3- and 4-kinase; SKIP, SKP-interacting partner; UGT, UDP-glycosyltransferase; VIP, vanabin-interacting protein.

seem likely that such a system plays a role in metal chelators under metal stress and nutrient-deficient conditions. As we showed in Figures 2 and 3, some metabolites that need to be transported into the vacuole revealed correlation in gene expression between those genes involved in their biosynthesis and those involved in their transport system. In this study, we used barley as an example of the power of combining data from metabolite profiling, subcellular proteomics, and in silico coexpression analysis in order to better understand the storage metabolome. Using this approach, we were able to characterize the storage metabolome of this species in spite of the fact that the genome sequence of barley is not yet available.

In summary, the results presented here provide the basis for a more comprehensive understanding of the vacuolar systems of transport and metabolism. While the partitioning of a number of these metabolites is in keeping with what would be anticipated, we additionally provide information on a number of other metabolites. That said, comparison between the barley vacuole data presented here and that found in Arabidopsis suggests that while many factors are conserved between species, there are subtle differences also. While it is possible that some of these differences are due to the different approaches used between the two studies, it would appear likely that the vacuolar metabolome, like the cellular metabolome (Fiehn, 2007; Krueger et al., 2011), is context dependent. With this in mind, it will be highly interesting in future studies to compare the storage metabolomes of a broad range of tissues types and its dynamic response to cellular circumstances, including exposure to salt stress, drought, and conditions that induce senescence.

#### MATERIALS AND METHODS

#### Plant Materials and Harvest

Barley (*Hordeum vulgare* var Baraka) was grown in vermiculite (Vermica) in a controlled-environment chamber (16 h of light/8 h of dark, 300 mE m<sup>-2</sup> s<sup>-1</sup>, 227°C, 60% relative humidity) and watered with Luwasa hydroculture nutrient solution (Interhydro). The primary leaves were harvested between 8 and 9 AM as described by Rentsch and Martinoia (1991) and Endler et al. (2006).

#### Protoplast and Vacuole Isolation for Metabolomics

Barley protoplasts and vacuoles for metabolomics were isolated by a slight modification of the procedure described by Rentsch and Martinoia (1991). Briefly, for protoplast isolation, after digestion of the leaves, protoplasts were collected by centrifugation at 1,200g for 10 min on a cushion constituted of digestion medium (500 mM sorbitol, 1 mM CaCl<sub>2</sub>, and 10 mM MES-KOH, pH 6) containing 30% Percoll (v/v). The supernatant was removed, the protoplasts were mixed with the Percoll cushion, and additional osmotically dissolved Percoll (digestion medium solved in Percoll) was added to give a final Percoll concentration of about 40%. For the metabolomics of the protoplasts, the protoplast solution was overlaid with medium C (500 mM sorbitol, 20% Percoll, pH 6, 1 mM CaCl<sub>2</sub>, and 10 mM MES-KOH, pH 5.6) and medium D (500 mM betaine, 1 mM CaCl<sub>2</sub>, and 10 mM MES-KOH, pH 5.6). The volume used depended on the quantity of protoplasts isolated. After centrifugation for 10 min at 1,200g, the protoplasts were recovered from the upper interphase. For the metabolomic studies of protoplasts, 200  $\mu$ L of a silicon oil mixture (AR200:

AR20, 8:2) was placed at the bottom of polyethylene microcentrifugation tubes (400- $\mu$ L capacity). Two milliliters of protoplasts was mixed with 2 mL of medium D, and 200  $\mu$ L of this mix was placed on the top of the silicon oil mixture. Protoplasts were sedimented by a 2 × 20-s centrifugation at 10,000g. The polyethylene tubes (50 per experiment) were frozen overnight. The bottoms of the frozen tubes were cut with a razor blade and collected. One milliliter of water was added and vortexed, and the aqueous fraction containing a tiny amount of silicon oil but no tube tips was transferred in a fresh tube. The mixture was centrifuged at 4,000g for 10 min. The water phase was removed, lyophilized, and used for the metabolomic analysis.

For the vacuole isolation, protoplasts were purified using a similar gradient but replacing medium C by C' (400 mM sorbitol, 20% Percoll, pH 7.2, 30 тм KCl, 20 тм HEPES-KOH, pH 7.2, 0.1% bovine serum albumin, and 0.2 тм dithiothreitol) and medium D by D' (400 mM betaine, 30 mM KCl, 20 mM HEPES-KOH, pH 7.2, and 0.2 mM dithiothreitol). Vacuoles were released from protoplasts by forcing the protoplasts through a syringe and purified by flotation as described by Rentsch and Martinoia (1991). This procedure occurred at 4°C. In order to get rid of the isolation medium, 30 µL of vacuoles was added to 70  $\mu L$  of a medium containing 22% Percoll, pH 7.2, 400 mm sorbitol, 30 mM KCl, 20 mM HEPES-KOH, pH 7.2, 0.1% bovine serum albumin, and 0.2 mM dithiothreitol. The samples were overlaid with 200  $\mu$ L of silicone oil AR200 and 60 µL of water. After mixing for 6 min, the vacuoles were floated by centrifugation at 10,000g for 20 s. Tubes were frozen overnight, and after defreezing, 45  $\mu L$  of the aqueous solution from each tube (50–60 per experiment) was recovered and pooled. The pooled fraction was lyophilized and used for the metabolomic analysis.

#### **Extraction for Metabolite Analysis**

Lyophilized sample was homogenized using a ball mill precooled with liquid nitrogen and extracted in 1,400  $\mu$ L of methanol, and 60  $\mu$ L of internal standard (0.2 mg ribitol mL<sup>-1</sup> water) was subsequently added as a quantification standard. Samples were homogenized by the Retsch Muhle mixer mill for 2 min at 25 L s<sup>-1</sup>. One thousand two hundred microliters of supernatant after centrifugation at 14,000 rpm for 10 min was transferred to a glass vial. After mixing with 750  $\mu$ L of CHCl<sub>3</sub>, 1,500  $\mu$ L of water was added and vortexed for 15 s. After centrifugation for 15 min at 4,000 rpm, supernatant was taken (50 and 100  $\mu$ L for GC-MS analysis and 1,000  $\mu$ L for LC-MS analysis) from the upper polar phase into fresh Eppendorf tubes (also 2  $\mu$ L). Sample dried by the SpeedVac for at least 2 h without heating was used for metabolite analysis.

#### Derivatization and Analysis of Primary Metabolites Using GC-MS

Metabolite extraction for GC-MS was performed by a method modified from that described by Roessner-Tunali et al. (2003). The extraction, derivatization, standard addition, and sample injection were exactly as described previously (Lisec et al., 2006). Both chromatograms and mass spectra were evaluated using either TAGFINDER (Luedemann et al., 2008) or the MAS-SLAB program (ThermoQuest), and the resulting data were prepared and presented as described (Roessner et al., 2001).

#### **UPLC-FT-MS of Secondary Metabolites**

UPLC separation of secondary metabolites was performed according to a previously published protocol (Giavalisco et al., 2009) using a Waters Acquity UPLC system. The UPLC apparatus was equipped with an HSS T3  $\mathrm{C}_{\mathrm{18}}$ reverse-phase column (100  $\times$  2.1 mm i.d., 1.8- $\mu$ m particle size; Waters), which was operated at a temperature of 40°C. The mobile phases consisted of 0.1% formic acid in water (solvent A) and 0.1% formic acid in acetonitrile (solvent B). The flow rate of the mobile phase was 400  $\mu$ L min<sup>-1</sup>, and 2  $\mu$ L of sample was loaded per injection. The following gradient profile was applied: after a 1-min isocratic run at 99% A, a linear 12-min gradient was applied to 65% A; this was immediately followed by a 1.5-min gradient to 30% A before a 1-min gradient to 1% A; then, a 1.5-min isocratic period at 1% A followed, before switching back to 99% A to reequilibrate the column for 2.5 min, before the next sample could be injected. The UPLC device was connected to the FT-ICR apparatus via a TriVersa NanoMate (Advion). The UPLC flow rate at 400  $\mu$ L min<sup>-1</sup> was split 1:1,000 with a T-valve (Advion). One-tenth of 1% (400 nL min<sup>-1</sup>) was directly loaded to the FT-ICR-MS device, while 99.9% was discarded. The sample was infused into the mass spectrometer via a nanospray chip (type A; Advion) by applying a voltage of 1.8 kV in the positive and 1.9 kV in the negative ionization mode. Spray sensing was used between 1 and 17 min of the UPLC gradient. The mass spectra were acquired using the LTQ FT-ICR-Ultra mass spectrometer (Thermo-Fisher). The spectra were recorded using full-scan mode, covering a mass range from m/z 100 to 1,500. Resolution was set to 50,000, and maximum loading time for the ICR cell was set to 500 ms. The transfer capillary temperature was set to 200°C, and the MS spectra were recorded from 0 to 19 min of the UPLC gradient. Molecular masses, retention times, and associated peak intensities were extracted from the raw files using the RefinerMS software (version 5.3; GeneData). Peak prediction and annotation in the data matrix by LC-MS was conducted by databases (for review, see Tohge and Fernie, 2009) such as MASSBANK (Horai et al., 2010) and KNApSAcK (Shinbo et al., 2006), based on accurate m/z analyzed by FT-MS.

#### **Coexpression Analysis via Proteome Data**

The coexpression network analysis of wheat (*Triticum aestivum*) was performed by PlaNet (Mutwil et al., 2011) using Supplemental Table S2.

#### Supplemental Data

- The following materials are available in the online version of this article.
- Supplemental Figure S1. Global comparison of all detected compounds by GC-MS and LC-MS in protoplast and vacuole.
- Supplemental Figure S2. Probe identifiers of genes shown in the framework of the coexpression network in Figure 2.
- Supplemental Figure S3. Probe identifiers of genes shown in the framework of the coexpression network in Figure 3.
- Supplemental Table S1. Secondary metabolite-related peaks detected by UPLC-FT-MS
- Supplemental Table S2. The barley vacuole proteins used for coexpression analysis.

Received August 19, 2011; accepted September 21, 2011; published September 26, 2011.

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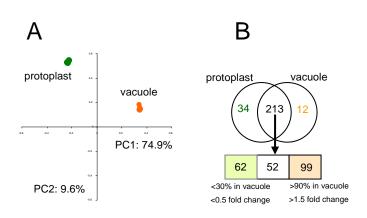
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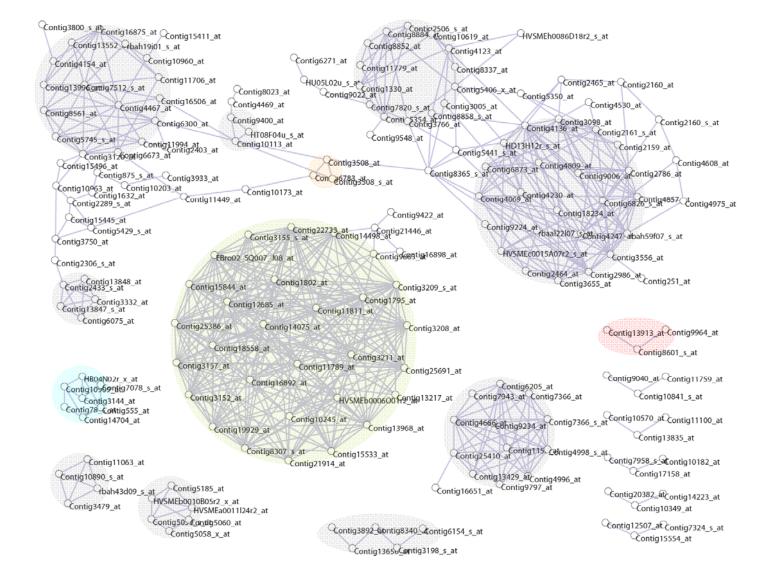
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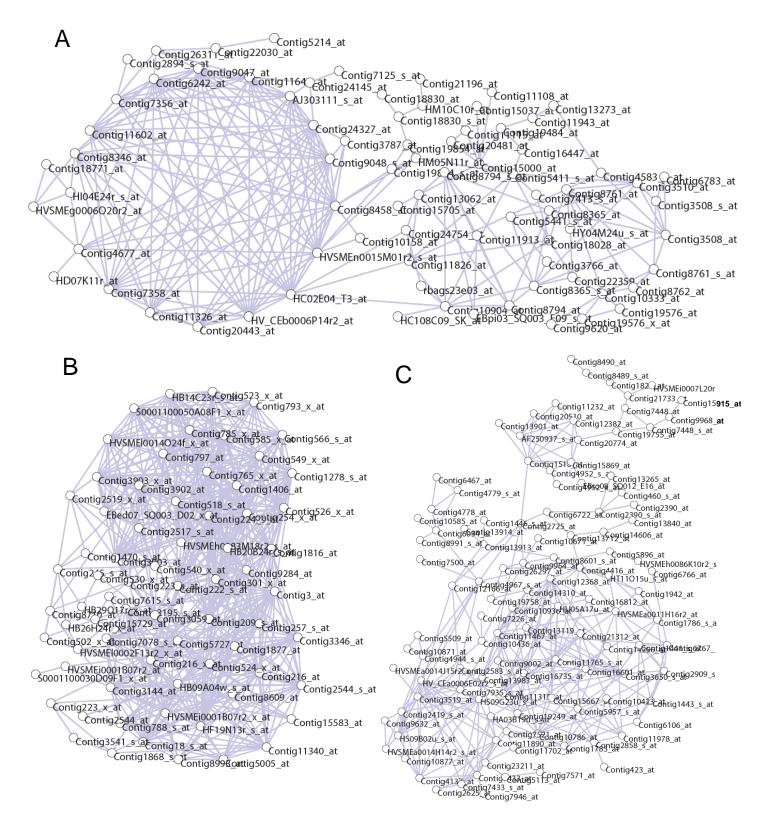


Supplemental Figure 1. Global comparison of all detected compounds by GC-MS and LC-MS in protoplast and vacuole. (A) Principal component analysis (PCA) using all detected peaks in GC-MS or LC-MS (260 peaks). (B) Venn Diagram to show common peaks and specific peaks.

## Tohge et al., Fig S2



Supplemental Figure 2. Probe ID of genes shown in the framework of co-expression network in Figure 2.



Supplemental Figure 3. Probe ID of genes shown in the framework of co-expression network in Figure 3A-C.